



UNIVERSIDADE ESTADUAL DE MARINGÁ
CENTRO DE CIÊNCIAS AGRÁRIAS
Programa de Pós-Graduação em Ciência de Alimentos

Caracterização de curdlana produzida por *Agrobacterium* sp. IFO 13140 e seus géis por meio de FT-IR, FT-Raman e análises térmicas. Avaliação da morfologia, propriedades físico-químicas, tecnológicas, reológicas e aplicação em alimentos

CAMILA SAMPAIO MANGOLIM

Maringá

2017

CAMILA SAMPAIO MANGOLIM

Caracterização de curdlana produzida por *Agrobacterium* sp. IFO 13140 e seus géis por meio de FT-IR, FT-Raman e análises térmicas. Avaliação da morfologia, propriedades físico-químicas, tecnológicas, reológicas e aplicação em alimentos

Tese apresentada ao programa de Pós-Graduação em Ciência de Alimentos da Universidade Estadual de Maringá, como parte dos requisitos para obtenção do título de doutor em Ciência de Alimentos.

Maringá

2017

Dados Internacionais de Catalogação na Publicação (CIP)
(Biblioteca Central - UEM, Maringá, PR, Brasil)

M277c Mangolim, Camila Sampaio, 1988-
Caracterização de curdlana produzida por
Agrobacterium sp. IFO 13140 e seus géis por meio de
FT-IR, FT-Raman e análises térmicas : avaliação da
morfologia, propriedades físico-químicas,
tecnológicas, reológicas e aplicação em alimentos /
Camila Sampaio Mangolim. -- Maringá, 2017.
125 f. : il. color., figs., tabs.

Orientadora: Prof.^a Dr.^a Graciette Matioli. Tese
(doutorado) - Universidade Estadual de Maringá,
Centro de Ciências Agrárias, Programa de
Pós-Graduação em Ciência de Alimentos, 2017.

1. Beta-glucana. 2. Polissacarídeo microbiano. 3.
Melhorador de textura - Alimentos. 4. Espessante -
Alimentos. 5. Gelificante - Alimentos. I. Matioli,
Graciette, orient. II. Universidade Estadual de
Maringá. Centro de Ciências Agrárias. Programa de
Pós-Graduação em Ciência de Alimentos. III. Título.

CDD 23.ed. 660.6

GVS-003729

CAMILA SAMPAIO MANGOLIM

**“CARACTERIZAÇÃO DE CURDLANA PRODUZIDA POR
AGROBACTERIUM SP. IFO 13140 E SEUS GÉIS POR MEIO DE FT-IR, FT-
RAMAN E ANÁLISES TÉRMICAS. AVALIAÇÃO DA MORFOLOGIA,
PROPRIEDADES FÍSICO-QUÍMICAS, TECNOLÓGICAS, REOLÓGICAS E
APLICAÇÃO EM ALIMENTOS”.**

Tese apresentada à Universidade Estadual de Maringá, como parte das exigências do Programa de Pós-graduação em Ciência de Alimentos, para obtenção do grau de Doutor em Ciência de Alimentos.



Profa. Dra. Rosane Marina Peralta



**Profa. Dra. Vanessa Aparecida Marcolino
Pittarelli**



Prof. Dr. Antonio Medina Neto



Profa. Dra. Márcia Portilho



**Profa. Dra. Graciette Matioli
Orientadora**

Maringá - 2017

Orientadora

Profa. Dra. Graciette Matioli

BIOGRAFIA

Camila Sampaio Mangolim nasceu em 8 de fevereiro de 1988 na cidade de Paranaíba, Paraná. Possui graduação em Engenharia de Alimentos pela Universidade Estadual de Maringá e mestrado em Ciência de Alimentos pela mesma universidade. Tem experiência na área de Biotecnologia, atuando principalmente nos seguintes temas: produção de ciclodextrinas e suas aplicações em alimentos; produção e caracterização de biopolímeros microbianos para aplicação em alimentos.

Dedico

À minha família, meu alicerce, meu porto seguro. Que sempre me fez acreditar que era possível...

*“Porque eu sou do tamanho do que vejo
E não, do tamanho da minha altura.”*

(Fernando Pessoa)

*“Não coloque limites nos seus sonhos,
coloque fé.”*

(Autor desconhecido)

AGRADECIMENTOS

A Deus, pela sua compaixão, bondade e por todas as graças concedidas.

Aos meus pais, Olidio e Joselita, que nunca deixaram de me apoiar. Por tudo o que fizeram e fazem por mim, pelo amor dedicado, pelo incentivo, por serem meus exemplos e por terem me ensinado todos os valores éticos e morais que conheço.

À minha irmã Amanda, meu amor incondicional. Por estar sempre presente em todos (ou quase todos) os momentos, sendo além de irmã uma grande amiga.

À minha orientadora Profa. Dra. Graciette Matioli, pelo ensino, aprendizado, confiança, paciência, amizade e por todo o amadurecimento que me proporcionou durante a minha caminhada de iniciação científica à pós-graduação.

Aos meus familiares, pela companhia constante e tão querida. Pelo carinho e apoio, por estarem próximos e fazendo esta vida valer a pena cada dia mais.

Aos meus amigos de perto e de longe, que compartilharam comigo as felicidades e também tristezas. Pelo convívio, apoio, compreensão, cumplicidade, ajuda e amizade.

Aos meus colegas de laboratório, por todos os momentos que passamos juntos, pela troca de conhecimentos e experiências, pelas dicas e sugestões durante este trabalho. Por compartilharem angústias, vitórias, choros e inquietações pessoais e profissionais.

Aos professores e alunos de outros departamentos, pela ajuda nos experimentos, pelas intervenções e debates. Pelo tempo dispensado a esta pesquisa, pelas ideias, auxílios e correções.

Aos servidores da UEM, que tornam possível o desempenho das nossas pesquisas.

A todos aqueles que de alguma forma doaram um pouco de si para que a conclusão deste trabalho se tornasse possível.

APRESENTAÇÃO

Esta tese de doutorado está apresentada na forma de dois artigos científicos:

1. **AUTORES:** Camila Sampaio Mangolim, Thamara Thaianne da Silva, Vanderson Carvalho Fenelon, Luciana Numata Koga, Sabrina Barbosa de Souza Ferreira, Marcos Luciano Bruschi e Graciette Matioli.

TÍTULO: Description of recovery method used for curdlan produced by *Agrobacterium* sp. IFO 13140 and its relation to the morphology and physicochemical and technological properties of the polysaccharide.

REVISTA: Plos One.

Artigo publicado.

2. **AUTORES:** Camila Sampaio Mangolim, Thamara Thaianne da Silva, Vanderson Carvalho Fenelon, Adriane do Nascimento, Francielle Sato e Graciette Matioli.

TÍTULO: Use of FT-IR, FT-Raman and thermal analysis to evaluate the gel formation of curdlan produced by *Agrobacterium* sp. IFO 13140 and determination of its rheological properties with food applicability.

REVISTA: Food Chemistry.

Artigo submetido.

RESUMO GERAL

INTRODUÇÃO. Curdlana é um polissacarídeo linear, neutro, composto por repetidas unidades de glicose unidas por ligações β -(1,3), sendo considerada uma fibra alimentar. Possui especial interesse devido as suas propriedades de formação de gel, as quais permitem seu uso como agente gelificante para melhorar a textura, capacidade de retenção de água e estabilidade térmica de vários produtos alimentícios. Quando aquecida, sua suspensão aquosa forma dois tipos de gel: o gel *low-set*, o qual é termorreversível e obtido pelo aquecimento em temperaturas entre 55 e 60 °C, seguido de resfriamento, e o gel *high-set*, que é termo-irreversível e obtido pelo aquecimento acima de 80 °C. Devido a sua elevada firmeza e estabilidade, os géis de curdlana são utilizados na elaboração de massas, molhos, alimentos congelados e enlatados, além de serem biodegradáveis, comestíveis e não tóxicos para humanos e para o ambiente.

Apesar de a curdlana ser um polissacarídeo linear, sua estrutura conformacional exerce grande influência nas suas propriedades de dispersão e gelificação. Uma vez que a forma de recuperação do meio e a purificação do polissacarídeo podem influenciar na sua estrutura conformacional, o estudo estrutural da curdlana, bem como das suas propriedades físico-químicas e tecnológicas, são essenciais para melhor adequar seu emprego na indústria de alimentos.

OBJETIVOS. Considerando o interesse na curdlana devido às suas propriedades de dispersão, viscosidade e gelificação, o presente trabalho objetivou avaliar a estrutura, morfologia e propriedades físico-químicas e tecnológicas da curdlana produzida por *Agrobacterium* sp. IFO 13140, comparativamente com a curdlana comercial. Também foi objetivo desta pesquisa a aplicação da curdlana em alimentos, como massa alimentícia caseira e iogurte.

MATERIAL E MÉTODOS. A cepa de *Agrobacterium* sp. IFO 13140 adquirida na forma liofilizada foi reativada e adicionada ao meio de produção de curdlana utilizando glicose como fonte de carbono. Duas metodologias foram utilizadas para a recuperação do polissacarídeo: a de pré-gelificação e a de precipitação. Em ambas as metodologias a curdlana foi solubilizada com solução de NaOH 3 mol/L. Posteriormente foi neutralizada com HCl 3 mol/L (pré-gelificação) ou com água (precipitação). A curdlana comercial também foi submetida ao processo de pré-gelificação e, então, denominada comercial pré-gelificada.

As metodologias empregadas para recuperação da curdlana foram relacionadas com suas propriedades funcionais. As curdlanas comercial, comercial pré-gelificada e produzidas por *Agrobacterium* sp. IFO 13140 (recuperadas pelos dois métodos) foram avaliadas quanto a sua estrutura, morfologia, propriedades físico-químicas e capacidade de dispersão e gelificação. Após esta etapa, as duas curdlanas submetidas ao processo de pré-gelificação foram utilizadas para o preparo de dispersões e géis. Foram estudados os mecanismos de formação de géis de curdlana por espectroscopias de FT-IR e FT-Raman, bem como suas temperaturas de formação por análises térmicas de DSC e TGA.

As curdlanas submetidas ao processo de pré-gelificação, assim como a curdlana comercial, foram avaliadas quanto as suas propriedades tecnológicas de absorção de água, óleo e solubilidade em água. Posteriormente, foram aplicadas em massa alimentícia caseira e iogurte nas proporções de 1% e 1,5%, em relação às quantidades de farinha e leite, respectivamente. As características de cozimento e o perfil de textura das massas foram avaliados. Dos iogurtes, foram avaliadas reologia, textura e sinérese.

RESULTADOS E DISCUSSÃO. A análise de FT-Raman das curdlanas comercial, comercial pré-gelificada e produzida por *Agrobacterium* sp. IFO 13140 (recuperada pelos métodos de pré-gelificação e precipitação) revelou similaridade estrutural entre todas as amostras, o que significa que a curdlana produzida pelo micro-organismo tem estrutura similar à comercial e, também, que o método empregado para recuperação do polissacarídeo não influencia na sua estrutura. Entretanto, dados da microscopia eletrônica de varredura revelaram que tanto o método de secagem quanto o método de recuperação implicam significativamente na forma e no tamanho dos grânulos da curdlana. A amostra comercial (atomizada) apresentou grânulos grandes e com concavidades. As demais amostras (liofilizadas) exibiram partículas menores. Ainda, as duas curdlanas submetidas ao processo de pré-gelificação apresentaram o menor tamanho de partícula, o que pode ter influenciado na sua melhor dispersão em água.

Os resultados obtidos nas análises físico-químicas revelaram que a curdlana comercial apresentou o maior teor de carboidratos, seguida da produzida pelo micro-organismo e recuperada por precipitação, demonstrando o maior grau de pureza dessas amostras. As curdlanas submetidas ao processo de pré-gelificação apresentaram menor teor de carboidratos devido ao sal incorporado no processo (estimado pelo teor de sódio). Quando avaliadas as características de dispersão e gelificação das curdlanas, as amostras pré-gelificadas formaram facilmente uma dispersão homogênea em água, além de desenvolverem gel firme e homogêneo. A curdlana comercial não formou dispersão homogênea com facilidade e nem gel homogêneo, assim como a curdlana produzida por *Agrobacterium* sp. IFO 13140 recuperada por precipitação, que não formou gel pela ausência de sódio. Desta forma, as análises térmicas, reológicas e de força de gel foram realizadas apenas com as curdlanas pré-gelificadas. A curdlana comercial foi comparada com as pré-gelificadas em relação às propriedades tecnológicas de absorção de água, de óleo e solubilidade em água, além da aplicação em alimentos.

As análises de FT-IR e FT-Raman revelaram similaridade estrutural entre a curdlana comercial e a produzida por *Agrobacterium* sp. IFO 13140. Não foi possível identificar variações estruturais decorrentes da formação de gel pela técnica de FT-IR. Entretanto, a técnica de FT-Raman evidenciou essas variações, sendo possível observar variações relacionadas às ligações de hidrogênio e interações hidrofóbicas, que ocorrem com a formação dos géis *low-set* e *high-set*, respectivamente. As temperaturas de formação de ambos os géis foram determinadas por DSC. As amostras apresentaram um pico endotérmico entre 40 e 55 °C, decorrente do entumescimento da curdlana, e um pico entre 70 e 80 °C devido a interações hidrofóbicas entre as moléculas. Ainda, as análises de reologia e força do gel revelaram que a curdlana produzida por *Agrobacterium* sp. IFO 13140 mostrou capacidade espessante superior à curdlana comercial pré-gelificada. Porém, a última mostrou capacidade gelificante mais elevada, com gel de força 17% superior e módulo elástico seis vezes maior, a 95 °C. Esta grande diferença entre as capacidades gelificantes das curdlanas é devido aos seus distintos graus de polimerização (DP_n) ou massa molecular, pois enquanto a amostra comercial apresentou DP_n de 334 ± 8 , a amostra produzida pelo micro-organismo apresentou DP_n de 232 ± 10 , os quais correspondem a massas moleculares de 54000 e 38000, respectivamente.

Quando avaliadas as propriedades de absorção de água e óleo e solubilidade em água das curdlanas comercial e das submetidas ao processo de pré-gelificação, confirmou-se a baixa solubilidade e baixa absorção de água de todas as amostras. Entretanto, as curdlanas submetidas ao processo de pré-gelificação apresentaram valores de absorção de óleo elevados e superiores aos de absorção de água, dado este que torna estes ingredientes úteis nas interações estruturais de diversos alimentos.

O emprego das curdlanas submetidas ao processo de pré-gelificação em massa alimentícia e iogurte melhorou as características de textura dos produtos. Na massa, as curdlanas pré-gelificadas tanto aumentaram seu peso cozido (cerca de 10%) quanto aumentaram significativamente sua dureza, adesividade e gomusidade. A curdlana comercial não provocou melhora significativa em nenhum dos parâmetros da massa porque, devido a sua dificuldade de homogeneização, não gelifica de forma homogênea após o cozimento do produto. De forma semelhante, quando aplicadas no iogurte e submetidas a tratamento térmico, as curdlanas pré-gelificadas aumentaram significativamente os parâmetros de firmeza e adesividade e diminuíram a coesividade dos produtos, devido à estruturação provocada pela formação do gel de curdlana. A sinérese foi reduzida nos iogurtes com curdlana, sendo mais eficiente para a curdlana comercial pré-gelificada que, apesar de ter menor influência na firmeza que a produzida por *Agrobacterium* sp. IFO 13140, formou um gel mais estável ao armazenamento. Os dados reológicos corroboraram com os dados do perfil de textura dos iogurtes, pois as amostras submetidas ao tratamento térmico apresentaram maiores valores de viscosidade em toda a faixa de taxa de deformação avaliada.

CONCLUSÕES. As características de dispersão, de gelificação e reológicas da curdlana dependem demasiadamente dos métodos de recuperação empregados após a sua produção, pois as curdlanas comercial e produzida por *Agrobacterium* sp. IFO 13140 submetidas ao processo de pré-gelificação dispersaram melhor em água e atuaram como espessantes, além de formarem géis homogêneos. Os dados obtidos nas análises estruturais mostraram similaridade entre a curdlana produzida por *Agrobacterium* sp. IFO 13140 e a comercial. Dentre as análises empregadas, a técnica de FT-Raman mostrou-se especialmente valiosa no estudo das variações estruturais decorrentes da formação dos géis de curdlana, evidenciando as ligações de hidrogênio e as interações hidrofóbicas que ocorrem com a formação dos géis *low-set* e *high-set*, respectivamente. Apesar da similaridade estrutural, as propriedades tecnológicas das curdlanas apresentaram diferenças significativas, as quais estão relacionadas com seus diferentes graus de polimerização. A curdlana produzida por *Agrobacterium* sp. IFO 13140 (método de pré-gelificação), que apresentou o menor grau de polimerização, mostrou maior capacidade espessante e de retenção de água e óleo que a comercial pré-gelificada, porém a última revelou capacidade gelificante superior. Como consequência, apesar de ambas terem aumentado a dureza e absorção de água das massas alimentícias e terem aumentado a firmeza e a viscosidade dos iogurtes e diminuído sua sinérese, a curdlana produzida por *Agrobacterium* sp. IFO 13140 (método de pré-gelificação) provocou aumento maior nos parâmetros de textura dos produtos, mas com formação de gel menos estável que a comercial pré-gelificada. Conclui-se que os géis de curdlana avaliados mostraram estrutura rígida e estável, com potencial de aplicabilidade no aprimoramento da textura de produtos alimentícios. Desta forma, o estudo estrutural dos géis de curdlana e das suas propriedades físico-químicas e tecnológicas realizados nesta pesquisa são fundamentais para ajustar seu emprego na indústria de alimentos.

Palavras-chave: β -glucana, polissacarídeo microbiano, melhorador de textura, espessante, gelificante.

GENERAL ABSTRACT

INTRODUCTION. Curdlan is a linear, neutral polysaccharide composed of repeated β -(1,3)-linked glucose residues, and is considered a food fiber. It is of special interest due to its gel formation properties, which allow it to be used as a gelling agent to improve the texture, water retention capacity and thermal stability of various food products. When heated, its aqueous suspension forms two types of gel: low-set gel, which is thermo-reversible and obtained by heating at temperatures between 55 and 60 °C, followed by cooling, and high-set gel, which is thermo-irreversible and obtained by heating at over 80 °C. Because of their high firmness and stability, curdlan gels are used in the manufacture of pasta, sauces, frozen and canned foods, and are biodegradable, edible and non-toxic to humans and the environment.

Although curdlan is a linear polysaccharide, its conformational structure has a great influence on its dispersion and gelling properties. As the recovery of the medium and the purification of the polysaccharide can influence its conformational structure, the structural study of curdlan, as well as its physicochemical and technological properties, is essential for its more effective use in the food industry.

AIMS. Considering the interest in curdlan due to its dispersion, viscosity and gelling properties, the present study aimed to evaluate the structure, morphology and physical-chemical and technological properties of curdlan produced by *Agrobacterium* sp. IFO 13140, compared to commercial curdlan. A further objective of this study was the application of curdlan in foods, such as homemade pasta and yogurt.

MATERIAL AND METHODS. The strain of *Agrobacterium* sp. IFO 13140 acquired in lyophilized form was reactivated and added to the curdlan production medium using glucose as a carbon source. Two methodologies were used to recover the polysaccharide: pre-gelation and precipitation. In both methodologies the curdlan was solubilized with NaOH 3 mol/L solution. Subsequently it was neutralized with HCl 3 mol/L (pre-gelation) or with water (precipitation). The commercial curdlan was also subjected to the pre-gelation process and entitled pre-gelled commercial curdlan.

The methodologies used to recover the curdlan were related to their functional properties. The commercial and pre-gelled commercial curdlans and the curdlan produced by *Agrobacterium* sp. IFO 13140 (recovered by both methods) were evaluated for their structure, morphology, physicochemical properties and dispersion and gelation capacity. After this step, the two curdlans submitted to the pre-gelation process were used for the preparation of the dispersions and gels. The mechanisms of formation of the curdlan gels were examined by FT-IR and FT-Raman spectroscopy, and their formation temperatures were evaluated by DSC and TGA analyzes.

The curdlans submitted to the pre-gelation process, as well as the commercial curdlan, were evaluated for their technological properties of water absorption and oil and water solubility. Subsequently, they were applied in homemade pasta and yogurt in proportions of 1% and 1.5%, in relation to the quantities of flour and milk, respectively. The cooking characteristics and the texture profile of the pasta were evaluated. Rheology, texture and syneresis were evaluated from the yoghurt.

RESULTS AND DISCUSSION. FT-Raman analysis of the commercial and pre-gelled commercial curdlans and the curdlan produced by *Agrobacterium* sp. IFO 13140 (recovered by the pre-gelation and precipitation methods) revealed a structural similarity between all the samples, meaning that the curdlan produced by the microorganism has a

similar structure to the commercial curdlan, and that the method used for the recovery of the polysaccharide does not influence its structure. However, scanning electron microscopy data revealed that both the drying method and the recovery method significantly affected the shape and size of the curdlan granules. The commercial (atomized) sample had large granules with concavities. The other samples (lyophilized) exhibited smaller particles. Also, the two curdlans submitted to the pre-gelation process had the smallest particle size, which may have influenced their improved dispersion in water.

The results obtained in the physical-chemical analyzes revealed that the commercial curdlan had the highest carbohydrate content, followed by that produced by the microorganism and recovered by precipitation, demonstrating the greater purity of these samples. The curdlans submitted to the pre-gelation process presented a lower carbohydrate content, due to the salt incorporated in this process (calculated by the sodium content). When the characteristics of the dispersion and gelation of the curdlans were evaluated, the pre-gelled samples dispersed easily and homogeneously in water, and developed a firm and homogeneous gel. The commercial curdlan did not easily form a homogeneous dispersion or gel. The same was true for the curdlan produced by *Agrobacterium* sp. IFO 13140 recovered by precipitation, which did not gel due to the absence of sodium. Therefore, the thermal, rheological and gel strength analyzes were performed using the pre-gelled curdlans only. The commercial curdlan was compared with the pre-gelled curdlans in terms of the technological properties of absorption of water and oil and solubility in water, as well as application in foods.

The FT-IR and FT-Raman analyses revealed a structural similarity between the commercial curdlan and the curdlan produced by *Agrobacterium* sp. IFO 13140. It was not possible to identify the structural variations resulting from the gel formation using the FT-IR technique. However, the FT-Raman technique did identify these variations, allowing variations related to the hydrogen bonding and hydrophobic interactions to be observed. The formation temperatures of both gels were determined by DSC. The samples revealed an endothermic peak between 40 and 55 °C, due to the swelling of the curdlan, and another peak between 70 and 80 °C due to hydrophobic interactions between the molecules. Additionally, rheology and gel strength analysis revealed that the curdlan produced by *Agrobacterium* sp. IFO 13140 had a greater thickening capacity than the pre-gelled commercial curdlan. However, the latter exhibited a greater gelling capacity, resulting in a gel with a 17% greater strength and a six times greater elastic modulus at 95 °C. This difference between the gelling properties of the curdlans is due to their different degrees of polymerization (DP_n) or molecular mass, as while the commercial sample had a DP_n of 334 ± 8 , the sample produced by the microorganism had a DP_n of 232 ± 10 , corresponding to molecular masses of 54000 and 38000, respectively.

When the water and oil absorption and water solubility properties of the commercial curdlan and the curdlans from the pre-gelation process were evaluated, the low water solubility and absorption of all the samples were confirmed. However, the curdlans submitted to the pre-gelation process presented high oil absorption values that were greater than those of water absorption, making these ingredients useful in the structural interactions of various foods.

The use of curdlans submitted to the pre-gelation process in pasta and yoghurt improved the texture characteristics of these products. In the pasta, pre-gelled curdlans increased the cooked weight (about 10%) and significantly increased hardness, adhesiveness and gumminess. The commercial curdlan did not result in a significant improvement in any of the parameters of mass as, due to its difficulty with

homogenization, it did not gel homogeneously after cooking. Similarly, when applied to the yogurt and subjected to heat treatment, the pre-gelled curdlans significantly increased the firmness and adhesiveness parameters and decreased the cohesiveness of the products, due to the structuring caused by the formation of the curdlan gel. Syneresis was reduced in the yoghurts with curdlan, and was more efficient with the pre-gelled commercial curdlan which, although having less influence on firmness than the curdlan produced by *Agrobacterium* sp. IFO 13140, formed a more stable gel for storage. The rheological data corroborated with the texture profile data of the yoghurts, as the samples that underwent heat treatment showed higher values of viscosity across the entire deformation rate range evaluated.

CONCLUSIONS. The dispersion, gelling and rheological characteristics of curdlan are highly dependent on the recovery method employed after its production, as the commercial curdlan and the one produced by *Agrobacterium* sp. IFO 13140, submitted to the pre-gelation process, dispersed more effectively in water and acted as a thickener, in addition to forming homogeneous gels. The data obtained in the structural analyzes showed a similarity between curdlan produced by *Agrobacterium* sp. IFO 13140 and commercial curdlan. Among the analyzes employed, the FT-Raman technique proved to be especially valuable in the study of the structural variations resulting from the formation of curdlan gels, identifying the hydrogen bonds and the hydrophobic interactions that occur with the formation of the low-set and high-set gels, respectively. Despite the structural similarity, the technological properties of curdlans presented significant differences, related to their different degrees of polymerization. The curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-gelation method), which had the lowest degree of polymerization, exhibited a greater thickening and water and oil retention capacity than the pre-gelled commercial curdlan, although the latter demonstrated a superior gelling capacity. Therefore, although both increased the hardness and water absorption of the pasta and increased the firmness and viscosity of the yoghurts and decreased syneresis, the curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-gelation method) resulted in a greater increase in the texture parameters of the products, but with a less stable gel formation than the pre-gelled commercial curdlan. It can be concluded that the curdlan gels evaluated exhibited a rigid and stable structure, with potential application in the improvement of the texture of food products. In this way, the structural study of the curdlan gels and their physical-chemical and technological properties undertaken in this research is fundamental for refining their use in the food industry.

Keywords: β -glucan, microbial polysaccharide, texture improver, thickener, gelling agent.

ARTICLE 1

**Description of Recovery Method Used for Curdlan Produced by *Agrobacterium* sp.
IFO 13140 and its Relation to the Morphology and Physicochemical and
Technological Properties of the Polysaccharide**

Camila Sampaio Mangolim¹, Thamara Thaianne da Silva², Vanderson Carvalho
Fenelon³, Luciana Numata Koga³, Sabrina Barbosa de Souza Ferreira³, Marcos Luciano
Bruschi³ and Graciette Matioli^{1,3,*}

¹*Postgraduate Program in Food Science, State University of Maringá (UEM), Av.*

Colombo, 5790 - 87020-900, Maringá, PR, Brazil

²*Department of Food Engineering, State University of Maringá (UEM), Av. Colombo,*

5790 - 87020-900, Maringá, PR, Brazil

³*Postgraduate Program in Pharmaceutical Science, State University of Maringá*

(UEM), Av. Colombo, 5790 - 87020-900, Maringá, PR, Brazil

* Corresponding author

E-mail: gmatioli@uem.br (GM)

Abstract

Curdlan is a linear polysaccharide considered a dietary fiber and with gelation properties. This study evaluated the structure, morphology and the physicochemical and technological properties of curdlan produced by *Agrobacterium* sp. IFO 13140 recovered by pre-gelation and precipitation methods. Commercial curdlan submitted or

26 otherwise to the pre-gelation process was also evaluated. The data obtained from
27 structural analysis revealed a similarity between the curdlan produced by *Agrobacterium*
28 sp. IFO 13140 (recovered by both methods) and the commercial curdlans. The results
29 showed that the curdlans evaluated differed significantly in terms of dispersibility and
30 gelation, and only the pre-gelled ones had significant potential for food application,
31 because this method influence on the size of the particles and in the presence of NaCl.
32 In terms of technological properties, the curdlan produced by *Agrobacterium* sp. IFO
33 13140 (pre-gelation method) had a greater water and oil holding capacity (64% and 98%
34 greater, respectively) and a greater thickening capacity than the pre-gelled commercial
35 curdlan. The pre-gelled commercial curdlan displayed a greater gelling capacity at 95
36 °C than the others. When applied to food, only the pre-gelled curdlans significantly
37 improved the texture parameters of yogurts and reduced syneresis. The curdlan gels,
38 which are rigid and stable in structure, demonstrated potential for improving the texture
39 of food products, with potential industrial use.

40

41 **Introduction**

42 Polysaccharides are an extremely diverse family of natural biopolymers, which
43 are industrially used as thickeners, stabilizers and gelling agents in foodstuffs [1]. There
44 is currently growing interest in their biological functions, such as their antioxidant and
45 prebiotic activity. Although polysaccharides are derived from various sources such as
46 microorganisms, algae and higher plants, the market is dominated by polysaccharides
47 obtained from algae (such as carrageenans, alginates and agar) and higher plants (such
48 as starch, cellulose and pectin) [2].

49 The synthesis of microbial polysaccharides has emerged as an important source
50 of new biopolymers for industrial use [3]. It is an attractive alternative as

51 microorganisms can grow under controlled conditions and produce a large variety of
52 polysaccharides with unique properties. Production can be carried out in large quantities
53 by biotechnological routes and from renewable and low-cost raw materials, which are
54 easily recovered [4]. One of the microbial polysaccharides of industrial interest is
55 curdlan, a neutral exopolysaccharide composed exclusively of glucose residues joined
56 by β -(1 \rightarrow 3) bonds, commercially produced by bacterial species of *Agrobacterium* [2,5-
57 7].

58 After being discovered by Harada in 1966, curdlan has received considerable
59 attention in both the food and non-food industry due to its physicochemical properties,
60 which are unique when compared to other polysaccharides commonly used, such as
61 starch. When in aqueous suspension, curdlan may form a thermoreversible gel (low-set
62 gel) when heated to temperatures close to 55 °C with subsequent cooling. This gel has
63 similar behavior to agar and gelatin gels. In addition, curdlan can form a
64 thermoirreversible gel (high-set gel) when its aqueous suspension is heated to
65 temperatures above 80 °C, which is very stable. Furthermore, gels with different
66 strengths can be formed by varying the temperature, the heating time and the
67 concentration of curdlan [1,6].

68 Curdlan was approved for use in foodstuffs in Korea, Taiwan and Japan in 1989
69 and registered in 1996 by the FDA (*Food and Drug Administration*) in the USA as a
70 food additive with the following functions: a formulation and processing aid, and a
71 stabilizer, thickener and texturizer [8]. It is widely used due to its ability to form the
72 thermoirreversible gel, which exhibits great stability during the industrial processes of
73 autoclaving, frying and freezing-thawing cycles. It has no taste, color or odor, can form
74 gels at a wide pH range (from 2 to 10) and can mimic the palatability of foods
75 containing fat [9]. It is also considered a dietary fiber, thus improving the functionality

76 of various food products. The versatility of its applications associated with its health
77 benefits make curdlan a valuable tool in the development of innovative food systems
78 [1,3].

79 The dispersion and gelation characteristics of curdlan in an aqueous system, as
80 well as its mechanical properties, are associated with several factors such as
81 concentration, temperature, heating time, dispersal method, and the presence of ions,
82 salts and low molecular weight sugars [10]. Therefore, the form of recovery from the
83 culture medium and the purification of the polysaccharide are steps which decisively
84 influence the physical and technological properties of curdlan, which directly affect the
85 texture characteristics of foods in which the polysaccharide is inserted. So, it is
86 necessary to study these properties to define the most appropriate form of the
87 implementation of the polysaccharide in the industry [11].

88 Considering the interest in the properties of dispersion, viscosity, gelation and
89 the health benefits of curdlan, the present study aimed to characterize the
90 physicochemical and technological properties of curdlan produced by *Agrobacterium*
91 sp. IFO 13140 and recovered using the pre-gelation and precipitation methods, as well
92 as commercial curdlan subjected or otherwise to pre-gelation treatment. The
93 methodologies employed for the recovery of the curdlans were studied. The structures
94 of all curdlans were compared by FT-Raman spectroscopy. In addition, the application
95 of curdlans in yogurts was evaluated.

96

97

98

99

100

101 **Materials and methods**

102

103 **Materials**

104 The bacterial strain *Agrobacterium* sp. IFO 13140 was purchased in lyophilized
105 form from the Institute for Fermentation of Osaka (Japan). Commercial curdlan was
106 acquired from Wako Pure Chemical Industries, Ltd. (Osaka, Japan). All solvents were
107 of analytical grade.

108

109 **Curdlan production by *Agrobacterium* sp. IFO 13140**

110 The culture medium used to reactivate the microorganism was proposed by the
111 supplier (g L^{-1}), pH 7: polypeptone (10), yeast extract (2), $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ (1). 30 mg of
112 the lyophilized bacteria were incubated in 100 mL of the medium at 30 °C and 120 rpm
113 for 48 h. The cells were recovered through centrifugation ($6000 \times g$, 10 min), washed
114 with 9 g L^{-1} NaCl and transferred to the production medium. For curdlan production, the
115 liquid medium described by Martinez et al. [12], pH 7, was used (g L^{-1}): glucose (50),
116 KH_2PO_4 (2.7), NH_4Cl (1.6), MgSO_4 (0.5) and trace elements (10 mL L^{-1}). The
117 composition of the trace elements (g L^{-1}) in $\text{HCl } 0.1 \text{ mol L}^{-1}$ was: $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ (1),
118 $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$ (1), ZnCl (1), CaCl_2 (1) and CaCO_3 (0.03). The reactivated
119 microorganisms were transferred to Erlenmeyer flasks containing 100 mL of the
120 production medium and maintained at 30 °C and 150 rpm for 5 days.

121

122 **Recovery of curdlan from production medium**

123 Two methodologies were used to recover curdlan from the medium. In pre-
124 gelation method [12,13], $\text{NaOH } 3 \text{ mol L}^{-1}$ was added to the Erlenmeyer flasks

125 containing the production medium at a ratio of 1.8:1 (NaOH:medium) for curdlan
126 solubilization. This mixture was centrifuged (18000 \times g, 15 min, 4 °C) to separate the
127 cells. HCl 3 mol L⁻¹ was added to the supernatant until pH 6-7 to obtain the curdlan gel,
128 which was recovered by centrifugation (18000 \times g, 15 min, 4 °C). Subsequently, it was
129 washed three times with distilled water and lyophilized. The commercial curdlan was
130 subjected to the same treatment and it was named pre-gelled commercial curdlan.

131 The precipitation method is an adaptation of the industrial purification of
132 curdlan. Industrially produced curdlan is purified by dissolution in a strong alkaline
133 solution and dried in a spray-dryer, then washed with water until neutralization [14].
134 NaOH 3 mol L⁻¹ was added to the Erlenmeyer flasks containing the production medium
135 at a ratio of 1.8:1 (NaOH: medium) to solubilize the curdlan and the mixture was
136 centrifuged at 18000 \times g for 15 min at 4 °C to separate the cells, before being
137 lyophilized. The dried mixture was resuspended in water, filtered to remove larger
138 impurities and then ultra-filtered using the system described by Fenelon et al. [15] with
139 a 30 kDa membrane and pressure of 1 kgf cm⁻². During ultrafiltration, the material was
140 washed with water until the pH was neutralized, causing the precipitation of the curdlan,
141 that was thereafter lyophilized.

142

143 **Structural analysis of curdlans by FT-Raman**

144 Samples of the commercial and commercial pre-gelled curdlans and those
145 produced by *Agrobacterium* sp. IFO 13140 (pre-gelation and precipitation methods)
146 were analyzed by Fourier transform Raman spectroscopy (FT-Raman) using a Fourier
147 Transform infrared spectrometer (Vertex 70v model with Ram II module, Bruker,
148 Germany) equipped with a Germanium detector cooled with liquid nitrogen. A Nd:YAG

149 laser was used for excitation at 1064 nm. The spectra were based an average of 200
150 scans with a resolution of 4 cm⁻¹.

151

152 **Morphology of curdlans**

153 The morphology of the commercial and commercial pre-gelled curdlans, and the
154 curdlans produced by *Agrobacterium* sp. IFO 13140 (pre-gelation and precipitation
155 methods) was analyzed in a scanning electron microscope (SS-550 model, Superscan,
156 Shimadzu, Japan), at an accelerating voltage of 15 kV.

157

158 **Physicochemical characterization of curdlans**

159 The commercial and commercial pre-gelled curdlans and those produced by
160 *Agrobacterium* sp. IFO 13140 (pre-gelation and precipitation methods) were
161 characterized for their carbohydrate, moisture and sodium content. The carbohydrate
162 content was determined through the phenol-sulfuric method [16]. The moisture content
163 was determined by the gravimetric method [17] and sodium was identified using atomic
164 absorption spectrometry [18].

165

166 **Technological properties of curdlans**

167

168 **Water dispersion and gel formation capacities of curdlans**

169 To verify the dispersal capacity in water of the commercial and commercial pre-
170 gelled curdlans and those produced by *Agrobacterium* sp. IFO 13140 (pre-gelation and
171 precipitation methods), they were subjected to stirring in a mixer and/or in a magnetic

172 stirrer until homogeneous dispersion was achieved. The dispersions were heated at 95
173 °C/1 h to assess the gel formation capacity.

174

175 **Rheological characteristics and gel strength of curdlans**

176 Dispersions of the pre-gelled commercial curdlan and the curdlan produced by
177 *Agrobacterium* sp. IFO 13140 (pre-gelation method) were prepared at three
178 concentrations: 20 g L⁻¹, 40 g L⁻¹ and 80 g L⁻¹ in water. The curdlans were dispersed in
179 water using a mixer at room temperature for 5 min, sonicated for 10 min and then stirred
180 in a magnetic stirrer for 12 h at 40 °C. The samples were analyzed in a controlled stress
181 rotational rheometer (HAAKE MARS II model, Thermo Fisher Scientific Inc.,
182 Newington, Germany), with steel cone/plate geometry (60 mm, gap 0.052 mm). The
183 elastic (G') and viscous (G'') modulus, and the dynamic viscosity were measured
184 depending on temperature (20-60 °C) at a frequency of 10 Hz. For gel strength
185 evaluation, the dispersions of both samples at the concentration of 20 g L⁻¹ were kept or
186 not in a water bath at 61 °C/1 h to prepare the low-set gel and at 95 °C/1 h for
187 preparation of the high-set gel [19]. After heating, they were cooled at room
188 temperature. The strength of the suspensions and gels was evaluated in triplicate in a
189 Stable Micro Systems texturometer (TA-XT Plus model, Texture Technologies Corp.,
190 UK), using a 36 mm probe for compression analysis.

191

192 **Water holding capacity (WHC), Oil holding capacity (OHC) and** 193 **Water solubility index (WSI)**

194 Samples of 0.25 g of the commercial and pre-gelled commercial curdlans and
195 curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-gelation method) were diluted
196 in 10 mL of distilled water or soya oil at 30 °C, homogenized in a magnetic stirrer for

197 30 min and centrifuged (1500 ×g, 10 min). The WHC was expressed as g of water
198 absorbed per g of curdlan sample and the OHC was described as g of oil absorbed per g
199 of curdlan sample. To determine the WSI, the supernatant of the WHC analysis was
200 oven dried at 105 °C and the ratio between the weight of the solid residue present in the
201 supernatant after drying and the weight of the curdlan sample was calculated [11].

202

203 **Application of curdlan in yogurt**

204

205 **Preparation of yogurts**

206 Eight yogurt samples were prepared: A1) without curdlan or heat treatment; A2)
207 without curdlan and with heat treatment; B1) with commercial curdlan and without heat
208 treatment; B2) with commercial curdlan and heat treatment; C1) with pre-gelled
209 commercial curdlan and without heat treatment; C2) with pre-gelled commercial curdlan
210 and heat treatment; D1) with curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-
211 gelation method) without heat treatment; D2) with curdlan produced by *Agrobacterium*
212 sp. IFO 13140 (pre-gelation method) with heat treatment.

213 The curdlan was mixed in UHT (ultra-high temperature) whole milk at a
214 concentration of 15 g L⁻¹ using a mixer. Subsequently, the milk of the samples subjected
215 to heat treatment was heated at 90 °C for 120 seconds and cooled to 42 °C. For samples
216 without heat treatment, the milk was directly heated to 42 °C. Next, all samples were
217 inoculated with 2 g L⁻¹ of mixed culture of *Streptococcus thermophilus* and
218 *Lactobacillus delbrueckii* subsp. *bulgaricus*, and maintained at 42 °C for 6 h. The
219 yogurts were cooled and refrigerated for 24 h for rheological, texture and syneresis
220 analysis [3,20].

221

222 **Texture profile analysis (TPA)**

223 The TPA of the yogurts was carried out in a TA-XT Plus texturometer (Stable
224 Micro Systems, Godalming, UK), equipped with Texture Expert software (Stable Micro
225 Systems, Godalming, UK). A 10 mm cylindrical probe was used (ref. P/0.5R, Stable
226 Micro Systems). Two cycles were applied, at a velocity of 2 mm s^{-1} and a depth of 15
227 mm, producing force-time curves which were used to determine the firmness,
228 cohesiveness, adhesiveness, springiness, gumminess and chewiness of the samples. The
229 analysis was performed in triplicate, at $8 \text{ }^{\circ}\text{C}$.

230

231 **Syneresis**

232 The syneresis analysis of each preparation was performed after 28 days of
233 storage of the yogurt at $8 \text{ }^{\circ}\text{C}$ in triplicate. Yogurt samples (30-40 g) were centrifuged
234 ($222 \times \text{g}$, 10 min, $4 \text{ }^{\circ}\text{C}$) and the supernatant was separated and centrifuged again (222
235 $\times \text{g}$, 10 min, $4 \text{ }^{\circ}\text{C}$), weighed and the syneresis (%) was calculated from the ratio between
236 the mass of the supernatant and the initial mass of the yogurt [20].

237

238 **Rheological analysis**

239 The rheological properties of flow of the yogurts were analyzed in triplicate, in a
240 controlled stress rotational rheometer (HAAKE MARS II model, Thermo Fisher
241 Scientific Inc., Newington, Germany), with steel cone/plate geometry (60 mm, gap
242 0.052 mm). The measurements were performed at $8 \text{ }^{\circ}\text{C}$. The scan of the deformation
243 rate was performed from 0 to 116 s^{-1} , obtaining the outward and return data. The flow
244 and viscosity curves were obtained by determining the stress and viscosity versus the
245 shear rate [20]. The rheological parameters K (consistency index) and n (flow behavior
246 index) were calculated using the Ostwald de Waele model, while the parameter τ_0 (yield

247 stress) was calculated by the Herschel-Bulkley model. The hysteresis was calculated
248 from the area of the flow curves of the yogurts using the RheoWin 4.10.000 software
249 program (HAAKE software, Thermo Fisher Scientific Inc., Newington, Germany).

250

251 **Statistical analysis**

252 Data were analyzed by analysis of variance (ANOVA), and means were
253 compared with the Tukey Test ($p < 0.05$) using the Statistica 8.0/2008 software package
254 (Stat Soft, Inc., Tulsa, USA).

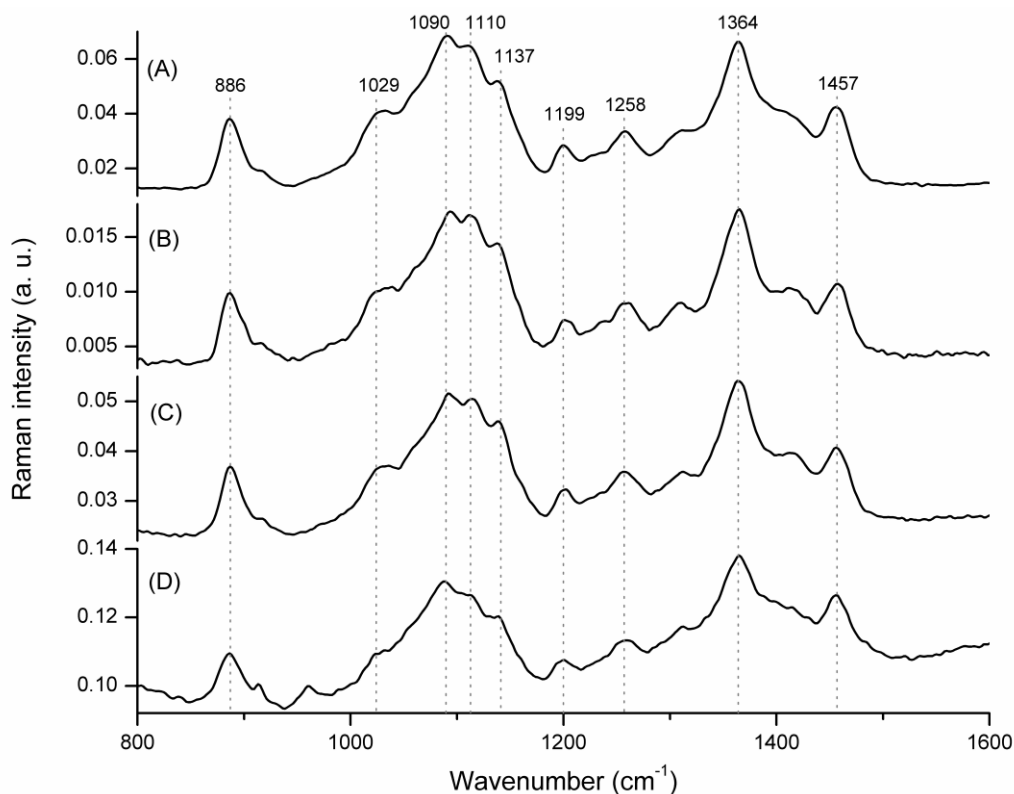
255

256 **Results and Discussion**

257

258 **Structural analysis of curdlans by FT-Raman**

259 Fig 1 shows FT-Raman spectra of commercial and pre-gelled commercial
260 curdlans and those produced by *Agrobacterium* sp. IFO 13140 (pre-gelation and
261 precipitation methods). From the data, it is verified a structural similarity between all
262 samples, which means that the curdlan produced by the microorganism has a similar
263 structure to commercial curdlan and it also means that the method employed to recover
264 curdlan from the medium did not modify the structure of the polysaccharide.



265

266 **Fig 1. FT-Raman spectra of: (A) commercial curdlan, (B) pre-gelled commercial**
 267 **curdlan, (C) curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-gelation**
 268 **method), (D) curdlan produced by *Agrobacterium* sp. IFO 13140 (precipitation**
 269 **method). The dotted lines show the characteristic peaks of the samples.**

270

271 Several authors described the chemical bonds present in the structure of curdlan
 272 molecule based in FT-IR analysis [3,14,21]. They stated that, from the FT-IR data for
 273 curdlan, the bands around 890, 1080 and 1160 cm^{-1} correspond to the β -(1,3)-glucan
 274 linkages. Despite the difference in band assignments for FT-IR and FT-Raman
 275 spectroscopies, the spectra of carbohydrates in both techniques show a characteristic
 276 absorption band of β -anomeric configuration at $\sim 890 \text{ cm}^{-1}$ (886 cm^{-1} in this work) [22].
 277 The Raman bands and shoulders at 1090 and 1137 cm^{-1} are typical for β -glucans.
 278 Intense highly overlapped Raman bands between 990 and 1200 cm^{-1} are attributed to
 279 COC and CC stretching vibrations of polysaccharides. The features between 1200 and

280 1440 cm^{-1} are mainly assigned to in-plane ring deformation including CH and OH
281 bending modes. Finally, the band at 1457 cm^{-1} is assigned to CH_2 in-plane bending in
282 CH_2OH of the molecule [23].

283

284 **Morphology of curdlans**

285 Fig 2 shows scanning electron microscopy images of the commercial curdlan
286 (Figs 2A and 2B), the pre-gelled commercial curdlan (Figs 2C and 2D) and the curdlans
287 produced by *Agrobacterium* sp. IFO 13140 recovered by pre-gelation and precipitation
288 methods (Figs 2E-2H) with a range of magnifications.

289

290

291

292

293

294

295

296

297

298

299

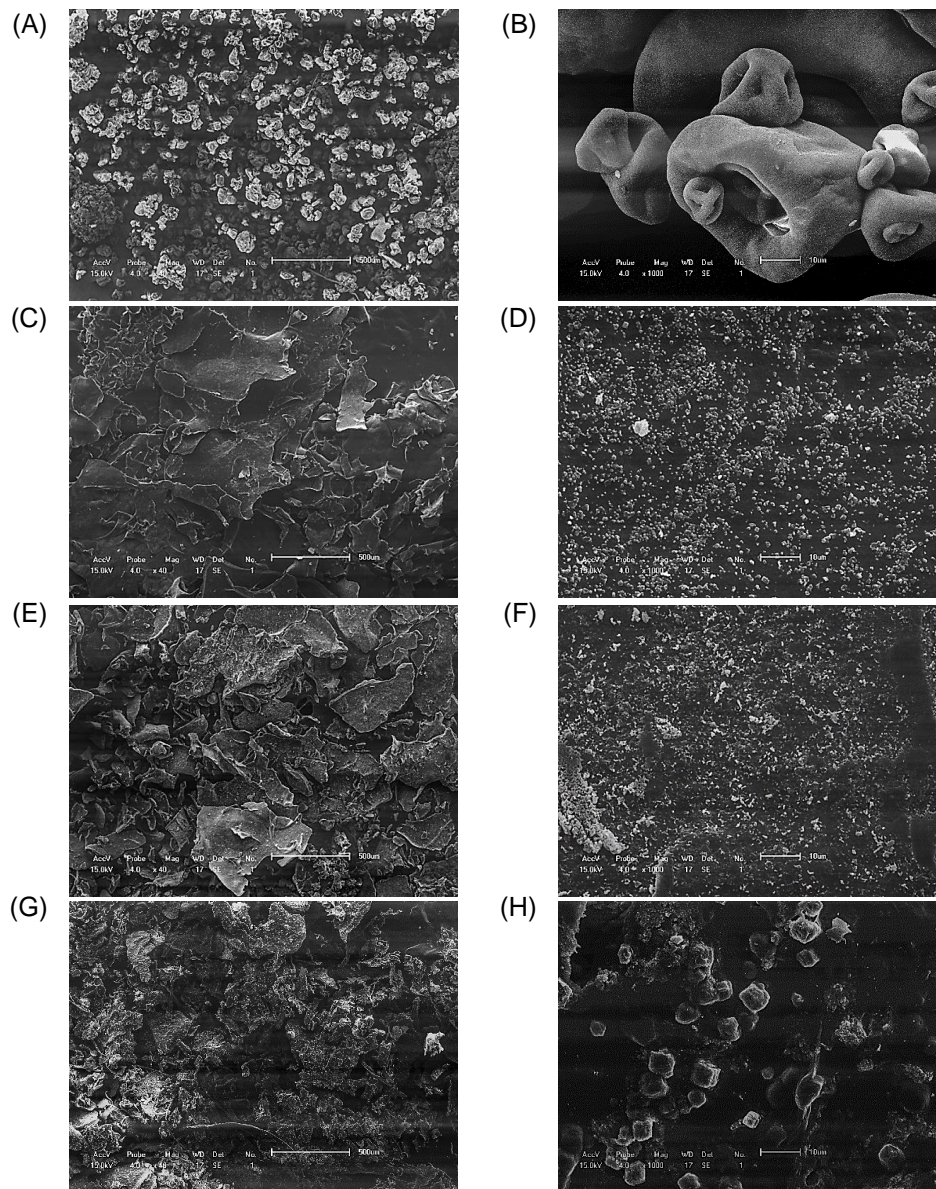
300

301

302

303

304



305

306

307 **Fig 2. Scanning electron microscopy of: commercial curdlan – (A) 40x and (B)**
 308 **1000x magnification; pre-gelled commercial curdlan – (C) 40x and (D) 1000x**
 309 **magnification; curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-gelation**
 310 **method) – (E) 40x and (F) 1000x magnification; curdlan produced by**
 311 ***Agrobacterium* sp. IFO 13140 (precipitation method) – (G) 40x and (H) 1000x**
 312 **magnification.**

313

314 Marchessault and Deslandes [24] define the shape of the granules of commercial
315 curdlan as collapsed or invaginated, which coincides with the commercial curdlan shape
316 illustrated in Figs 2A and 2B. The granules have widely differing sizes, varying from
317 around 10 to 100 μm in diameter. Industrially produced curdlan is dried by spray-drying
318 [14] and the type of drying has a major influence on the particle bead structure [25]. The
319 expected range of particle size of polymers dried by ordinary spray dryers is 5-150 μm .
320 Furthermore, spherical microspheres or those with pores/concavities, as shown in Figs
321 2A and 2B, are characteristic of the drying of products by a spray-dryer. The formation
322 of pores or concavities is associated with the rapid evaporation of the liquid particles of
323 this process [26].

324 The pre-gelled commercial curdlan and both the curdlans produced by the
325 microorganism, when viewed at low magnification (40x), were in flake form and
326 displayed irregularities. These characteristics are expected for lyophilized products,
327 since the working conditions of the lyophilizer exert great pressure on the particles to be
328 dried, meaning that a product dried by lyophilization remains amorphous in comparison
329 with a spray dried product [25]. Furthermore, at a higher magnification (1000x), it can
330 be seen that the particles obtained for the three samples that underwent the
331 lyophilization process (Figs 2D, 2F and 2H), were considerably smaller than the
332 commercial curdlan (obtained by spray-dryer). Therefore, it can be inferred that the
333 lyophilization process can facilitate the dispersion of curdlan molecules in water,
334 especially the pre-gelled commercial curdlan and the one produced by *Agrobacterium*
335 sp. IFO 13140 (pre-gelation method), which have smaller particle sizes (Figs 2D and
336 2F). The smaller diameter increases the accessibility of water molecules to the inside of
337 the particles, facilitating dispersion.

338 The structure displayed by the commercial curdlan is maintained by a large
339 amount of hydrogen bonds. But, when added to an alkaline solution, these bonds are
340 broken due to their ionization, and the granule loses its structure. The neutralization of
341 the curdlan suspension causes a reshaping of the hydrogen bonds that depends on the
342 neutralizing agent used. With the addition of HCl (pre-gelation method), the
343 reassociation causes the formation of a firm gel, with small structures of less than 1 μm
344 in diameter (Figs 2D and 2F). With water (precipitation method), reassociation causes
345 the precipitation of the curdlan into small particles with diameters greater than those
346 obtained by neutralization with HCl, ranging from 2-10 μm in diameter (Fig 2H).
347 Therefore, the difference between the sizes of the curdlan particles produced by the
348 microorganism using the two methods is due to the different forms of reassociation of
349 the hydrogen bonds, which are dependent on the recovery method employed, and can
350 also influence the characteristics of water dispersion of the polysaccharide.

351

352 **Physicochemical characterization of curdlans**

353 As an exopolysaccharide, curdlan is secreted to the extracellular medium in the
354 form of biofilms, and has the advantage of being easy to recover. As a result, the
355 polysaccharide should present a low degree of impurities, and consequently the
356 carbohydrate content represents an indirect measure of the purity of the samples. The
357 levels of carbohydrates and moisture found in the curdlan samples are shown in Table 1.

358

359

360

361

362 **Table 1. Carbohydrate, moisture and sodium content (%) of different samples of**
 363 **curdlan. Values indicate mean \pm standard-deviation.**

Curdlan sample	Carbohydrate	Moisture	Sodium
Commercial	94.5 \pm 0.2	4.1 \pm 0.4	1.2 \pm 0.3
Commercial pre-gelled	82 \pm 1	4.0 \pm 0.3	4.4 \pm 0.8
Microbial* (pre-gelation method)	70.6 \pm 0.6	5.4 \pm 0.9	0.024 \pm 0.001
Microbial* (precipitation method)	85.4 \pm 0.7	4.8 \pm 0.6	< 0.001

364 *Produced by *Agrobacterium* sp. IFO 13140.

365

366 The highest carbohydrate content was found for the commercial sample. The
 367 pre-gelled commercial sample had a carbohydrate content lower than the commercial
 368 curdlan due to the salt (NaCl) incorporated in the pre-gelling process. The two samples
 369 produced by *Agrobacterium* sp. IFO 13140 had different carbohydrate contents, with
 370 that produced by precipitation method having greater purity due to the successive
 371 washings performed. The impurities in the last samples consist of remnants of the
 372 production medium and the salt formed in the recovery step.

373 The sodium content of curdlans was determined because curdlan is a glucose
 374 polymer with a low amount of inorganic salts, mainly sodium chloride [27]. Thus,
 375 sodium content is an indirect estimate of the salt content of the material. The sodium
 376 concentrations disposed in Table 1 directly influence the dispersion and gelling
 377 characteristics of the polysaccharide, which will be further discussed in the next section.

378

379

380

381

382 **Technological properties**

383

384 **Water dispersibility and gelling capacities**

385 Curdlan can be used as a thickener, stabilizer and texturizer in food industry
386 [8,27]. However, as curdlan is insoluble in water, it is necessary to use an efficient
387 homogenizer to obtain a homogeneous dispersion and apply a further stirring of the
388 dispersion prior to the carrying out of analyses requiring uniformity [21].

389 The pre-gelled commercial curdlan and the curdlan produced by *Agrobacterium*
390 sp. IFO 13140 (pre-gelation method) dispersed easily in water when subjected to stirring
391 in a mixer for less than 5 min, and also acted as thickening agents. Both dispersions
392 remained visibly homogeneous for days and, when heated at 95 °C/1 h, formed a firm
393 and homogeneous gel. The commercial curdlan did not easily disperse in water using
394 the mixer. After 24 hours of stirring in a magnetic stirrer, a homogeneous dispersion
395 formed for a brief period, displaying a phase separation after 10 min of rest. After 48 h
396 of stirring, phase separation began to be observed at 30 min; after 72 h, at 1 hour; and
397 after 96 h, phase separation began at 1.5 h. The gelling of commercial curdlan also
398 occurred after heating at 95 °C/1 h, but there was phase separation in the gel formed,
399 which became more pronounced as the stirring period for formation of the dispersion
400 was reduced. Marchessault and Deslandes [24] have previously identified the formation
401 of a non-homogenous gel from curdlan in its native form. Probably, the difficulty of
402 dispersion of commercial curdlan compared with those submitted to pre-gelation
403 process is related to the particle size of the sample (as noted in morphological analysis),
404 which makes difficult the accessibility of water molecules to the inside of the particles.

405 Another important factor related to the dispersion and gelling of curdlan is the
406 presence of salts. In the gelling of curdlan, the swelling phase of the molecule is

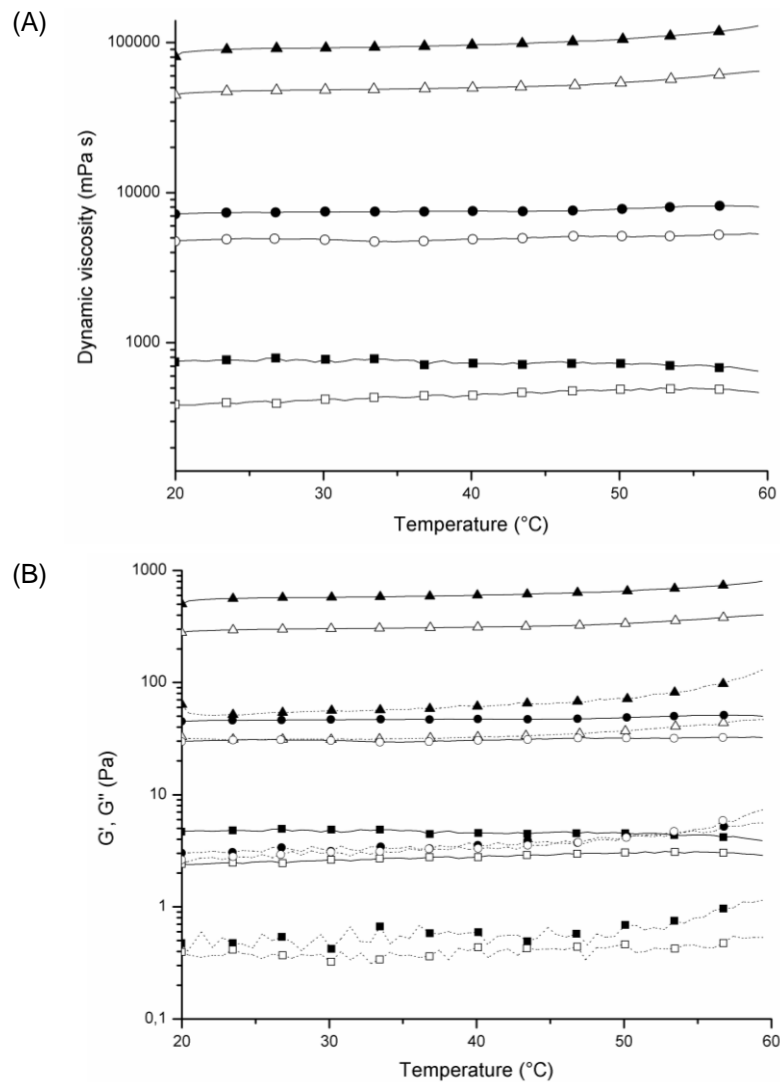
407 essential and may be influenced by salts, as they affect the mobility of the water
408 molecules during hydration [10]. The swelling of curdlan is facilitated with the salts
409 present because it increases intermolecular association and consequently increases the
410 viscosity of curdlan water dispersions. The curdlan produced by *Agrobacterium* sp. IFO
411 13140 (precipitation method) displayed greater difficulty in forming a homogenous
412 dispersion in water and, in contrast to the other curdlans, did not form a gel when
413 heated. This can be explained by its considerably lower sodium content, due to the
414 countless washings with water used to achieve neutralization. Samples of the other
415 curdlans were also washed in the ultrafiltration device and, following such process, no
416 longer had the same dispersion and gelling capacities. This proves therefore, that as with
417 other biopolymers, for curdlan, the mechanical characteristics depend excessively on the
418 methods of recovery employed following production. Due to the characteristics
419 displayed, the technological characteristics of the curdlan produced by *Agrobacterium*
420 sp. IFO 13140 (precipitation method) were not evaluated, and this sample was not
421 applied in yogurt. Additionally, dispersions of commercial curdlan were not evaluated
422 by rheological and gel strength analysis because the commercial curdlan did not act as a
423 thickener with the methodology employed for preparation of the dispersions.

424

425 **Rheological characteristics and gel strength of curdlans**

426 Thermal scanning rheological measurements were made to evaluate both the
427 dynamic viscosity and the gelation behavior of curdlan dispersions. The temperature
428 dependence of dynamic viscosity and of G' and G'' modulus for aqueous dispersions of
429 pre-gelled commercial curdlan and of the curdlan produced by *Agrobacterium* sp. IFO
430 13140 (pre-gelation method) at 20, 40 and 80 g L⁻¹ are shown in Figs 3A and 3B.

431



432

433 **Fig 3. Temperature dependency of: (A) dynamic viscosity and (B) G' (continuous**
 434 **line) and G'' (dotted line) modulus of the aqueous dispersions of curdlan. Pre-gelled**
 435 **commercial curdlan (empty symbol) and curdlan produced by *Agrobacterium* sp. IFO**
 436 **13140 recovered by the pre-gelation method (full symbol) at (■) 20, (●) 40 and (▲) 80**
 437 **g L⁻¹.**

438

439 The curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-gelation method)
 440 exhibited considerably higher viscosity than pre-gelled commercial curdlan (Fig 3A).
 441 For example, at 20 °C, the dynamic viscosity of curdlan produced by the microorganism
 442 was almost 100% higher than pre-gelled commercial curdlan at 20 and 80 g L⁻¹.

443 Equally, G' and G'' modulus (Fig 3B) were also higher for the curdlan produced by the
444 microorganism, revealing the highest thickening potential of this curdlan when
445 compared to the pre-gelled commercial one.

446 Both the dynamic viscosity and the G' module of both curdlans exhibited a
447 different behavior as a function of temperature for each concentration employed. At 20
448 g L^{-1} , both parameters increased with temperature for pre-gelled commercial curdlan,
449 while for curdlan produced by the microorganism they decreased. This behavior is
450 related to the swelling of curdlan because of the breakage of hydrogen bonds during
451 heating. The difference between the curdlans is because, for pre-gelled commercial
452 curdlan, inter-molecular entanglements between the particles that were swollen formed
453 pseudo-links, which contributed to the little increase in G' . This phenomenon did not
454 occur for curdlan produced by the microorganism. However, for higher concentrations
455 (40 and 80 g L^{-1}), the decrease in the viscosity and in G' no longer occurred with
456 temperature for any sample, because the high concentration favors the inter-molecular
457 entanglements between the particles, which contribute to the increase of both
458 parameters. Additionally, the increase in G' and G'' modulus and in dynamic viscosity of
459 curdlan dispersions at about 50-55 °C, especially at 80 g L^{-1} , suggests the beginning of
460 the formation of the thermo-irreversible gels due to hydrophobic interactions.

461 Jin et al. [21] stated that G' decreased until 50 °C for curdlan suspensions at 20 g
462 L^{-1} , which is related to the swelling of curdlan because of the breakage of hydrogen
463 bonds, as observed for curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-gelation
464 method) in this study. However, Funami et al. [28] stated only increasing in G'
465 parameter for dispersions of curdlans at 20, 40 and 100 g L^{-1} with increasing
466 temperature. The authors attributed this behavior to inter-molecular entanglements
467 between the particles that were swollen forming pseudo-links, being this result similar

468 to the one obtained for pre-gelled commercial curdlan in all concentrations evaluated and
 469 for curdlan produced by the microorganism for concentrations from 40 to 80 g L⁻¹ in
 470 this work.

471 The strength data of suspensions and gels of the pre-gelled commercial curdlan
 472 and the curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-gelation method) at 20
 473 g L⁻¹ are presented in Table 2. According to the strength values of the suspensions and
 474 gels, the curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-gelation method) has a
 475 better thickening capacity than the pre-gelled curdlan, as the suspension of the same
 476 without heat treatment was approximately 10% stronger. This result agrees with the
 477 results of rheological analysis. However, the gelling capacity of the pre-gelled
 478 commercial curdlan is considerably higher.

479

480 **Table 2. Strength ($\times 10^{-3}$ N) of pre-gelled commercial curdlan and the curdlan**
 481 **produced by *Agrobacterium* sp. IFO 13140 (pre-gelation method) samples after**
 482 **undergoing different heat treatments. Values indicate mean \pm standard-deviation.**

Heat treatment of the dispersion in a concentration of 20 g L ⁻¹	Strength ($\times 10^{-3}$ N)	
	Pre-gelled commercial curdlan	Curdlan from microorganism (pre-gelation method)
Without heat treatment	69 \pm 2 ^d	76 \pm 2 ^c
Low-set gel (61 °C/1 h)	79 \pm 1 ^c	77 \pm 2 ^c
High-set gel (95 °C/1 h)	96.8 \pm 0.8 ^a	83 \pm 1 ^b

483 ^{a-d} Means with different letters are significantly different (p < 0.05).

484

485 The strength of the low-set gel of the pre-gelled commercial curdlan was around
 486 15% greater than that of the suspension without heat treatment, and statistically equal to
 487 that of the low-set gel of the curdlan produced by the microorganism. The high-set gel

488 of the pre-gelled commercial curdlan had strength 40% greater than the suspension
489 without heat treatment, and 17% greater than the high-set gel produced by the
490 microorganism.

491 When an aqueous dispersion of curdlan is heated around 55 °C, the low-set gel
492 formed is maintained by intramolecular hydrogen bonds and the curdlan chains adopt a
493 predominantly single helix conformation. But when heated above 80 °C, the high-set gel
494 formed is maintained by intermolecular hydrophobic interactions and the curdlan chains
495 adopt a predominantly triple helix conformation [14,29], forming an organized and rigid
496 gel configuration. As such, the increase in the strength of the high-set curdlan gel (95
497 °C/1 h) is due to the greater presence of the triple helix conformation.

498 The fact that the curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-
499 gelation method) have formed a considerably weaker gel than the pre-gelled commercial
500 curdlan, as well as the fact that it does not exhibit the same behavior that pre-gelled
501 commercial sample in rheological analysis at low concentrations (increase of G' and
502 dynamic viscosity with temperature), are probably due to a difference of molecular
503 weight/degree of polymerization of the polymers. A variety of physical properties of β -
504 (1 \rightarrow 3)-glucans, including gel strength (or gel-forming ability) is related to the molecular
505 weight/degree of polymerization of the biopolymer. The higher the degree of
506 polymerization of the polysaccharide, the greater is its gel forming ability with heating
507 [30,31].

508 Nakanishi et al. [32] studied the formation of a complex of curdlan with aniline
509 blue dye and noted a relation between the rate of interaction of the polymer with the dye
510 and the concentration, degree of polymerization, and gel-forming ability of the polymer.
511 The variation in absorbance at 590 nm is proportional to the concentration of curdlan
512 and to its gel-forming ability (and consequently to its degree of polymerization). S1

513 Appendix contains the methodology used by the authors described to evaluate curdlan
514 interaction with aniline blue dye and the results of the relationship between absorbance
515 variation and concentration of the pre-gelled commercial curdlan and curdlan produced
516 by the microorganism. By comparing the results of this study to those obtained by
517 Nakanishi et al. [32] and using the relation obtained by these authors between
518 absorbance variation and degree of polymerization, it is concluded that the degree of
519 polymerization of the pre-gelled commercial curdlan is about 45% higher than of the
520 curdlan produced by the microorganism, which explains the lowest gel strength and the
521 decrease in dynamic viscosity and G' in low concentration dispersions of the polymer.

522

523 **Water holding capacity (WHC), Oil holding capacity (OHC) and**
524 **Water solubility index (WSI)**

525 Determining the technological properties of polysaccharides is of great
526 importance when predicting their possible industrial applications. The values for the
527 properties of water solubility index (WSI) water holding capacity (WHC) and oil
528 holding capacity (OHC) of the curdlan samples are described in Table 3.

529

530

531

532

533

534

535

536

537 **Table 3. Water holding capacity (WHC), oil holding capacity (OHC) and water**
 538 **solubility index (WSI) (g g^{-1}) of the commercial, pre-gelled commercial curdlans**
 539 **and the one produced by *Agrobacterium* sp. IFO 13140 (pre-gelation method).**
 540 **Values indicate mean \pm standard-deviation.**

Curdlan sample	WHC	OHC	WSI
Commercial	4.6 ± 0.4^a	0.62 ± 0.08^c	0.006 ± 0.002^a
Commercial pre-gelled	2.20 ± 0.08^c	4.4 ± 0.2^b	$0.00923 \pm 7\text{E-}5^a$
Microbial (pre-gelation method)	3.6 ± 0.3^b	8.7 ± 0.1^a	0.0068 ± 0.003^a

541 ^{a-c} Means in the same column with different letters are significantly different ($p < 0.05$).

542

543 The three curdlans had very low and statistically equal WSIs, which was
 544 expected as curdlan is insoluble in water. Despite this, however, the curdlans presented
 545 some water absorption, with the commercial curdlan achieving the highest value,
 546 followed by the curdlan produced by the microorganism and the pre-gelled commercial
 547 curdlan. Seguchi and Kusunose [33] found water absorption rates for curdlan between
 548 5.244 g g^{-1} and 7.724 g g^{-1} .

549 Compared to other polysaccharides employed in food industry as gums, curdlan
 550 has low water holding capacity; the guar and xanthan gums have a WHC of 25.77 g g^{-1}
 551 and 27.33 g g^{-1} , respectively [34]. However, both are soluble in water. The low
 552 solubility and water holding capacity of curdlans is explained by the existence of a large
 553 amount of intra/intermolecular hydrogen bonds within the polymer. This also explains
 554 the fact that commercial curdlan has a higher water absorption index than the other
 555 types as the recovery by the pre-gelation process employed in the latter two types favors
 556 the formation of large quantities of hydrogen bonds in the polymer. Thus, the
 557 polysaccharide interacts more strongly with itself than with water [21,34].

558 The three evaluated curdlans also differed with respect to oil holding capacity.
559 With the exception of the commercial curdlan, they presented greater OHC values than
560 WHC, indicating that the samples have a higher lipolytic than hydrophilic capacity. Oil
561 absorption values greater than three make curdlan a potentially useful ingredient in
562 structural interactions in foods, especially in aroma retention, improved palatability and
563 maintenance of the stability of the product during storage [11]. This result justifies the
564 use of curdlan as a fat replacer or mimicker of fat in the food industry [9,35].

565

566 **Application of curdlan in yogurt**

567 The main characteristics that define the quality of yogurts are its texture and
568 propensity for serum separation (syneresis). Typically, polysaccharides such as xanthan,
569 guar, gellan, pectin, carrageenan are used to give the product a firmer texture, increase
570 its stability and hence make it more acceptable to the consumer [3,20]. Table 4 displays
571 the parameters of texture and syneresis after 28 days of storage of yogurt samples with
572 and without curdlan, submitted or not to heat treatment with the aim of gelling the
573 curdlan in the milk prior to the fermentation process.

574

575

576

577

578

579

580

581 **Table 4. Texture parameters and syneresis of yogurts with and without curdlan**
 582 **and submitted or not to heat treatment. Values indicate mean \pm standard-**
 583 **deviation.**

Sample	Firmness ($\times 10^{-3}$ N)	Cohesiveness	Adhesiveness (N \times mm)	Springiness (mm)	Gumminess (N)	Chewiness (N)	Syneresis (%)
A1	77 \pm 2 ^c	0.77 \pm 0.02 ^a	0.02 \pm 0.01 ^c	1.13 \pm 0.01 ^{ab}	5.9 \pm 0.3 ^c	6.7 \pm 0.4 ^{bcd}	49.3 \pm 0.8 ^a
A2	84 \pm 8 ^c	0.71 \pm 0.07 ^{ab}	0.04 \pm 0.03 ^c	1.00 \pm 0.04 ^{bc}	5.9 \pm 0.1 ^c	5.9 \pm 0.3 ^{cde}	48.7 \pm 0.4 ^a
B1	83 \pm 8 ^c	0.66 \pm 0.02 ^{bc}	0.08 \pm 0.03 ^c	0.94 \pm 0.04 ^c	5.5 \pm 0.7 ^c	5.1 \pm 0.7 ^e	45.6 \pm 0.9 ^b
B2	129 \pm 7 ^b	0.47 \pm 0.04 ^d	0.4 \pm 0.1 ^b	0.97 \pm 0.05 ^{bc}	6.1 \pm 0.2 ^{bc}	5.9 \pm 0.4 ^{cde}	38.2 \pm 0.3 ^c
C1	94 \pm 9 ^c	0.65 \pm 0.02 ^{bc}	0.16 \pm 0.04 ^c	1.2 \pm 0.1 ^a	6.0 \pm 0.6 ^{bc}	7.1 \pm 0.3 ^{bc}	26.4 \pm 0.8 ^d
C2	125.1 \pm 0.6 ^b	0.58 \pm 0.02 ^c	0.58 \pm 0.02 ^{ab}	1.01 \pm 0.01 ^{bc}	7.2 \pm 0.2 ^b	7.3 \pm 0.2 ^b	29.2 \pm 0.7 ^d
D1	128 \pm 6 ^b	0.46 \pm 0.02 ^d	0.68 \pm 0.04 ^a	0.96 \pm 0.03 ^c	5.8 \pm 0.3 ^c	5.6 \pm 0.5 ^{de}	28.8 \pm 0.2 ^d
D2	160 \pm 7 ^a	0.59 \pm 0.01 ^c	0.5 \pm 0.1 ^{ab}	0.99 \pm 0.06 ^{bc}	9.5 \pm 0.6 ^a	9.3 \pm 0.3 ^a	35.8 \pm 0.9 ^c

584 ^{a-c} Means within the same column with different letters are significantly different ($p < 0.05$).

585 Yogurts: A1) without curdlan or heat treatment; A2) without curdlan and with heat treatment;
 586 B1) with commercial curdlan and without heat treatment; B2) with commercial curdlan and heat
 587 treatment; C1) with pre-gelled commercial curdlan and without heat treatment; C2) with pre-gelled
 588 commercial curdlan and heat treatment; D1) with curdlan produced by *Agrobacterium* sp. IFO 13140
 589 (pre-gelation method) without heat treatment; D2) with curdlan produced by *Agrobacterium* sp. IFO
 590 13140 (pre-gelation method) with heat treatment.

591

592 It can be seen that the use of curdlan produced by *Agrobacterium* sp. IFO 13140
 593 (pre-gelation method) produced significant alterations in the parameters of firmness,
 594 cohesiveness and adhesiveness of the yogurt, even without being subjected to heat
 595 treatment. This is due to the thickening potential of the material. Heat treatment caused
 596 major changes in these parameters, both for the yogurts with commercial curdlan and
 597 pre-gelled commercial curdlan, although the highest values for firmness, gumminess
 598 and chewiness were obtained for the yogurt with curdlan produced by the
 599 microorganism. Heat treatment promotes the formation of a firm gel due to the
 600 intermolecular hydrophobic interactions that structure the system, making the yogurt

601 more consistent, resulting in greater difficulty to separate in the mouth and making it
602 denser during chewing, which therefore takes longer.

603 It was notable that the pre-gelled commercial curdlan formed a stronger gel in
604 water, but had less effect on the texture parameters than that produced by
605 *Agrobacterium* sp. IFO 13140 (pre-gelation method) when applied to the yogurt, which
606 was not expected. Thus, it is likely that the latter has a greater ability to interact with
607 water and the other components of milk, especially proteins, more efficiently stiffening
608 the protein network formed after the fermentation process. As described before, the
609 curdlan produced by the microorganism has the capacity to absorb 64% more water and
610 98% more oil than the pre-gelled commercial curdlan, with these parameters being
611 important for products such as yogurts prepared with whole milk (3% fat).

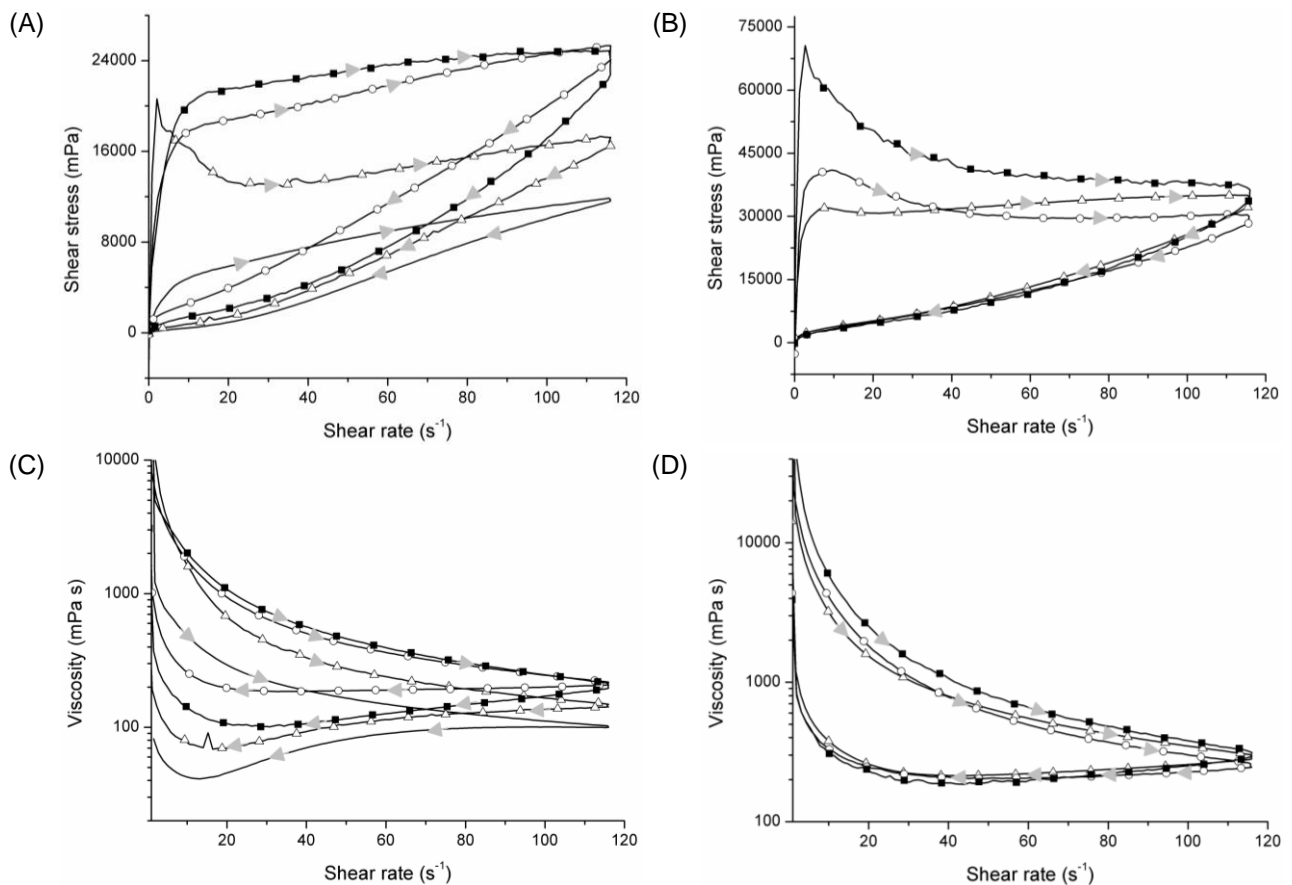
612 The use of commercial curdlan without heat treatment did not produce many
613 changes in the texture parameters or in the syneresis of the yogurts. This is because it
614 was not effectively homogenized in the milk as it dispersed with greater difficulty,
615 interfering little in the formation of the protein network during fermentation.

616 It was found that syneresis was reduced for the yogurt samples with curdlan,
617 particularly the pre-gelled commercial variety and the curdlan produced by
618 *Agrobacterium* sp. IFO 13140 (pre-gelation method). The latter displayed higher
619 syneresis after the first 28 days of storage because its gel is less stable than the pre-
620 gelled commercial curdlan. As such, the pre-gelled commercial curdlan demonstrated a
621 greater ability to preserve the structure of the yogurt, avoiding rearrangements in the
622 casein network due to the time of storage and avoiding whey expulsion. Martinez et al.
623 [3] produced yogurts with added curdlan and achieved a significant reduction in the
624 syneresis of the products compared to the control (without curdlan). The authors found
625 that the high stability obtained by the yogurt with curdlan is attributed to interactions

626 between the curdlan molecules, either between one another or with the proteins, which
 627 cause the formation of a more compact and continuous three-dimensional protein
 628 network, which effectively entraps the protein molecules and water in its structure.

629 Importantly, there was no difference between any of the texture parameters and
 630 the syneresis of yogurts without curdlan submitted or not to heat treatment. This was
 631 expected as the aim of treatment was to gel the curdlan and, therefore, the A2 sample
 632 was not submitted to rheological analysis. Fig 4 shows the flow curves (Figs 4A and
 633 4B) and the viscosity curves (Figs 4C and 4D) for the yogurts produced with curdlan
 634 with and without heat treatment of the milk prior to fermentation.

635



636

637 **Fig 4. Flow curves of yogurt samples without heat treatment (A) with heat**
 638 **treatment (B); viscosity curves of yogurt samples without heat treatment (C) and**

639 **with heat treatment (D)**. Yogurt without curdlan (–), with commercial curdlan (Δ),
640 with pre-gelled commercial curdlan (\circ) and with curdlan produced by *Agrobacterium*
641 sp. IFO 13140 (pre-gelation method) (\blacksquare). The direction of the gray arrows indicates the
642 ascendant and descendant curves.

643

644 All the yogurts behaved as non-Newtonian pseudo-plastic fluids, as their
645 viscosity decreased due to the shear rate applied. It can be seen that the samples
646 subjected to heat treatment of the milk had higher viscosity values across the entire
647 shear rate analyzed due to the gelling of the curdlan. However, the decrease in viscosity
648 versus the shear rate in the return data was much more significant for these samples. All
649 the samples also exhibited thixotropic characteristics, due to the difference of tension
650 and viscosity between the ascending and descending shear rate curves. The hysteresis
651 observed is due to the breakdown of the gel structure formed by coagulation of the
652 protein during fermentation, in the presence and absence of curdlan. Hysteresis is
653 measured as the area between the ascending and descending curves, where the greater
654 the area (when positive), the greater the thixotropic effect. Hysteresis, together with the
655 other rheological parameters, are described in Table 5.

656

657

658

659

660

661

662 **Table 5. Rheological parameters of yogurt samples with and without curdlan and**
 663 **submitted or not to heat treatment. Hysteresis, consistency index (K), flow**
 664 **behavior index (n) and yield stress (τ_0). Values indicate mean \pm standard-deviation.**

<i>Sample</i>	<i>K (mPa sⁿ)</i>	<i>n</i>	<i>τ_0 (Pa)</i>	<i>Hysteresis (Pa s⁻¹)</i>
A	98 \pm 3 ^d	0.997 \pm 0.006 ^a	2 \pm 1 ^c	362 \pm 43 ^e
B1	223 \pm 26 ^d	0.938 \pm 0.001 ^b	12 \pm 1 ^d	1088 \pm 75 ^d
B2	1685 \pm 18 ^a	0.662 \pm 0.004 ^e	30 \pm 4 ^c	2241 \pm 39 ^b
C1	922 \pm 35 ^b	0.73 \pm 0.04 ^d	12 \pm 1 ^d	1134 \pm 92 ^d
C2	--*	--*	36 \pm 2 ^b	2200 \pm 49 ^b
D1	467 \pm 56 ^c	0.87 \pm 0.02 ^c	10.9 \pm 0.5 ^d	1623 \pm 36 ^c
D2	--*	--*	75 \pm 3 ^a	3461 \pm 10 ^a

665 ^{a-c} Means within the same column with different letters are significantly different (p< 0.05).

666 *It was not possible to calculate the parameters by the Oswald de Waele model.

667

668 In Fig 4B, the heat-treated samples of yogurt with pre-gelled commercial curdlan
 669 and curdlan produced by the microorganism displayed a peak in shear strain at the
 670 beginning of the growth of the deformation rate. This peak is related to the lack of
 671 homogeneity of the product, because during the gelling process of curdlan small lumps
 672 are formed, which are not necessarily noticeable to the palate, but which alter rheology.
 673 With the presence of these lumps, the tension needed for a small deformation is high,
 674 and as a result, it was not possible to calculate the K and n parameters for samples C2
 675 and D2 by the Oswald de Waele model.

676 From the data of Table 5 it is clear that the higher viscosity samples in Figs 4C
 677 and 4D, which are those subjected to heat treatment, had more evident thixotropic
 678 characteristics (higher hysteresis), as they underwent a major reduction in dynamic
 679 viscosity with time, in a rate constant with shearing. This is due to the breakdown of the
 680 organized yogurt structure when submitted to a determined strain. After remained
 681 stationary, the samples returned to their original state more quickly, but with lower

682 viscosity during the return, being this recovery dependent on time. Therefore, the gel
683 formed with heat treatment presented low stability, in particular the gel of the curdlan
684 produced by the microorganism, corroborating with the syneresis data of the yogurt.

685 Curiously, it was observed that of the samples without heat treatment, that which
686 had the greater viscosity in the increasing shear rate curve did not present higher K and
687 n values, or in other words, the sample with pre-gelled commercial curdlan was greater
688 than the curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-gelation method) in the
689 consistency index and flow behavior. This was due to the sharp drop in viscosity in the
690 decreasing curve of the latter, revealing again the lower stability of the gel formed.

691 The results of the yield strength parameter (τ_0) were consistent with the firmness
692 of the yogurt, as the yogurt with curdlan produced by *Agrobacterium* sp. IFO 13140
693 with thermic treatment withstood the greatest tension before suffering deformation,
694 followed by the yogurts with pre-gelled commercial curdlan and commercial curdlan
695 with heat treatment. It is important to note that yield stress refers to the maximum stress
696 that the material can bear before yielding (in the elastic deformation regime) and that the
697 greatest τ_0 values are those of the firmer samples. These samples displayed a minimum
698 stress for deformation that was more difficult to break due to the increased organization
699 and stiffening of the protein network formed, originating from the intermolecular
700 hydrophobic interactions of the curdlan with heating.

701

702 **Conclusions**

703 The characteristics of dispersion, gelation and the rheological properties of
704 curdlan depended greatly on the recovery methods employed after its production, as the
705 curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-gelation method) and the
706 commercial pre-gelled curdlan dispersed better in water, acted as thickeners and formed

707 more homogeneous gels. The use of the pre-gelation method exerts a major influence on
708 the size of the particles obtained and in the presence of NaCl, which contributes
709 significantly to the dispersion and gelation characteristics described. However, the
710 recovery method employed did not influence the structure of the polysaccharide. Even
711 the pre-gelled commercial curdlans and those produced by the microorganism through
712 the pre-gelation method had significantly different technological properties. The curdlan
713 produced by *Agrobacterium* sp. IFO 13140 (pre-gelation method) showed a greater
714 thickening and water and oil holding capacity than the pre-gelled commercial curdlan,
715 while the latter demonstrated a considerably greater gelling capacity, that is related to
716 degree of polymerization of the polysaccharide. As a consequence, although both types
717 of curdlan increased firmness, viscosity and reduced the syneresis of yogurts, the
718 curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-gelation method) caused a
719 greater increase in the parameters, but also showed a less stable gel formation than the
720 pre-gelled commercial curdlan. The commercial curdlan caused less significant effect in
721 the firmness and syneresis of yogurts tested. Nevertheless, the curdlans recovered by the
722 pre-gelation method provided rigid gels with a stable structure, allowing improvements
723 in the texture of a range of products, and therefore, has great potential for application in
724 the food industry.

725

726 **Acknowledgements**

727 The authors are thankful to the Brazilian Agencies CAPES, CNPq, FINEP and
728 Fundação Araucária for their financial support of this work.

729

730

731 **Author Contributions**

732 Conceived and designed the experiments: CSM GM. Performed the experiments:
733 CSM TTS VCF LNK SBSF. Analyzed the data: CSM MLB GM. Contributed
734 reagents/materials/analysis tools: MLB GM. Wrote the paper: CSM GM.

735

736 **References**

737 1. Zhang R, Edgar K. Properties, chemistry and applications of the bioactive
738 polysaccharide curdlan. *Biomacromolecules*. 2014; 15: 1079-1096.

739

740 2. Donot F, Fontana A, Baccou JC, Schorr-Galindo S. Microbial exopolysaccharides:
741 Main examples of synthesis, excretion, genetics and extraction. *Carbohydr Polym*.
742 2012; 87: 951-962.

743

744 3. Martinez CO, Ruiz SP, Fenelon VC, Morais GR, Baesso ML, Matioli G.
745 Characterization of curdlan produced by *Agrobacterium* sp. IFO 13140 cells
746 immobilized in loofa sponge matrix, and application of this biopolymer in the
747 development of functional yogurt. *J Sci Food Agric*. 2016; 96: 2410-2417.

748

749 4. Triveni R, Shamala TR, Rastogi NK. Optimised production and utilization of
750 exopolysaccharide from *Agrobacterium radiobacter*. *Process Biochem*. 2001; 36: 787-
751 795.

752

- 753 5. Lee I, Seo WT, Kim GJ, Kim MK, Park CS, Park YH. Production of curdlan using
754 sucrose or sugar cane molasses by two-step fed-batch cultivation of *Agrobacterium*
755 species. *J Ind Microbiol Biotechnol.* 1997; 18: 255-259.
756
- 757 6. Wu C, Yuan C, Chen S, Liu D, Ye X, Hu Y. The effect of curdlan on the rheological
758 properties of restructured ribbonfish (*Trichiurus* spp.) meat gel. *Food Chem.* 2015; 179:
759 222-231.
760
- 761 7. Li J, Zhu L, Lu G, Zhan X, Lin C, Zheng Z. Curdlan β -1,3-glucooligosaccharides
762 induce the defense responses against *Phytophthora infestans* infection of potato
763 (*Solanum tuberosum* L. cv. McCain G1) leaf cells. *PLoS ONE.* 2014; 9: e97197.
764
- 765 8. Wang M, Chen C, Sun G, Wang W, Fang H. Effects of curdlan on the color,
766 syneresis, cooking qualities and textural properties of potato starch noodles. *Starke.*
767 2010; 62: 429-434.
768
- 769 9. Chen C, Wang R, Sun G, Fang H, Ma D, Yi S. Effects of high pressure level and
770 holding time on properties of duck muscle gels containing 1% curdlan. *Innov Food Sci*
771 *Emerg Technol.* 2010; 11: 538-542.
772
- 773 10. Funami T, Nishinari K. Gelling characteristics of curdlan aqueous dispersions in the
774 presence of salts. *Food Hydrocoll.* 2007; 21: 59-65.
775

- 776 11. Chel-Guerrero L, Pérez-Flores V, Betancur-Ancona D, Dávila-Ortiz G. Functional
777 properties of flours and protein isolates from *Phaseolus lunatus* and *Canavalia*
778 *ensiformis* seeds. J Agric Food Chem. 2002; 50: 584-591.
779
- 780 12. Martinez CO, Ruiz SP, Nogueira MT, Bona E, Portilho M, Matioli G. Effective
781 immobilization of *Agrobacterium* sp. IFO 13140 cells in loofa sponge for curdlan
782 biosynthesis. Molecules. 2015; 20: 7957-7973.
783
- 784 13. Shin H, Liu L, Kim M, Park Y, Chen R. Metabolic engineering of *Agrobacterium*
785 sp. ATCC 31749 for curdlan production. J Ind Microbiol Biotechnol. 2016; 43: 1323-
786 1331.
787
- 788 14. Gagnon M, Lafleur M. From curdlan powder to the triple helix gel structure: an
789 attenuated total reflection-infrared study of the gelation process. Appl Spectrosc. 2007;
790 61: 374-378.
791
- 792 15. Fenelon VC, Aguiar MFA, Miyoshi JH, Martinez CO, Matioli G. Ultrafiltration
793 system for cyclodextrin production in repetitive batches by CGTase from *Bacillus*
794 *firmus* strain 37. Bioprocess Biosyst Eng. 2015; 38: 1291-1301.
795
- 796 16. Dubois M, Gilles KA, Hamilton JK, Reber PA, Smith F. Colorimetric method for
797 determination of sugars and related substances. Anal Chem. 1956; 28: 350-356.
798
- 799 17. AOAC International. Official Methods of Analysis of AOAC International (method
800 926.12). 1996.

- 801 18. AOAC International. Official Methods of Analysis of AOAC International (method
802 985.35). 1995.
803
- 804 19. Hirashima M, Takaya T, Nishinari K. DSC and rheological studies on aqueous
805 dispersions of curdlan. *Thermochim Acta*. 1997; 306: 109-114.
806
- 807 20. Keogh MK, O’Kennedy BT. Rheology of stirred yogurt as affected by added milk
808 fat, protein and hydrocolloids. *J Food Sci*. 1998; 63: 108-112.
809
- 810 21. Jin Y, Zhang H, Yin Y, Nishinari K. Comparison of curdlan and its
811 carboxymethylated derivative by means of Rheology, DSC, and AFM. *Carbohydr Res*.
812 2006; 341: 90-99.
813
- 814 22. Mikkelsen MS, Jespersen BM, Moller BL, Laerke HN, Larsen FH, Engelsen SB.
815 Comparative spectroscopic and rheological studies on crude and purified soluble barley
816 and oat β -glucan preparations. *Food Res Int*. 2010; 43: 2417-2424.
817
- 818 23. Novák M, Synytsya A, Gedeon O, Slepíčka P, Procházka V, Synytsya A, et al.
819 Yeast $\beta(1-3),(1-6)$ -D-glucan films: preparation and characterization of some structural
820 and physical properties. *Carbohydr Polym*. 2012; 87: 2496-2504.
821
- 822 24. Marchessault RH, Deslandes Y. Finestructure of (1 \rightarrow 3)- β -D-glucans: curdlan and
823 paramylon. *Carbohydr Res*. 1979; 75: 231-242.
824

- 825 25. Che Man YB, Irwandi J, Abdullah WJW. Effect of different types of maltodextrin
826 and drying methods on physico-chemical and sensory properties of encapsulated durian
827 flavour. *J Sci Food Agric*. 1999; 79: 1075-1080.
828
- 829 26. Favaro-Trindade CS, Santana AS, Monterrey-Quintero ES, Trindade MA, Netto
830 FM. The use of spray drying technology to reduce bitter taste of casein hydrolysate.
831 *Food Hydrocoll*. 2010; 24: 336-340.
832
- 833 27. Spicer EJJ, Goldenthal EI, Ikeda T. A toxicological assessment of curdlan. *Food*
834 *Chem Toxicol*. 1999; 37: 455-479.
835
- 836 28. Funami T, Funami M, Yada H, Nakao Y. A rheological study on the effects of
837 heating rate and dispersing method on the gelling characteristics of curdlan aqueous
838 dispersions. *Food Hydrocoll*. 2000; 14: 509-518.
839
- 840 29. Zhang H, Nishinari K, Williams MAK, Foster TJ, Norton IT. A molecular
841 description of the gelation mechanism of curdlan. *Int J Biol Macromol*. 2002; 30: 7-16.
842
- 843 30. Wu J, Zhan X, Liu H, Zheng Z. Enhanced production of curdlan by *Alcaligenes*
844 *faecalis* by selective feeding with ammonia water during the cell growth phase of
845 fermentation. *Chin J Biotechnol*. 2008; 24: 1035-1039.
846
- 847 31. Saitô H, Yokoi M, Yoshioka Y. Effect of hydration on conformational change or
848 stabilization of (1→3)- β -D-glucans of various chain lengths in the solid state as studied

849 by high-resolution solid-state ^{13}C NMR spectroscopy. *Macromolecules*. 1989; 22: 3892-
850 3898.

851

852 32. Nakanishi I, Kimura K, Kusui S, Yamazaki E. Complex formation of gel-forming
853 bacterial (1 \rightarrow 3)- β -D-glucans (curdlan-type polysaccharides) with dyes in aqueous
854 solution. *Carbohydr Res*. 1974; 32: 47-52.

855

856 33. Seguchi M, Kusunose C. Lipophilization of curdlan granules by heat-treatment or
857 chlorination. *Food Hydrocoll*. 2001; 15: 177-183.

858

859 34. Sciarini LS, Maldonado F, Ribotta PD, Pérez GT, León AE. Chemical composition
860 and functional properties of *Gleditsia triacanthos* gum. *Food Hydrocoll*. 2009; 23: 306-
861 313.

862

863 35. Chan TWD, Tang KY. Analysis of a bioactive β -(1 \rightarrow 3) polyssacharide (curdlan)
864 using matrix-assisted laser desorption/ionization time-of-flight mass spectrometry.
865 *Rapid Commun Mass Spectrom*. 2003; 17: 887-896.

866

867

Supporting information captions

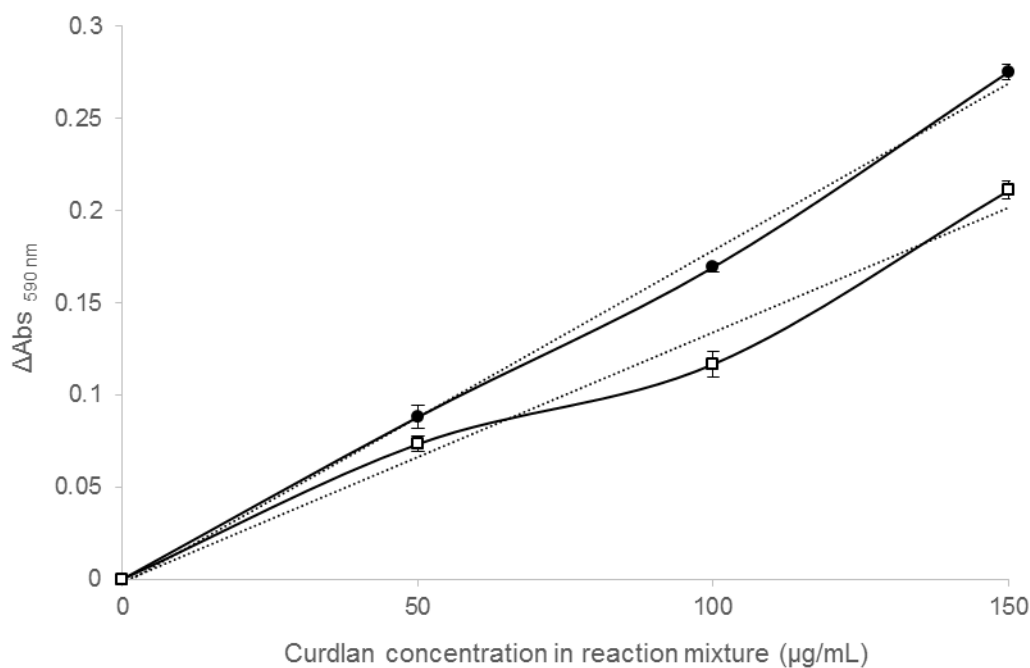
868

869

870 S1 Appendix. Curdlan interaction with aniline blue dye.

871 Solutions of the following concentrations of pre-gelled commercial curdlan and
872 of curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-gellation method) were
873 prepared in NaOH 5 mmol L⁻¹: 0, 50, 100 and 150 µg mL⁻¹. These solutions were mixed
874 with an equal volume of phosphate buffer (0.5 mol L⁻¹, pH 7) containing 160 mg L⁻¹ of
875 aniline blue and kept for 120 min at room temperature. Then, the absorbance of the
876 mixtures was measured at 590 nm. The analysis was performed in triplicate.

877



878

879 **Fig S1. Relationship between absorbance difference and concentration of**
880 **curdlans.** Pre-gelled commercial curdlan (●) and curdlan produced by *Agrobacterium*
881 sp. IFO 13140 recovered by the pre-gellation method (□).

882

ARTICLE 2

1
2
3 **Use of FT-IR, FT-Raman and thermal analysis to evaluate the gel formation of**
4 **curdlan produced by *Agrobacterium* sp. IFO 13140 and determination of its**
5 **rheological properties with food applicability**
6

7 Camila Sampaio Mangolim^a, Thamara Thaianne da Silva^b, Vanderson Carvalho Fenelon^c,
8 Adriane do Nascimento^d, Francielle Sato^d and Graciette Matioli^{a,c,*}

9
10 *^aPostgraduate Program in Food Science, State University of Maringá (UEM), Av.*

11 *Colombo, 5790 - 87020-900, Maringá, PR, Brazil*

12 *^bDepartment of Food Engineering, State University of Maringá (UEM), Av. Colombo,*

13 *5790 - 87020-900, Maringá, PR, Brazil*

14 *^cPostgraduate Program in Pharmaceutical Science, State University of Maringá*

15 *(UEM), Av. Colombo, 5790 - 87020-900, Maringá, PR, Brazil*

16 *^dPostgraduate Program in Physics, State University of Maringá (UEM), Av. Colombo,*

17 *5790 - 87020-900, Maringá, PR, Brazil*

18
19 * Corresponding author. Tel.: +55 44 3011-3868; fax: +55 44 3011-4119

20 E-mail address: gmatioli@uem.br (G. Matioli)

21
22 **ABSTRACT**

23 Curdlan is a linear polysaccharide composed of glucose units joined by β -(1,3)
24 bonds that possesses unique gelation properties. This study aimed to characterize the
25 structure and evaluate the gelling properties of curdlan produced by *Agrobacterium* sp.

26 IFO 13140 and its gels, as well as apply it in food. FT-Raman analysis highlighted the
27 structural changes that occurred during the formation of gels, with variations related to
28 the hydrogen bonds and hydrophobic interactions, which occur with the formation of the
29 low-set and high-set gels, respectively. Rheological analysis showed that the pre-gelled
30 commercial curdlan and the curdlan produced by *Agrobacterium sp.* IFO 13140 differed
31 in terms of gelation properties, which depends of the degree of polymerization of the
32 polysaccharide, but when applied to pasta products, both improved the texture
33 parameters. The curdlan gels were found to have great potential as gelling agents to
34 improve texture, water retention capacity and stability of food products.

35

36 **Keywords:** microbial polysaccharide, gelling agent, β -glucan, texture improver,
37 thickener.

38

39 **1 INTRODUCTION**

40

41 Curdlan is a neutral linear polysaccharide composed of repeating glucose units
42 linked by β -(1,3) bonds, and is classified as a β -glucan. It was discovered in 1966 when
43 Harada and collaborators investigated the production of succinoglycan using a non-
44 pathogenic bacillus found in soil. The microorganism, *Alcaligenes faecalis* variety
45 *myxogenes* (today named *Agrobacterium sp.* by the ATCC), produced curdlan in an
46 unexpected form by a mutant strain known as 10C3K (Amemura, Hisamatsu & Harada,
47 1977). This mutant strain became the precursor of others that preserved curdlan gel-
48 forming ability. Amongst these strains are ATCC 21680 (or IFO 13140), ATCC 31749
49 and ATCC 31750 (McIntosh, Stone & Stanisich, 2005).

50 The molecule was named curdlan due to its capacity to curdle when heated. This
51 property has allowed it to be used as a gelling agent for improving texture, water
52 retention capacity and thermal stability of several food products (Shih, Yu, Hsieh & Wu,
53 2009). Curdlan is one of the few bacterial additives approved by the US Food and Drug
54 Administration (FDA). Its gelling capacity has major potential for applications in food
55 and pharmaceutical industries; curdlan gels can be used as fat replacers and they are
56 colorless, flavorless and indigestible (Gagnon & Lafleur, 2007). Curdlan is therefore a
57 useful additive for the preparation of pasta products, frozen foods and canned meats,
58 being biodegradable, edible and non-toxic to humans and the environment (Shih et al.,
59 2009).

60 A number of studies of the application of curdlan have recently been carried out
61 in a range of industrial sectors. In the food industry, its use has been reported as a
62 texture enhancer (Wang, Chen, Sun, Wang & Fang, 2010), fat replacer (Chen, Wang,
63 Sun, Fan, Ma & Yi, 2010), encapsulant of active ingredients (Ferreira, Faria, Grosso &
64 Mercadante, 2009), for the lowering oil uptake during deep-fat frying (Funami, Funami,
65 Tawada & Nakao, 1999), and even as a potential prebiotic ingredient in rat feeding
66 (Shimizu, Tsuchihashi, Kudoh, Wada, Takita & Innami, 2001). In the biomedical area,
67 recent works have employed curdlan and its derivatives to prepare nanoparticles for
68 application as release-controlled drug vehicles (Na, Park, Kim & Bae, 2000; Popescu,
69 Pelin, Butnaru, Fundueanu & Suflet, 2013). Curdlan has also been recently used in the
70 preparation of edible blend films (Wu et al., 2012) and as a support matrix for enzyme
71 immobilization (Saudagar & Singhal, 2004).

72 In the food industry, curdlan is of great interest because of its solubility and gel
73 formation properties. It is insoluble in water, but its aqueous dispersion forms two types
74 of gels through heating, which are known as low-set or high-set. Low-set gel is obtained

75 by heating the dispersion at temperatures between 55 and 60 °C, with subsequent
76 cooling to below 40 °C, and is thermo-reversible. High-set gel is obtained by heating the
77 aqueous dispersion to above 80 °C and is irreversible (Funami & Nishinari, 2007).

78 Many factors such as the form of recovery of the polysaccharide from the
79 medium, concentration, temperature, heating time and rate, as well as the dispersion
80 method, can influence the conformational structure of curdlan and consequently affect
81 its mechanical and gelling properties. The transition temperature of the gel from thermo-
82 reversible to irreversible is dependent on its concentration and can also depend on the
83 molecular weight of the molecule (Funami & Nishinari, 2007). Even curdlans produced
84 by the same strain can differ in terms of mechanical properties depending on the
85 conditions of culture of the microorganism. Thus, the structural study of curdlan and its
86 gelling mechanisms and properties are essential to better define its use in the food
87 industry.

88 The structure and gelling mechanisms of curdlan have been studied by many
89 authors using a variety of methods. Techniques such as carbon nuclear magnetic
90 resonance (^{13}C -NMR) (Saitô, Ohki & Sasaki, 1977), infrared spectroscopy (FT-IR)
91 (Gagnon & Lafleur, 2007), X-ray diffraction (Harada, Okuyama, Konno, Koreeda &
92 Harada, 1994), electronic microscopy (Marchessault & Deslandes, 1979), and
93 differential scanning calorimetry (Hirashima, Takaya & Nishinari, 1997), have been
94 used to identify both the chemical bonds involved in the gelling process and the
95 structural conformation of the molecule. However, the use of Raman spectroscopy (FT-
96 Raman) for the study of the molecule and its gels is both rare and original, and in
97 accordance with Synytsya, Čopíková, Matějka and Machovič (2003), FT-IR and FT-
98 Raman techniques are complementary for the structural analysis of carbohydrates.

99 Considering the interest in the gelling properties of curdlan for food application,
100 the present study aimed to characterize the structure of curdlan produced by
101 *Agrobacterium* sp. IFO 13140 and evaluate its gels properties. The mechanisms of
102 formation of the curdlan gels were studied by FT-IR and FT-Raman spectroscopies, as
103 well as their formation temperature by thermal analysis. In addition, the application of
104 curdlan in homemade pasta was evaluated.

105

106 **2 MATERIALS AND METHODS**

107

108 **2.1 Materials**

109 The bacterial strain *Agrobacterium* sp. IFO 13140 was acquired from the
110 Institute for Fermentation of Osaka (Japan) in lyophilized form. Commercial curdlan
111 was obtained from Wako Pure Chemical Industries, Ltd. (Osaka, Japan). According to
112 the supplier, commercial curdlan is produced as a water-insoluble linear polysaccharide
113 without branching by *Alcaligenes faecalis* var. *myxogenes*. The other reagents were of
114 analytical grade.

115

116 **2.2 Production of curdlan by *Agrobacterium* sp. IFO 13140**

117 The reactivation of the microorganism was conducted using the culture medium
118 proposed by the supplier (g/L): polypeptone (10), yeast extract (2), MgSO₄·7H₂O (1)
119 and pH 7. Thirty milligrams of the lyophilized bacteria were incubated in 100 mL of the
120 medium at 30 °C and 120 rpm for 48 h. The free cells were recovered through
121 centrifugation at 6000 ×g for 10 min, washed with 0.9% NaCl (w/v) and transferred to
122 the production medium. The production of curdlan was performed in Erlenmeyer flasks
123 containing 100 mL of the liquid medium described by Martinez, Ruiz, Nogueira, Bona,

124 Portilho and Matioli (2015) (g/L): glucose (50), KH_2PO_4 (2.7), NH_4Cl (1.6), MgSO_4
125 (0.5), pH 7 and trace elements (10 mL/L). The composition of the trace elements (g/L)
126 in HCl 0.1 mol/L was: $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ (1), $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$ (1), CaCl_2 (1), ZnCl (1) and
127 CaCO_3 (0.03). The reactivated microorganisms were transferred to the production
128 medium and maintained at 30 °C and 150 rpm for 5 days. Curdlan production was
129 carried out in 5 batches, each containing 10 Erlenmeyer flasks with production medium.

130 To recover curdlan from the medium, NaOH 3 mol/L was added to the
131 Erlenmeyer flasks containing the production medium at a ratio of 1.8:1 (NaOH:medium)
132 to solubilize curdlan. Subsequently, the mixture was centrifuged at 18000 ×g for 15 min
133 at 4 °C to separate the cell pellets. The curdlan present in the supernatant was
134 precipitated by the addition of HCl 3 mol/L to pH 6-7 and recovered by centrifugation at
135 18000 ×g for 15 min at 4 °C. It was then washed three times with distilled water (20 °C)
136 and subsequently lyophilized. The curdlan yield (the relation between the amount of
137 curdlan produced and the initial amount of glucose) was also calculated. The
138 commercial curdlan was subjected to the same treatment, and was therefore described as
139 pre-gelled commercial curdlan to be differentiated from commercial curdlan. According
140 to Gagnon and Lafleur (2007), commercial curdlan is purified by dissolution in a strong
141 alkaline solution and dried in a spray-dryer, then washed with water until neutralization.

142

143 **2.3 Preparation of curdlan dispersions and gels**

144 The dispersions of the commercial pre-gelled curdlan and the curdlan produced
145 by *Agrobacterium* sp. IFO 13140 were prepared at a concentration of 2% (w/v) in water,
146 in accordance with Hirashima et al. (1997), with modifications. The water-insoluble
147 polysaccharide curdlan was dispersed in water using a mixer at room temperature for 5
148 min, sonicated for 10 min and then stirred in a magnetic stirrer for 12 h at 40 °C. The

149 dispersions were kept in a water bath at 61 °C/1 h to prepare the low-set gel and at 95
150 °C/1 h for preparation of the high-set gel. The temperatures used to gel formation were
151 according to Gagnon and Lafleur (2007), with slight modifications due to the
152 concentration of curdlan dispersions. After heating, samples were kept under
153 refrigeration or lyophilized for further analysis.

154

155 **2.4 Structural characterization of curdlan and its gels by FT-IR and FT-Raman**

156 Samples of the commercial pre-gelled curdlan and the curdlan produced by
157 *Agrobacterium* sp. IFO 13140, their lyophilized dispersions and gels, and also a sample
158 of the commercial curdlan, were analyzed by Fourier Transform Infrared Spectroscopy
159 (FT-IR) and Fourier transform Raman scattering infrared spectroscopy (FT-Raman).

160 The FT-IR spectra of the samples were obtained using a Fourier Transform
161 infrared spectrometer (Vertex 70v model, Bruker, Germany). The spectral region was
162 400-4000 cm^{-1} with 128 scans and a resolution of 2 cm^{-1} . The samples were mixed in
163 KBr powder and pellets were made to perform the measurements. The Raman spectra of
164 the samples were obtained using a Fourier Transform infrared spectrometer (Vertex 70v
165 model with Ram module II, Bruker, Germany) equipped with a Germanium detector
166 cooled with liquid nitrogen. A Nd:YAG laser was used for excitation at 1064 nm. The
167 spectra were based an average of 200 scans with a resolution of 4 cm^{-1} . Both FT-IR and
168 FT-Raman analysis were performed in duplicate.

169

170 **2.5 Thermal analysis of curdlan dispersions**

171 Differential Scanning Calorimetry (DSC) and Thermogravimetry (TG) were
172 performed on the dispersions (2%) of the commercial pre-gelled curdlan and the curdlan
173 produced by *Agrobacterium* sp. IFO 13140, prepared as described in section 2.3. The

174 analyses were conducted in duplicate using a Simultaneous System of DSC/TGA
175 Thermal Analysis (STA 409 PG Luxx model, Netzsch-Gerätebau GmbH, Selb/Bavaria,
176 Germany). Samples were heated from 20 to 160 °C at a rate of 2° C/min under a
177 nitrogen atmosphere (flow rate 30 mL/min).

178

179 **2.6 Determination of number-average degree of polymerization (DP_n) of curdlans**

180 The determination of the degree of polymerization of commercial curdlan, pre-
181 gelled commercial curdlan and curdlan produced by *Agrobacterium* sp. IFO 13140 was
182 performed according to the methodology established by Zhang and Lynd (2005) to
183 determine the DP_n of insoluble cellulose. By this method, the DP_n is calculated as the
184 ratio of glucosyl monomer concentration determined by the phenol-sulfuric acid method
185 divided by the reducing-end concentration determined by the modified 2,2'-
186 bicinchoninate (BCA) method. The assay was performed in quadruplicate.

187

188 **2.7 Rheological characteristics of curdlan dispersions and gels**

189 The dispersions of the pre-gelled commercial curdlan and the curdlan produced
190 by *Agrobacterium* sp. IFO 13140 at a concentration of 2% were prepared as described in
191 section 2.3. The samples were prepared 24 h prior to analysis and were analyzed in a
192 controlled stress rotational rheometer (HAAKE MARS II model, Thermo Fisher
193 Scientific Inc., Newington, Germany), with steel cone/plate geometry (60 mm,
194 separated by a fixed distance of 0.052 mm in the center). The samples were applied to
195 the lower plate of the rheometer very carefully and allowed to equilibrate for 60 s prior
196 to analysis. First, the linear viscoelastic region (the region in which stress is directly
197 proportional to strain and the G' modulus remain constant) for each sample was
198 determined at 20 °C. After this determination, the elastic (G') and viscous (G'') modulus,

199 and the dynamic viscosity were measured depending on temperature (20-60 °C) at a
200 frequency of 10 Hz. Subsequently, the same parameters, both for the dispersions and the
201 gels of curdlans were evaluated at a temperature of 20 °C, with variations in frequency
202 (0.1 to 10 Hz), and with application of a constant stress of 10 mPa. The curves of the
203 temperature sweep test and of the dynamic oscillatory frequency test were then
204 calculated using the RheoWin 4.10.000 software program (HAAKE software, Thermo
205 Fisher Scientific Inc., Newington, Germany). In each case, the dynamic rheological
206 properties of at least four replicate samples were determined (Hirashima et al., 1997).

207

208 **2.8 Curdlan application in homemade pasta products**

209

210 *2.8.1 Preparation of homemade pasta products*

211 To prepare the pasta products, wheat flour, curdlan (1%), water (30%), salt
212 (1.4%) and eggs (40%) were used. The amounts of each ingredient are described in
213 relation to wheat flour. With the exception of the curdlan, the ingredients were
214 purchased in local market. Four pasta product formulations were prepared: A) without
215 curdlan; B) with commercial curdlan; C) with pre-gelled commercial curdlan; D) with
216 curdlan produced by *Agrobacterium* sp. IFO 13140. First, the dry ingredients (except
217 curdlan) were mixed together. The curdlan was hydrated in water with a mixer and
218 added to the dry ingredients, together with the eggs. The pasta products were kneaded
219 and molded with a cylinder (3 mm thick) and cut and stored under refrigeration for
220 further analysis.

221

222

223

224 2.8.2 *Cooking characteristics of pasta products*

225 The cooking of the pasta product formulations was conducted according to
 226 Wang et al. (2010), with variation only in cooking time due to differences in thickness
 227 and type of cereal product. The parameters evaluated were cooking loss (CL) and
 228 cooked weight (CW). Approximately 3 g of pasta were weighed and boiled in 200 mL
 229 of boiling deionized water for 30 min. The container remained closed during cooking to
 230 prevent loss by evaporation. The cooked pasta products were rinsed with 50 mL of
 231 distilled water to eliminate particles released during cooking that remained adhered in
 232 some way, and dried with a paper towel to remove excess water from its surface. The
 233 CL was measured by draining the cooking water and drying to constant weight at 105
 234 °C, and expressed as the percentage of solids lost in cooking (Eq. 1). The CW was
 235 calculated and expressed as a percentage of the mass gained during cooking (Eq. 2).

$$236 \quad CL (\%) = \frac{W_{after\ drying}}{W_{initial}} \times 100 \quad (1)$$

$$237 \quad CW (\%) = \frac{W_{cooked}}{W_{initial}} \times 100 \quad (2)$$

238 Where, $w_{after\ drying}$ is the weight of solids obtained after draining the cooking water and
 239 drying to constant weight, $w_{initial}$ is the weight of the pasta before cooking and w_{cooked} is
 240 the weight of the pasta after cooking. The analysis was performed in triplicate.

241

242 2.8.3 *Texture profile analysis (TPA) and tensile strength of the pasta products*

243 For texture profile analysis of the pasta products, a TA-XT Plus texturometer
 244 (Stable Micro Systems, Godalming, UK) equipped with a version of the Texture Expert
 245 software (Stable Micro Systems, Godalming, UK) was used. The pasta products were
 246 prepared as described in section 2.8.1 and manually cut into 4 × 4 cm pieces (length ×
 247 width). After, the pasta were cooked in boiling water for 30 min, cooled in cold water

248 (20 °C, 10 min) and subsequently dried with paper towels. A 35 mm cylindrical probe
249 was used for the TPA test (ref. P/35, Stable Micro Systems). The probe compresses a
250 single strip of pasta dough with a constant deformation rate of 1 mm/s to 70% of initial
251 thickness. The probe was withdrawn and remained stationary for 10 s before a second
252 compression was performed to 70% of the original thickness, thus obtaining the TPA
253 curve, from which the hardness, cohesiveness, adhesiveness, springiness, gumminess
254 and chewiness parameters were calculated. The measurements were performed in
255 quadruplicate for each formulation. For the tensile strength test, the texturometer was
256 equipped with traction claws (ref. A/TG, Stable Micro Systems) and the test conducted
257 at a speed of 0.5 mm/s. The pasta products were prepared as described in section 2.8.1
258 and manually cut into 10 × 5 cm pieces (length × width). After, the pasta were cooked
259 during 30 min (3 h prior to analysis, to remove moisture from the surface) and
260 subsequently dried with paper towels. The assay was performed in triplicate for each
261 formulation.

262

263 **2.9 Statistical analysis**

264 The data obtained were analyzed by analysis of variance (ANOVA). Means were
265 compared with the Tukey Test ($p < 0.05$) using the Statistica 8.0/2008 software package
266 (Stat Soft, Inc., Tulsa, USA).

267

268 **3 RESULTS AND DISCUSSION**

269 Curdlan production by *Agrobacterium* sp. IFO 13140 achieved 20.0 ± 0.8 g/L,
270 which corresponds to a curdlan yield (the relation between curdlan production and the
271 initial amount of glucose) of $44 \pm 2\%$. When discovering the production of curdlan by
272 *Alcaligenes faecalis* var. *myxogenes* 10C3K, Harada, Fujimori, Hirose and Masada

273 (1966) established a culture medium that resulted in a high polysaccharide yield,
274 achieving 19.8 g/L in a medium containing 40 g/L of glucose, a result very similar to
275 that obtained in the present study.

276 In their review of exopolysaccharides of industrial interest, Donot, Fontana,
277 Baccou and Schorr-Galindo (2012) found that strains of *Agrobacterium* sp. can produce
278 5.02 to 76 g/L of curdlan depending on the strain and culture conditions. Authors who
279 worked with smaller containers and without continuous control of process parameters,
280 such as Portilho, Matioli, Zanin, Moraes and Scamparini (2006), obtained a production
281 of approximately 20 g/L of curdlan by *Agrobacterium* sp. IFO 13140 and 22 g/L by
282 *Agrobacterium* sp. ATCC 31749, values that are very close to those found in the present
283 study, which shows that *Agrobacterium* sp. IFO 13140 is a high yield strain for the
284 production of curdlan. Authors who worked with higher volumes in a fermenter with
285 continuous control of process parameters obtained the highest values of curdlan
286 production. Kim, Ryu, Choi, Rhee and Lee (2003), using a mutant strain of
287 *Agrobacterium* sp. ATCC 31750 in a 300 L fermenter achieved a production of 76 g/L.

288

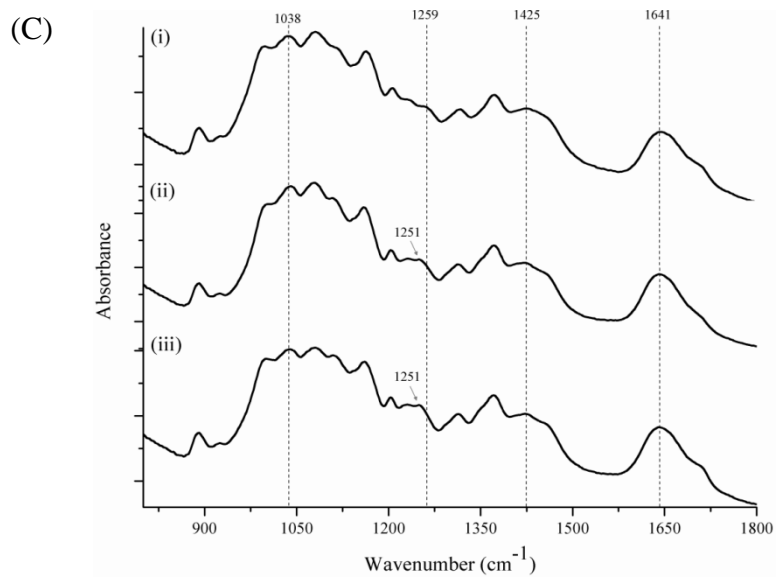
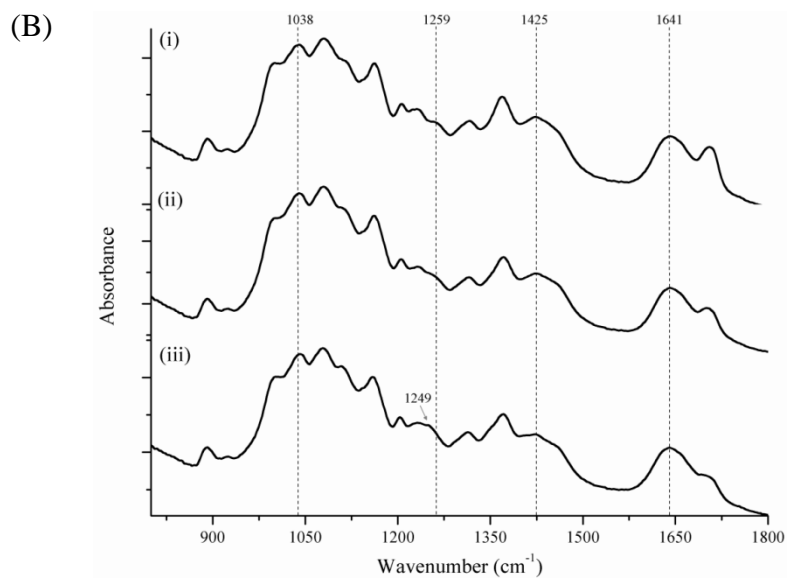
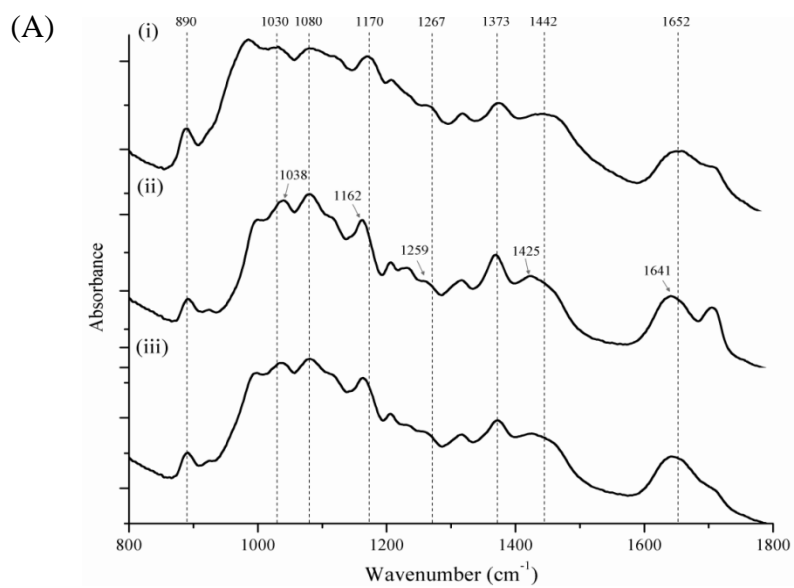
289 **3.1 Structural characterization of curdlan and its gels by FT-IR and FT-Raman**

290 Figure 1 shows the FT-IR spectra of the commercial curdlan, the pre-gelled
291 commercial curdlan and the curdlan produced by *Agrobacterium* sp. IFO 13140 (Fig.
292 1A), of the dispersion of the pre-gelled commercial curdlan and its gels (Fig. 1B) and of
293 the dispersion of curdlan produced by *Agrobacterium* sp. IFO 13140 and its gels (Fig.
294 1C).

295

296

297



299 **Figure 1:** FT-IR spectra: (A) curdlans – (i) commercial, (ii) pre-gelled commercial, (iii)
300 produced by *Agrobacterium* sp. IFO 13140; (B) dispersion of pre-gelled commercial
301 curdlan (i) without heat treatment, (ii) heated to 61 °C/1 h, (iii) heated to 95 °C/1 h; (C)
302 dispersion of curdlan produced by *Agrobacterium* sp. IFO 13140 (i) without heat
303 treatment, (ii) heated to 61 °C/1 h, (iii) heated to 95 °C/1 h. The dotted lines show the
304 characteristic peaks of the curdlans tested.

305

306 In the commercial curdlan spectrum (Fig. 1Ai) the most significant peaks for the
307 characterization of the polysaccharide are demonstrated and their tentative assignments
308 are described: the peak at 890 cm^{-1} , is attributed to the $\text{C}_1\text{-H}$ deformation of the
309 formation of the β -glycosidic bond, and the peak at approximately 1170 cm^{-1} , is
310 probably attributed to the vibration $\text{C}_1\text{-O-C}_3$ of the same glycosidic bond (Gagnon &
311 Lafleur, 2007). The peak at 1030 cm^{-1} is associated with the vibration of the $\text{C}_1\text{-O-C}_5$
312 bond of the ring of the glucose molecule (Wu, Cai & Sun, 2012). The peaks at 1080,
313 1267 and 1373 cm^{-1} are related to the vibrations of C–O, C–OH and C–H, respectively
314 (Jin, Zhang, Yin & Nishinari, 2006). The last peaks, at 1442 and 1652 cm^{-1} are
315 associated to the vibrations of the CH_2 groupings (carbon 6) and the association between
316 the water and the curdlan molecule, resulting in vibrations of the H–O–H grouping,
317 respectively (Popescu et al., 2013).

318 When pre-gelled, the commercial curdlan displayed the same variations in the
319 FT-IR spectrum as the curdlan produced by *Agrobacterium* sp. IFO 13140 (Figures 1Aii
320 and 1Aiii), revealing a structural similarity between the commercial curdlan and that
321 produced by the microorganism. The differences between the spectra of commercial
322 curdlan and the curdlans submitted to pre-gelling (pre-gelled commercial curdlan and
323 the one produced by *Agrobacterium* sp. IFO 13140), which are some displacements of

324 the peaks at 1030, 1170, 1267 and 1652 cm^{-1} of the original molecule, can be related to
325 the initial solubilization of curdlan in NaOH (that leads to water incorporation in the
326 molecule) and subsequent neutralization with HCl (that causes the restoration of
327 hydrogen bonds) due to the pre-gelation process. The bands at 1030, 1170 and 1267
328 cm^{-1} are associated to vibrations of C–O or C–OH of curdlan molecules, while the band
329 around 1640 cm^{-1} is associated with water.

330 Figures 1B and 1C show the FT-IR spectra of curdlan dispersions and its gels.
331 Modifications were envisaged, especially at the peaks at 1030, 1267 and 1442 cm^{-1} for
332 the gels formed, which are related to the vibrations of the C₁–O–C₅, C–OH and –CH₂
333 bonds, involved in the gelling processes. However, modifications had already appeared
334 in these peaks in both dispersions (Figs. 1Bi and 1Ci), and these were maintained in the
335 gels formed. As such, it is difficult to identify structural modifications in curdlan related
336 to its gelling using the FT-IR technique employed in this study. This is firstly due to the
337 fact that the obtaining or pre-gelling of curdlan involves the reorganization of the
338 hydrogen bonds of the molecule, a phenomenon that also occurs with the formation of
339 the low-set gel, and secondly due to the spectra obtained presenting significant
340 overlaying of the peaks, making the study of structure difficult.

341 Gagnon and Lafleur (2007) studied the processes of hydration and gelling of
342 curdlan molecule using the FT-IR–ATR technique and obtained similar results to those
343 found in the present study. When analyzing the region of 850 to 1200 cm^{-1} , they
344 identified the appearance of characteristic peaks of curdlan molecule, especially at 890
345 and 1170 cm^{-1} and also perceived that the hydration of the molecule significantly
346 modified the spectra of this region, as it is where the vibrations of the C–OH bonds of
347 the carbohydrate appear. Also, the authors observed that the low-set gel presented an
348 identical spectrum to the hydrated molecule, considering that the mechanism of

349 formation of the gel also occurs through the structuration of hydrogen bonds, and that
350 the high-set gel possesses subtle differences, making it difficult to identify the structural
351 modifications responsible for gelling.

352 According to Synytsya et al. (2003), Raman spectroscopy is widely used for
353 structural investigations of polysaccharides, being evident that the FT-IR and FT-Raman
354 spectroscopies are complementary for the structural analysis of carbohydrates. Some
355 vibrational bands, which are either very weak or overlaid by other stronger bands and,
356 therefore, cannot be detected using IR, can be identified or studied in FT-Raman
357 spectroscopy and vice-versa. In the case of the FT-IR spectra obtained in the present
358 study for curdlan and its gels, the region between 800 and 1600 cm^{-1} is replete with
359 overlaid bands, while the FT-Raman spectra are clearer and the peaks better defined in
360 this region. Figure 2 shows the FT-Raman spectra of commercial curdlan, pre-gelled
361 commercial curdlan and curdlan produced by *Agrobacterium* sp. IFO 13140 (Fig. 2A),
362 of the dispersion of the pre-gelled commercial curdlan and its gels (Fig. 2B) and of the
363 dispersion of curdlan produced by *Agrobacterium* sp. IFO 13140 and its gels (Fig. 2C).

364

365

366

367

368

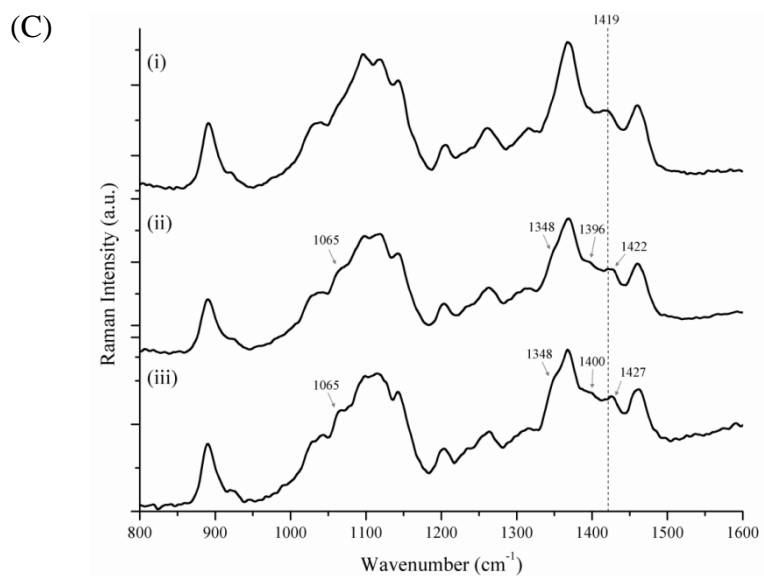
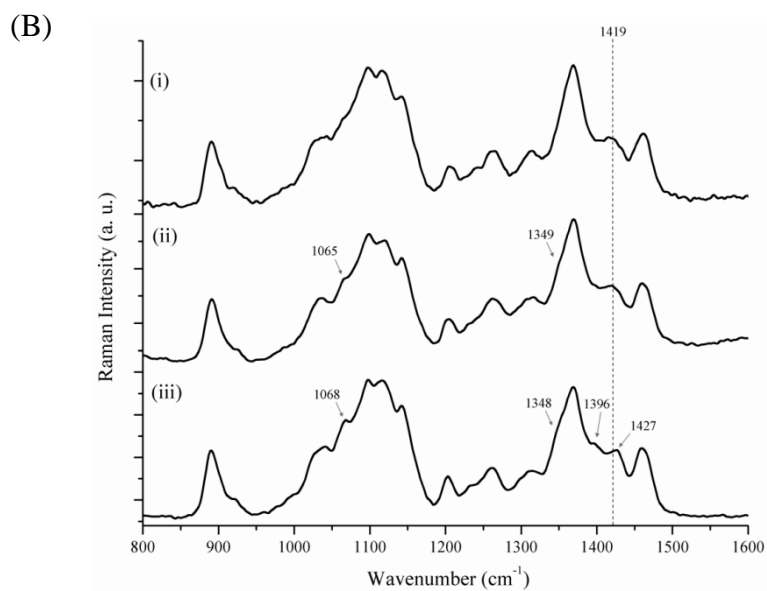
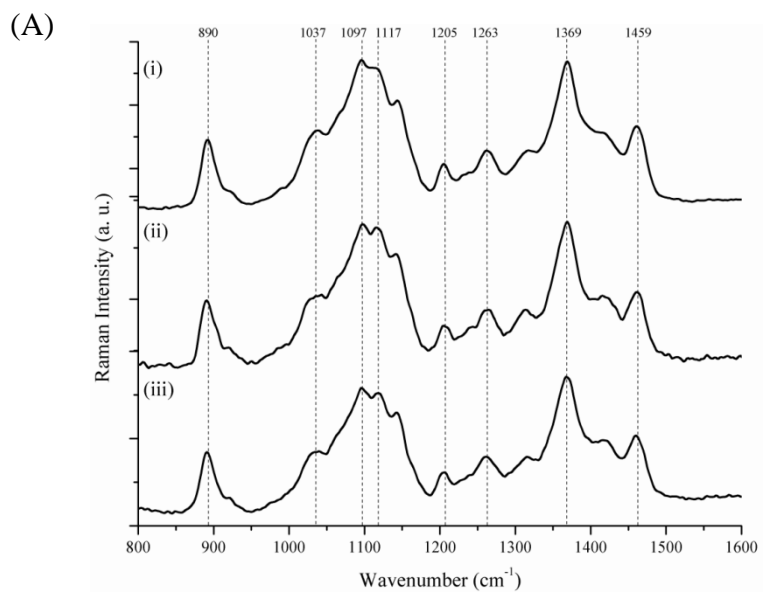
369

370

371

372

373



375 **Figure 2:** FT-Raman spectra: (A) curdlans – (i) commercial, (ii) pre-gelled commercial,
376 (iii) produced by *Agrobacterium* sp. IFO 13140; (B) dispersion of pre-gelled
377 commercial curdlan (i) without heat treatment, (ii) heated to 61 °C/1 h, (iii) heated to 95
378 °C/1 h; (C) dispersion of curdlan produced by *Agrobacterium* sp. IFO 13140 (i) without
379 heat treatment, (ii) heated to 61 °C/1 h, (iii) heated to 95 °C/1 h. The dotted lines show
380 the characteristic peaks of the curdlans tested.

381

382 Figure 2A shows the most significant FT-Raman peaks for the characterization
383 of polysaccharide. Again, the peak at 890 cm^{-1} is attributed to the $\text{C}_1\text{-H}$ deformation of
384 the formation of the glucoside bond in the β form. The peak at 1037 cm^{-1} is attributed to
385 the stretching of the C–C and C–OH bonds of the molecule (Synytsya et al., 2003),
386 while the peaks at 1097 and 1117 cm^{-1} are related to the symmetric and asymmetric
387 stretching of C–O–C of the glycosidic bonds. According to Kizil and Irudayaraj (2007),
388 the peak at 1205 cm^{-1} is attributed to the vibrations of the C–C–H bonds and the peak at
389 1263 cm^{-1} is the result of the vibrations of the CH_2OH group of the side chain of the
390 glucoses. Finally, the peak at 1369 cm^{-1} is attributed to the angular deformation of the
391 CH and $\text{C}_3\text{-OH}$ bonds, and the peak at 1459 cm^{-1} is found in the angular deformation
392 region of the CH_2 grouping.

393 Remaining with Figure 2A, it can be seen that the three samples have the same
394 structure. Then, Figures 2B and 2C can provide evidences of the structural variations
395 due to the formation of gels in the commercial pre-gelled curdlan and the curdlan
396 produced by *Agrobacterium* sp. IFO 13140. With the formation of low-set gel (Figs.
397 2Bii and 2Cii), there is the appearance of a shoulder at 1065 cm^{-1} , attributed to the
398 stretching of the C–O and C–C bonds of the glucose ring, a shoulder at 1349 cm^{-1} ,
399 related to the angular deformation of the C–OH bond and the twisting of the CH_2

400 radical, and another shoulder at 1396 cm^{-1} (more evident in the curdlan gel produced by
401 the microorganism), which can be attributed to the angular bending of the CH_2 radical
402 and the deformation of the C–H and C–OH bonds (Kizil & Irudayaraj, 2007). These
403 variations in these spectra are related to the formation mechanism of the low-set gel.
404 According to Gagnon and Lafleur (2007), the low-set gel can be considered an assembly
405 of swollen particles that interact via hydrogen bonding. The main conformation in these
406 gels is the single helix structure. Additionally, Tako and Hanashiro (1997) propose a
407 more specific mechanism that retains this single helix structure: the hydrogen bonding
408 between the oxygen of the glucose ring ($\text{C}_1\text{--O--C}_5$) and the hydroxyl of carbon 4, or
409 between hydroxyls of the CH_2 groupings of carbon 6.

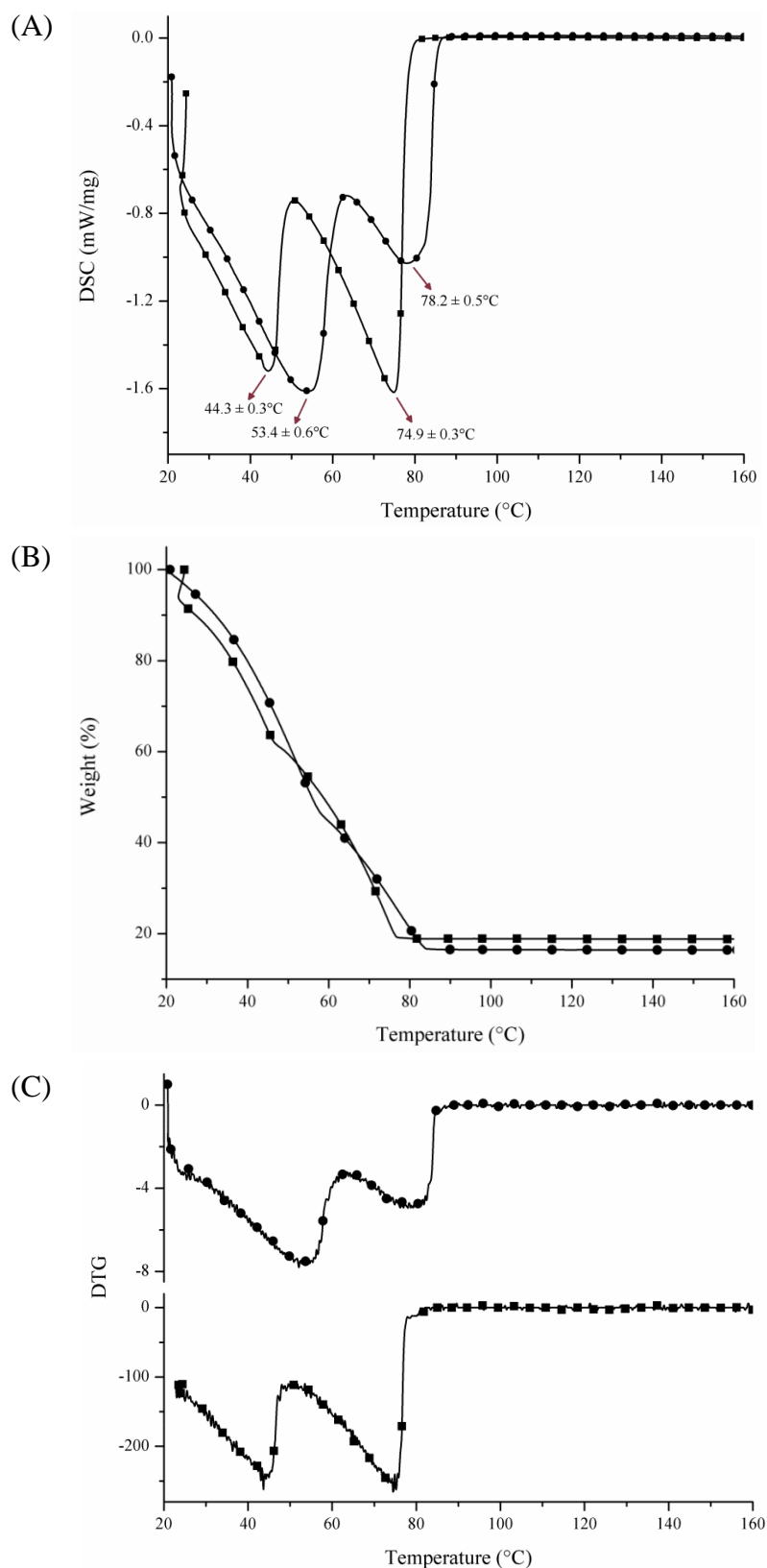
410 The formation of the high-set gel is evidenced by the accentuation of the
411 formation of the shoulders described for the low-set gel, added to the displacement of
412 the peak at 1419 cm^{-1} to 1427 cm^{-1} . This peak is related to the angular deformation of
413 the CH_2 grouping of the glycoses of the molecule. As described by many authors
414 (Gagnon & Lafleur, 2007; Saitô et al., 1977; Tako & Hanashiro, 1997) the high-set gel
415 is formed by intermolecular hydrophobic interactions which are associated with the
416 formation of a triple helix conformation. Tako and Hanashiro (1997) propose a more
417 specific mechanism that preserves this triple helix conformation: the intermolecular
418 hydrophobic interactions between CH_2 groupings of C-6. Additionally, Saitô et al.
419 (1977) observed by ^{13}C NMR some shift of C-4 signals, which is compatible with an
420 intramolecular hydrogen bond between O-4'---O-5 and are ascribed to a region of single
421 helical conformation. The same authors also observed very broad ^{13}C resonance peaks
422 of C-1–C-5 and C-6 that are ascribed to the triple-helical junction zones for the gel
423 structure and their vicinities. Also, Takeda, Yasuoka, Kasai and Harada (1978) stated by
424 X-ray diffraction studies that by heat-treatment, the single helical molecules of curdlan

425 in water change their structure to triple-stranded helices, showing a behavior that
426 suggest the existence of a compact hydrophobic structure after heat-treatment. All these
427 results are in accordance with the mechanisms described above for formation of low-set
428 and high-set gels of curdlan and also give support to the ones proposed by Tako and
429 Hanashiro (1997). The gel formation mechanisms proposed by Tako and Hanashiro
430 (1997) which are in agreement with the FT-Raman variations of this work are shown in
431 Supplementary material 1.

432

433 **3.2 Thermal analysis (DSC and TG) of curdlan dispersions**

434 Figure 3 shows the DSC (Fig. 3A), TG (Fig. 3B) and DTG (Fig. 3C) data of the
435 dispersions of the pre-gelled commercial curdlan and curdlan produced by
436 *Agrobacterium* sp. IFO 13140, in a concentration of 2%. In Fig. 3A, both samples
437 displayed two well defined peaks, with the former being between 40 and 55 °C and the
438 latter between 70 and 80 °C. The two samples showed great variation in peak
439 temperature, which made it necessary to analyze the degree of polymerization (DP_n), as
440 according to Zhang, Huang, Nishinari, Watase and Konno, 2000, the magnitude of the
441 DSC peaks and the temperature of the peaks are susceptible to changes depending on
442 the concentration, molecular weight and experimental conditions. The DP_n values
443 obtained were 334 ± 8 for the pre-gelled commercial curdlan and 232 ± 10 for the
444 curdlan produced by microorganism, which corresponds to molecular masses of
445 approximately 54000 and 38000, respectively, and which explains why the peaks of the
446 two dispersions were not at the same temperature (the higher molecular weights lead to
447 higher transition temperatures). A dispersion of commercial curdlan was not evaluated
448 in thermal analysis because the commercial curdlan ($DP_n = 332 \pm 12$) did not form
449 homogeneous dispersions and gels with the methodology employed in section 2.3.



450

451 **Figure 3:** DSC (A), TG (B) and DTG (C) data of the dispersions of pre-gelled
 452 commercial curdlan (●) and curdlan produced by *Agrobacterium* sp. IFO 13140 (■) in a
 453 concentration of 2%.

454 In Fig. 3A, the first peak is due to the swelling of curdlan caused by the breaking
455 of some hydrogen bonds, the first step of low-set gel formation. A number of studies
456 (Harada et al., 1994; Hirashima et al. 1997; Jin et al., 2006; Konno & Harada, 1991;
457 Zhang, Nishinari, Williams, Foster & Norton, 2002; Zhang et al., 2000) have performed
458 DSC of curdlan gels in aqueous dispersions of 1, 2, or 4%, with a heating rate of 1 or 2
459 °C/min. All these works found that between 40 and 60 °C a clear endothermic peak
460 appears due to swelling of curdlan molecules. Hirashima et al. (1997), for example,
461 observed an endothermic peak at close to 58 °C, which shifted slightly with the
462 increasing concentration of the curdlan dispersion (5 to 15%). The second peak shown
463 in Fig. 3A for both dispersions (between 70 and 80 °C) is due to the hydrophobic
464 interactions between curdlan molecules. Konno and Harada (1991) reported the
465 appearance of a wide and shallow peak between 70 and 105 °C, which indicates the
466 structural transition resulting from the formation of a firm gel stabilized by hydrophobic
467 interactions. In the present study, the peak that represents the formation of this gel is
468 thin and clearly defined.

469 According to Gagnon and Lafleur (2007), when an aqueous dispersion of
470 curdlan is heated, two types of gel may be formed depending on the heating
471 temperature. The first is a thermo-irreversible (~80 °C) high-set gel, maintained by
472 intermolecular hydrophobic interactions, and the second is a thermo-reversible (~55 °C)
473 low-set gel, maintained by intramolecular hydrogen bonds. The curdlan chains can
474 adopt two helical conformations: single or triple helices. The gelling property of the
475 polysaccharide is also attributed to transformations of its helical arrangement at
476 different temperatures. At room temperature and in the low-set gel, the structure is
477 predominantly formed by single helix curdlan chains, which are more strongly

478 associated in the low-set gel. In the high-set gel, the structure is predominantly formed
479 by associated triple helices, forming an organized gel configuration.

480 The temperature for formation of the gels indicated in Fig. 3A can also be
481 identified in Fig. 3B, which represents the loss of mass data of the curdlan dispersion
482 samples during heating. For both samples, two distinct stages of loss of mass appeared
483 at temperatures close to the endothermic peaks of DSC. In the first step, at up to 46.6
484 and 58.4 °C for the curdlan produced by *Agrobacterium* sp. IFO 13140 and the pre-
485 gelled commercial curdlan, respectively, the weight loss is more pronounced, as the
486 curdlan molecules are disorganized and swollen, which facilitates water loss. In the
487 second stage, at temperatures up to 77 and 84.2 °C for the curdlan produced by
488 *Agrobacterium* sp. IFO 13140 and the pre-gelled commercial curdlan, respectively, the
489 loss of mass was a little less pronounced, as with the formation of a firm gel it is more
490 difficult for water to become detached from the structure formed.

491 The two temperature stages seen in Fig. 3B, in which the mass loss was
492 observed, can be better understood by Fig. 3C, which shows differential mass loss
493 thermograms (DTG) of the curdlans, obtained from the weight loss curves (Fig. 3B).
494 This mathematical procedure provides the rate of mass loss with increasing temperature.
495 As a result, it can be seen that the DSC curves are recovered in the DTG thermograms
496 of the samples, evidencing the approximate transition temperatures for the two gel
497 formations for both the curdlan produced by *Agrobacterium* sp. IFO 13140 (~44 and 74
498 °C), and for the pre-gelled commercial curdlan (~54 and 78 °C).

499

500 **3.3. Rheological characteristics of curdlan dispersions and gels**

501 Determining some technological properties of polysaccharides is of great
502 importance when predicting their possible applications. To better define the use of

503 curdlan and its gels as thickeners and gelling agents in the food industry, a rheological
504 study that elucidate their mechanical properties is essential. Figure 4 shows the
505 rheological data of the dispersion samples of the pre-gelled commercial curdlan and the
506 curdlan produced by *Agrobacterium* sp. IFO 13140 based on temperature (Figure 4A)
507 and the oscillatory rheology of the curdlan dispersions and gels (Figures 4B and 4C).
508 Once again, dispersions and gels of commercial curdlan were not evaluated in
509 rheological analysis because the commercial curdlan did not form homogeneous
510 dispersions and gels with the methodology employed in section 2.3.

511

512

513

514

515

516

517

518

519

520

521

522

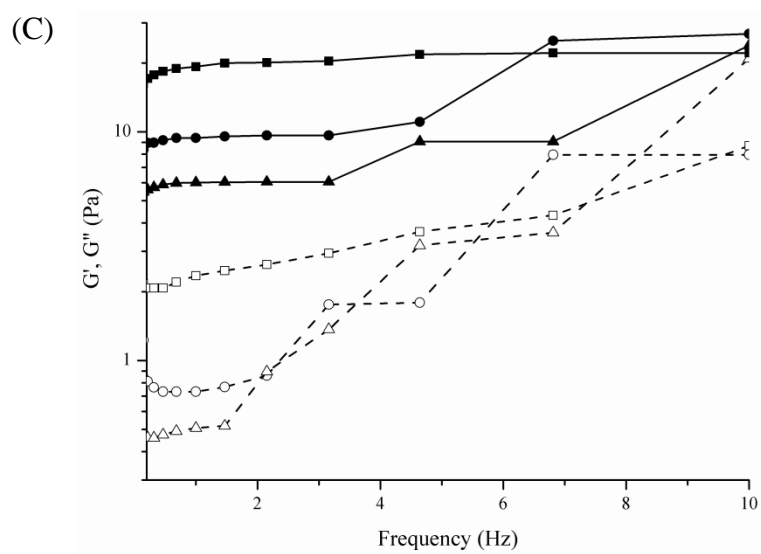
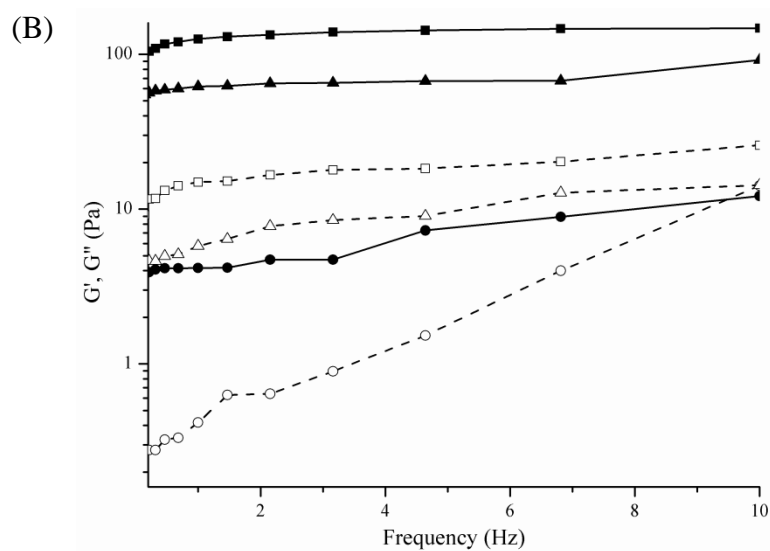
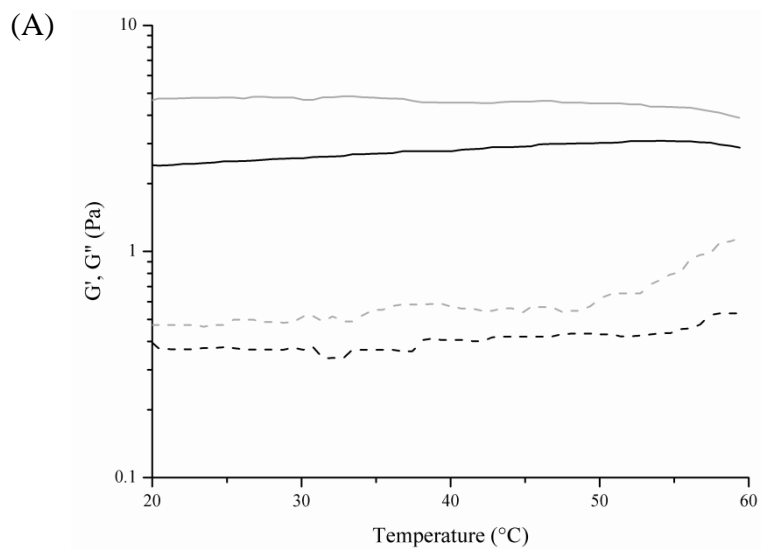
523

524

525

526

527



528 **Figure 4:** Temperature dependency of the G' (continuous line) and G'' (dotted line)
529 modules for the dispersions of pre-gelled commercial curdlan (black line) and of curdlan
530 produced by *Agrobacterium* sp. IFO 13140 (grey line) (A); the frequency dependency of
531 the modules G' (continuous line) and G'' (dotted line) for the dispersions of pre-gelled
532 commercial curdlan (B) and of curdlan produced by *Agrobacterium* sp. IFO 13140 (C)
533 without heat treatment (●), heated to 61 °C/1 h (▲) and heated to 95 °C/1 h (■).

534

535 In Figure 4A it can be seen that the dispersions of the pre-gelled commercial
536 curdlan and the curdlan produced by *Agrobacterium* sp. IFO 13140 display elastic
537 behavior in their gels, as the elastic module (G') is greater than the viscous module (G'').
538 Also, the curdlan produced by *Agrobacterium* sp. IFO 13140 presented higher values
539 for both modules, as well as higher values for dynamic viscosity (data not shown),
540 indicating that it has greater thickening potential than the pre-gelled commercial
541 curdlan, which can be due to its lower DP_n , that facilitates the swelling or hydration of
542 the molecule. The increase in temperature up to 60 °C did not have a widely significant
543 effect on the rheological parameters of the dispersions (Fig. 4A). This is because, in the
544 pre-gelling process, the neutralization of the alkaline solution of curdlan with acid
545 results in a gelatinous precipitate, which has similar properties to those of the low-set
546 gel (Kanzawa, Harada, Koreeda & Harada, 1987).

547 The G' values slightly decreased for the curdlan produced by the microorganism
548 with increasing temperature. A similar result was found by Salah et al. (2011),
549 evaluating the rheological properties of commercial curdlan depending on temperature
550 (up to 65 °C), who observed that increasing the temperature reduced the viscosity of the
551 curdlan. According to the authors, this is due to the fact that the temperature affects the
552 conformational structure of the polysaccharide in solution so that it undergoes a

553 transition from a rigid and organized structure to a disordered structure. For the pre-
554 gelled commercial curdlan the G' values increased slightly. This temperature
555 dependence shows that the pre-gelled commercial curdlan mobility is lower than the
556 mobility of curdlan produced by the microorganism, and its chains are therefore more
557 rigid. As such, the commercial curdlan possesses a more rigid structure than that
558 produced by *Agrobacterium* sp. IFO 13140, demonstrating greater structural stability
559 when subjected to temperature variation. Similar results were obtained by Zhang et al.
560 (2002), who when evaluating the rheological behavior of 2% curdlan dispersions
561 depending on temperature found that this module experienced a slight increase as
562 temperature increased.

563 The rheological parameters of the curdlan dispersions and their gels were
564 evaluated according to frequency (Figs. 4B and 4C). Once again, the dispersion of
565 curdlan (without being submitted to heat treatment) produced by *Agrobacterium* sp. IFO
566 13140 presented higher values for both G' and G'' modules, as well as higher values for
567 dynamic viscosity (data not shown), indicating that it has greater thickening potential
568 than the pre-gelled commercial curdlan. Despite this fact, the commercial curdlan had a
569 much greater gelling capacity, as the G' and G'' values for its high-set gel reached 100
570 and 10 Pa, respectively, while for the high-set gel of the curdlan produced by the
571 microorganism these values remained only slightly superior to the dispersion without
572 heat treatment. The great difference between gelling capacities of the curdlans is related
573 to their distinct DP_n (or molecular weight). The extent of forming the triple-stranded
574 helices may simply depend upon the molecular weight of the polysaccharide. Thus,
575 higher degree of cross-linking may be achieved for samples with larger DP_n . In other
576 words, the gel forming ability of curdlan increases with increasing DP_n of the
577 polysaccharide (Nakanishi, Kimura, Kusui & Yamazaki, 1974). Also, according to

578 Ogawa and Tsurugi (1973), DP_n around 200 may be the lower limit to form gels in
579 aqueous media (DP_n of curdlan produced by the microorganism is 232 ± 10).

580 Still in Figs. 4B and 4C, it was verified that the parameters G' and G'' increased
581 with the formation of gels, especially the parameter G' , which is the measure of the
582 energy stored within the gel network. This shows that the interlacing between the
583 curdlan molecules, which forms a three-dimensional gel network, becomes stronger
584 with increased temperature. Additionally, the values of the G''/G' relationship based on
585 frequency were greater for samples that were not submitted to heat treatment, and
586 declined as the temperature used for gel formation increased. For example, for the
587 dispersion sample of the pre-gelled commercial curdlan at 2%, the G''/G' values were
588 0.15 ± 0.01 to 1.2 ± 0.1 between the frequencies of 0.1 to 10 Hz for the sample without
589 heat treatment and 0.045 ± 0.009 to 0.18 ± 0.03 for the sample treated at $95\text{ }^\circ\text{C}/1\text{ h}$.
590 Analogous results were found for the dispersion samples of the curdlan produced by
591 microorganism at 2%. As the relationship G''/G' is a measure of the dynamic
592 characteristics of the intermolecular bonds in the gel network, low values indicate that
593 the gel reacts to stress in a relatively more elastic than viscous manner. This result
594 suggests that the number of elastically active chains of curdlan molecules increased with
595 gel formation, making the gel an elastic structure. Similar results were found by
596 Hirashima et al. (1997), who assessed the rheological parameters of 2% curdlan
597 dispersions, and observed an increase, notably in G' , as temperature increased, showing
598 the hardness of the gel formed, and also observed the reduction of the G''/G'
599 relationship.

600 The curdlan dispersions and their gels displayed an increase in rheological
601 parameters as the frequency applied was increased, but became more linear when higher
602 temperatures were employed for forming the gels (Figs. 4B and 4C), corroborating the

603 fact that the hardness obtained by the high-set gels makes this a stable structure for use
604 in various products.

605

606 **3.4 Application of curdlan in homemade pasta products**

607 Curdlan is an ingredient of great interest to both the food and food supplement
608 industries because, in addition to its ability to gel when heated, it is the most
609 concentrated source of β -(1,3)-insoluble glucan available, and so is considered a dietary
610 fiber. The addition of curdlan to foods therefore has many advantages. It can contribute
611 to a variety of textures in foods, depending on the heat treatment applied, the
612 concentration and the other ingredients present in the product. Among its main
613 properties of interaction with other ingredients, it has a high water absorption capacity,
614 and so can help obtain products with higher humidity and consequently a higher yield. It
615 also has a high capacity of fat absorption, making it an important structural agent for
616 food, especially in increasing palatability and the extension of shelf life (Jezequel,
617 1998). These properties influence various attributes of homemade pasta products.

618 According to Wang et al. (2010), the attributes of cooking loss and cooked
619 weight are of great importance for cooked pasta products and consist of indicators of the
620 loss of solids and the water absorbed during cooking, respectively. The addition of
621 curdlan did not influence the cooking loss (CL) as the four pasta products evaluated
622 (without curdlan, with commercial curdlan, with pre-gelled commercial curdlan and
623 with curdlan produced by *Agrobacterium* sp. IFO 13140) presented values between 4.6
624 $\pm 0.9\%$ and $5.9 \pm 0.5\%$ for this parameter, with no statistically significant difference
625 ($p < 0.05$). However, the use of curdlan increased statistically significantly ($p < 0.05$) the
626 water absorption or the cooked weight (CW) of the pasta products, with the pre-gelled
627 commercial curdlan and the curdlan produced by the microorganism presenting similar

628 CW values of $175 \pm 4\%$ and $173 \pm 3\%$, respectively. The commercial curdlan had no
629 effect on the cooked weight parameter, most likely because it is not easily homogenized
630 in the other ingredients. The CW value for the pasta products with commercial curdlan
631 was $158 \pm 5\%$, a value statistically equal ($p < 0.05$) to the pasta products without curdlan
632 ($156 \pm 5\%$).

633 There is a great difficulty in homogenizing or dispersing curdlan in food
634 products because it is insoluble in water. Therefore, it is necessary to use an efficient
635 homogenizer and, according to Zhang et al. (2002) and Jin et al. (2006), apply a further
636 stirring of the dispersion prior to the carrying out of applications requiring uniformity.
637 Both pre-gelled commercial curdlan and curdlan produced by *Agrobacterium* sp. IFO
638 13140 disperse easily in water when subjected to stirring and, when heated at $95\text{ }^{\circ}\text{C}$,
639 form a firm and homogeneous gel. However, the commercial curdlan do not easily
640 disperse in water and, even being gelled after heating at $95\text{ }^{\circ}\text{C}$, it occurs with a phase
641 separation in the gel formed. Other authors (Marchessault & Deslandes, 1979) have
642 previously identified the formation of a non-homogenous gel from curdlan in its native
643 form. Thus, the characteristics of dispersion of curdlan depend greatly on the recovery
644 method employed after its production. In the pre-gelling process, the neutralization of
645 the alkaline solution of curdlan with acid results in a gelatinous precipitate (Kanzawa et
646 al., 1987), which is much easier to rehydrate than commercial curdlan after dried
647 (commercial curdlan is neutralized with water and precipitates in the form of particles).
648 Thus, the use of the pre-gelling method in curdlans, which significantly increases the
649 ease application of the polysaccharide in food matrices and consequently results in
650 significant effects in the texture characteristics of pastas, is extremely important.

651 Wang et al. (2010) added different amounts of curdlan to macaroni pasta and
652 obtained similar results to the present study for the cooking loss parameter, since the

653 addition of curdlan in concentrations of up to 1% did not change the values found.
654 However, the same author found that an increase in the amount of curdlan added to the
655 pasta products resulted in a decrease in cooked weight, which varied from the result of
656 this work. It is likely that the lack of prior treatment of the curdlan is the reason for this
657 result, as pre-gelation allows a better dispersion of the polysaccharide in the pasta
658 products. Comparing the cooked weight data for the samples without curdlan and with
659 commercial curdlan in this study, no variation in the attribute was noted.

660 Commercial curdlan and the one produced by *Agrobacterium* sp. IFO 13140 are
661 odorless, tasteless and have a very light color (light grayish yellow color).
662 Supplementary material 2 exhibits a picture of both samples. Therefore, the addition of
663 curdlan caused no significant effects on the flavor and appearance of homemade pasta
664 products. Table 1 presents the texture parameters for these products prepared with and
665 without curdlan. The addition of pre-gelled commercial curdlan and curdlan produced
666 by *Agrobacterium* sp. IFO 13140 caused a significant increase in the hardness,
667 adhesiveness and gumminess of the pasta, and a less significant increase in chewiness.
668 This is because, with gelling occurring homogeneously in the pasta products, they
669 become firmer and more adhesive in the mouth during chewing, and maintains a more
670 persistent density, meaning that it spends more time in the mouth.

671

672

673

674

675

676

677 **Table 1:** Texture parameters of homemade pasta products with and without curdlan.

678 Values indicate mean \pm standard-deviation.

679

Texture parameter	Sample			
	A	B	C	D
<i>Hardness (N)</i>	135 ^b \pm 4	140 ^b \pm 6	183 ^a \pm 2	181 ^a \pm 9
<i>Cohesiveness</i>	0.86 ^a \pm 0.01	0.79 ^b \pm 0.01	0.73 ^b \pm 0.05	0.78 ^b \pm 0.01
<i>Adhesiveness (N \times mm)</i>	6 ^c \pm 4	37 ^b \pm 8	60 ^a \pm 8	58 ^a \pm 7
<i>Springiness (mm)</i>	0.98 ^a \pm 0.01	0.94 ^b \pm 0.02	0.93 ^b \pm 0.02	0.95 ^{ab} \pm 0.01
<i>Gumminess ($\times 10^2$ N)</i>	117 ^b \pm 3	110 ^b \pm 3	134 ^a \pm 9	142 ^a \pm 8
<i>Chewiness ($\times 10^2$ N)</i>	115 ^{bc} \pm 2	105 ^c \pm 4	125 ^{ab} \pm 10	134 ^a \pm 7
<i>Tensile strength (N)</i>	28 ^a \pm 1	28.4 ^a \pm 0.4	29.4 ^a \pm 0.5	28.3 ^a \pm 0.3
<i>Extensibility (mm)</i>	11 ^b \pm 1	14 ^b \pm 1	13 ^b \pm 2	20 ^a \pm 2

680 ^{a,b,c} Means within the same line with different letters are significantly different ($p < 0.05$).

681 Pasta product formulations: A) without curdlan; B) with commercial curdlan; C) with pre-gelled
682 commercial curdlan; D) with curdlan produced by *Agrobacterium* sp. IFO 13140.

683

684 The addition of commercial curdlan to the mass did not affect its hardness or
685 gumminess, but caused an increase (even if less significant than the other curdlans) in
686 adhesiveness. In terms of cohesiveness, the parameter related to degree of compression
687 of the food between the teeth before it breaks, all the curdlans caused the same reduction
688 in this parameter. The addition of curdlan had little influence on the parameters of
689 springiness, chewiness and tensile strength of the pastas. These changes are due to the
690 fact that the irreversible gel of curdlan is a firmer gel than it is elastic, and has greater
691 influence on parameters related to chewing force.

692 Wang et al. (2010) evaluated the firmness and tensile strength of the pasta
693 products with the addition of different concentrations of curdlan, and noted that adding
694 curdlan of over 0.3% increased the firmness of the pasta products and also the tensile
695 strength thereof. The authors also state that texture is a determining factor in pasta

696 quality: high firmness indicates high chewiness and high tensile strength indicates high
697 elasticity, which are desirable characteristics for cooked pasta products. Oishi et al.
698 (2009) added various concentrations of curdlan (up to 2%) to pasta products with
699 hypoallergenic wheat flour (in which gluten is partially hydrolyzed by enzymes) and
700 noticed that the addition of curdlan to this pasta increased the breaking strain and the
701 hardness of the pasta in addition to inhibiting cooking loss, as the concentration of the
702 curdlan added increased. The authors explain that this is due to the formation of an
703 irreversible gel which occurs at cooking temperatures.

704

705 **4 CONCLUSIONS**

706

707 The data obtained from structural analysis revealed a similarity between the
708 curdlan produced by *Agrobacterium* sp. IFO 13140 and the commercial curdlan. Among
709 the analyses performed, the FT-Raman technique proved to be especially useful for the
710 study of structural changes resulting from the formation of the curdlan gels, showing the
711 hydrogen bonds and hydrophobic interactions that occur with the formation of the low-
712 set and high-set gels, respectively, which could not be observed in FT-IR analysis.
713 Despite the structural similarities, the gelling properties of the curdlans displayed
714 significant differences, which are related to the difference between their degrees of
715 polymerization. The curdlan produced by *Agrobacterium* sp. IFO 13140, which has the
716 lowest DP_n , has a greater thickening capacity than the pre-gelled commercial curdlan,
717 while the latter has a considerably greater gelling capacity. In spite of that, both types of
718 curdlan increased the hardness and water absorption of pasta products, unlike
719 commercial curdlan. The characteristics of dispersion and consequently gelation and
720 rheological properties of curdlan depend greatly on the recovery methods employed

721 after its production, as the curdlan produced by *Agrobacterium* sp. IFO 13140 and the
722 commercial pre-gelled curdlan disperse better in water than the commercial curdlan in
723 its native form and are easily to disperse during their food application. In an innovative
724 manner, the use of pre-gelling method has been proven to be very important for a more
725 effective use of curdlan in foods. Nevertheless, the curdlan gels, especially the high-set
726 gel, had a rigid, stable and elastic structure, displaying great potential for use as a
727 gelling agent to improve the texture, water retention capacity and thermal stability of
728 several food products.

729

730 **5 ACKNOWLEDGEMENTS**

731

732 All authors are thankful to the Brazilian Agencies CNPq, CAPES, FINEP and
733 Fundação Araucária for their financial support.

734

735 **6 REFERENCES**

736

737 Amemura, A., Hisamatsu, M., & Harada, T. (1977). Spontaneous mutation of
738 polysaccharide production in *Alcaligenes faecalis* var. *myxogenes* 10C3. *Applied and*
739 *environmental Microbiology*, 34, 617–620.

740

741 Chen, C., Wang, R., Sun, G., Hongmei, F., Ma, D., & Yi, S. (2010). Effects of high
742 pressure level and holding time on properties of duck muscle gels containing 1%
743 curdlan. *Innovative Food Science and Emerging Technologies*, 11, 538–542.

744

- 745 Donot, F., Fontana, A., Baccou, J. C., & Schorr-Galindo, S. (2012). Microbial
746 exopolysaccharides: Main examples of synthesis, excretion, genetics and extraction.
747 *Carbohydrate Polymers*, 87, 951–962.
- 748
- 749 Ferreira, D. S., Faria, A. F., Grosso, C. R. F., & Mercadante, A. Z. (2009).
750 Encapsulation of blackberry anthocyanins by thermal gelation of curdlan. *Journal of the*
751 *Brazilian Chemical Society*, 20, 1908–1915.
- 752
- 753 Funami, T., & Nishinari, K. (2007). Gelling characteristics of curdlan aqueous
754 dispersions in the presence of salts. *Food Hydrocolloids*, 21, 59–65.
- 755
- 756 Funami, T., Funami, M., Tawada, T., & Nakao, Y. (1999). Decreasing oil uptake of
757 doughnuts during deep-fat frying using curdlan. *Journal of Food Science*, 64, 883–888.
- 758
- 759 Gagnon, M., & Lafleur, M. (2007). From curdlan powder to the triple helix gel
760 structure: an attenuated total reflection-infrared study of the gelation process. *Applied*
761 *Spectroscopy*, 61, 374–378.
- 762
- 763 Harada, T., Fujimori, K., Hirose, S., & Masada, M. (1966). Growth and β -glucan
764 10C3K production by a mutant of *Alcaligenes faecalis* var. *myxogenes* in defined
765 medium. *Agricultural and Biological Chemistry*, 30, 764–769.
- 766
- 767 Harada, T., Okuyama, K., Konno, A., Koreeda, A., & Harada, A. (1994). Effect of
768 heating on formation of curdlan gels. *Carbohydrate Polymers*, 24, 101–106.
- 769

- 770 Hirashima, M., Takaya, T., & Nishinari, K. (1997). DSC and rheological studies on
771 aqueous dispersions of curdlan. *Thermochimica Acta*, 306, 109–114.
772
- 773 Jezequel, V. (1998). Curdlan: a new functional β -glucan. *Cereal Foods World*, 43, 361–
774 364.
775
- 776 Jin, Y., Zhang, H., Yin, Y., & Nishinari, K. (2006). Comparison of curdlan and its
777 carboxymethylated derivative by means of Rheology, DSC, and AFM. *Carbohydrate*
778 *Research*, 341, 90–99.
779
- 780 Kanzawa, Y., Harada, T., Koreeda, T. A., & Harada, A. (1987). Curdlan gel formed by
781 neutralizing its alkaline solution. *Agricultural and Biological Chemistry*, 51, 1839–
782 1843.
783
- 784 Kim, M., Ryu, K., Choi, W., Rhee, Y., & Lee, I. (2003). Enhanced production of
785 (1→3)- β -D-glucan by a mutant strain of *Agrobacterium* species. *Biochemical*
786 *Engineering Journal*, 16, 163–168.
787
- 788 Kizil, R., & Irudayaraj, J. (2007). Rapid evaluation and discrimination of γ -irradiated
789 carbohydrates using FT-Raman spectroscopy and canonical discriminant analysis.
790 *Journal of the Science of Food and Agriculture*, 87, 1244–1251.
791
- 792 Konno, A., & Harada, T. (1991). Thermal properties of curdlan in aqueous suspension
793 and curdlan gel. *Food Hydrocolloids*, 5, 427–434.
794

- 795 Marchessault, R. H., & Deslandes, Y. (1979). Finestructure of (1→3)-β-D-glucans:
796 curdlan and paramylon. *Carbohydrate Research*, 75, 231–242.
797
- 798 Martinez, C. O., Ruiz, S. P., Nogueira, M. T., Bona, E., Portilho, M., & Matioli, G.
799 (2015). Effective immobilization of *Agrobacterium* sp. IFO 13140 cells in loofa sponge
800 for curdlan biosynthesis. *Molecules*, 20, 7957–7973.
801
- 802 McIntosh, M., Stone, B. A., & Stanisich, V. A. (2005). Curdlan and other bacterial
803 (1→3)-β-D-glucans. *Applied Microbiology and Biotechnology*, 68, 163–173.
804
- 805 Na, K., Park, K., Kim, S. W., & Bae, Y. H. (2000). Self-assembled hydrogel
806 nanoparticles from curdlan derivatives: Characterization, anti-cancer drug release, and
807 interaction with a hepatoma cell line (HepG2). *Journal of Controlled Release*, 69,
808 225–236.
809
- 810 Nakanishi, I., Kimura, K., Kusui, S., & Yamazaki, E. (1974). Complex formation of gel-
811 forming bacterial (1→3)-β-D-glucans (curdlan-type polysaccharides) with dyes in
812 aqueous solution. *Carbohydrate Research*, 32, 47–52.
813
- 814 Ogawa, K., & Tsurugi, J. (1973). The dependence of the conformation of a (1→3)-β-D-
815 glucan on chain-length in alkaline solution. *Carbohydrate Research*, 29, 397–403.
816
- 817 Oishi, Y., Udagawa, H., Shinozaki, Y., Moriyama, M., Taniguchi, H., Kobayashi-
818 Hattori, K., Arai, S., & Takita, T. (2009). Preparation of hypoallergenic wheat flour

- 819 noodles and evaluation of their physical properties. *Food Science and Technology*
820 *Research*, 15, 39–44.
- 821
- 822 Popescu, I., Pelin, I. M., Butnaru, M., Fundueanu, G., & Suflet, D. M. (2013).
823 Phosphorylated curdlan microgels. Preparation, characterization, and in vitro drug
824 release studies. *Carbohydrate Polymers*, 94, 889–898.
- 825
- 826 Portilho, M., Matioli, G., Zanin, G. M., Moraes, F. F., & Scamparini, A. R. P. (2006).
827 Production of insoluble exopolysaccharide of *Agrobacterium* sp. (ATTC 31749 and IFO
828 13140). *Applied Biochemistry and Biotechnology*, 129–132, 864–869.
- 829
- 830 Saitô, H., Ohki, T., & Sasaki, T. (1977). A ^{13}C nuclear magnetic resonance study of gel-
831 forming (1→3)- β -D-glucans. Evidence of the presence of single-helical conformation in
832 a resilient gel of a curdlan-type polysaccharide 13140 from *Alcaligenes faecalis* var.
833 *myxogenes* IFO 13140. *Biochemistry*, 16, 908–914.
- 834
- 835 Salah, R. B., Jaouadi, B., Bouaziz, A., Chaari, K., Blecker, C., Derrouane, C., Attia, H.,
836 & Besbes, S. (2011). Fermentation of date palm juice by curdlan gum production from
837 *Rhizobium radiobacter* ATCC 6466TM: purification, rheological and physico-chemical
838 characterization. *LWT - Food Science and Technology*, 44, 1026–1034.
- 839
- 840 Saudagar, P. S., & Singhal, R. S. (2004). Curdlan as a support matrix for immobilization
841 of enzyme. *Carbohydrate Polymers*, 56, 483–488.
- 842

- 843 Shih, I., Yu, J., Hsieh, C., & Wu, J. (2009). Production and characterization of curdlan
844 by *Agrobacterium* sp. *Biochemical Engineering Journal*, 43, 33–40.
845
- 846 Shimizu, J., Tsuchihashi, N., Kudoh, K., Wada, M., Takita, T., & Innami, S. (2001).
847 Dietary curdlan increases proliferation of bifidobacteria in the cecum of rats.
848 *Bioscience, Biotechnology, and Biochemistry*, 65, 466–469.
849
- 850 Synytsya, A., Čopíková, J., Matějka, P., & Machovič, V. (2003). Fourier transform
851 Raman and infrared spectroscopy of pectins. *Carbohydrate Polymers*, 54, 97–106.
852
- 853 Takeda, H., Yasuoka, N., Kasai, N., & Harada, T. (1978). X-ray structural studies of
854 (1→3)-β-D-glucan (curdlan). *Polymer Journal*, 10, 365–368.
855
- 856 Tako, M., & Hanashiro, I. (1997). Evidence for a conformational transition in curdlan.
857 *Polymer Gels and Networks*, 5, 241–250.
858
- 859 Wang, M., Chen, C., Sun, G., Wang, W., & Fang, H. (2010). Effects of curdlan on the
860 color, syneresis, cooking qualities and textural properties of potato starch noodles.
861 *Starch*, 62, 429–434.
862
- 863 Wu, C., Peng, S., Wen, C., Wang, X., Fan, L., Deng, R., & Pang, J. (2012). Structural
864 characterization and properties of konjac glucomannan/curdlan blend films.
865 *Carbohydrate Polymers*, 89, 497–503.
866

867 Wu, S., Cai, R., & Sun, Y. (2012). Degradation of curdlan using hydrogen peroxide.
868 *Food Chemistry*, 135, 2436–2438.

869

870 Zhang, H., Huang, L., Nishinari, K., Watase, M., & Konno. (2000). Thermal
871 measurements of curdlan in aqueous suspension during gelation. *Food Hydrocolloids*,
872 14, 121–124.

873

874 Zhang, H., Nishinari, K., Williams, M. A. K., Foster, T. J., & Norton, I. T. (2002). A
875 molecular description of the gelation mechanism of curdlan. *International Journal of*
876 *Biological Macromolecules*, 30, 7–16.

877

878 Zhang, Y. P., & Lynd, L. R. (2005). Determination of the number-average degree of
879 polymerization of cellodextrins and cellulose with application to enzymatic hydrolysis.
880 *Biomacromolecules*, 6, 1510–1515.

881

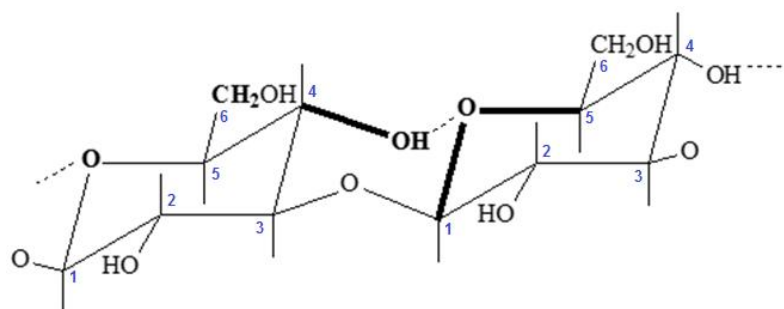
882

883

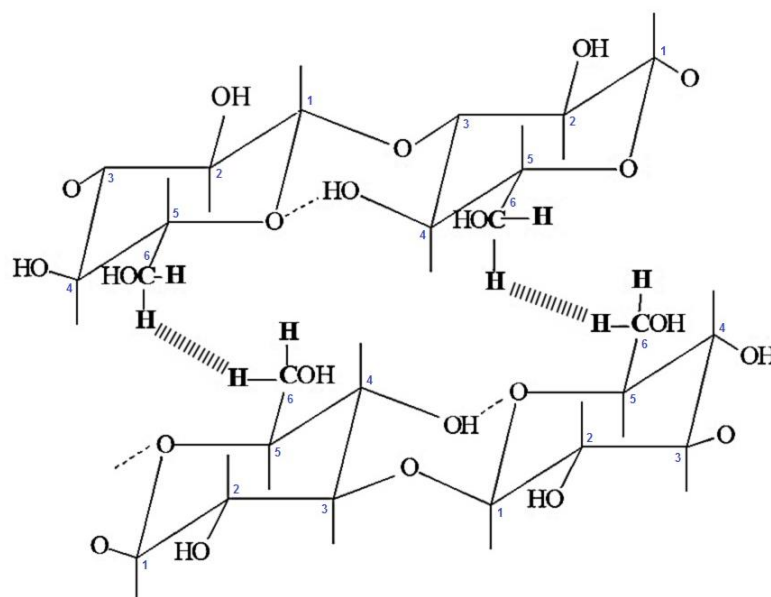
Supplementary material 1

884

(A)



(B)



885

886 **Figure S1:** Formation mechanisms of (A) low-set and (B) high-set gels described by

887 Tako and Hanashiro (1997) and suggested in the present study by FT-Raman.

888

889

890

Supplementary material 2

891

892



893

894 **Figure S2:** Pictures of (A) commercial curdlan and (B) of curdlan produced by

895 *Agrobacterium* sp. IFO 13140.

ANEXOS

ANEXO 1: Comprovante de aceite/publicação (Plos One)

01/03/2017 Fwd: Notification of Formal Acceptance for PONE-D-16-42057 - [EMID:4334ac65357a39ba] - camilamangolim@gmail.com - Gmail

Fwd: Notification of Formal Acceptance for PONE-D-16-42057 - [EMID:4334ac65357a39ba]

Entrada x



Graciette Matioli
para mim

21 de fev (Há 8 dias)



inglês

português

[Traduzir mensagem](#)

[Desativar para: inglês](#)

----- Forwarded message -----

From: PLOS ONE <em@editorialmanager.com>

Date: 2017-02-21 11:29 GMT-03:00

Subject: Notification of Formal Acceptance for PONE-D-16-42057 - [EMID:4334ac65357a39ba]

To: Graciette Matioli <gmatioli@uem.br>

CC: camilamangolim@gmail.com, thamarathaiane01@hotmail.com, vanderson2912@gmail.com, lnkoqa@uem.br, sbsferreira88@gmail.com

PONE-D-16-42057

Description of Recovery Method Used for Curdlan Produced by *Agrobacterium* sp. IFO 13140 and its Relation to the Morphology and Physicochemical and Technological Properties of the Polysaccharide

Dear Dr. Matioli:

I am pleased to inform you that your manuscript has been deemed suitable for publication in PLOS ONE. Congratulations! Your manuscript is now with our production department.

If your institution or institutions have a press office, please notify them about your upcoming paper at this point, to enable them to help maximize its impact. If they will be preparing press materials for this manuscript, please inform our press team within the next 48 hours. Your manuscript will remain under strict press embargo until 2 pm Eastern Time on the date of publication. For more information please contact onepress@plos.org.

For any other questions or concerns, please email plosone@plos.org.

Thank you for submitting your work to PLOS ONE.

With kind regards,

PLOS ONE Editorial Office Staff
on behalf of

Prof. Krishnendu Acharya
Academic Editor
PLOS ONE

RESEARCH ARTICLE

Description of recovery method used for curdlan produced by *Agrobacterium* sp. IFO 13140 and its relation to the morphology and physicochemical and technological properties of the polysaccharide

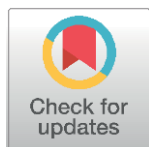
Camila Sampaio Mangolim¹, Thamara Thaianne da Silva², Vanderson Carvalho Fenelon³, Luciana Numata Koga³, Sabrina Barbosa de Souza Ferreira³, Marcos Luciano Bruschi³, Graciette Matioli^{1,3*}

1 Postgraduate Program in Food Science, State University of Maringá (UEM), Maringá, Paraná, Brazil,

2 Department of Food Engineering, State University of Maringá (UEM), Maringá, Paraná, Brazil,

3 Postgraduate Program in Pharmaceutical Science, State University of Maringá (UEM), Maringá, Paraná, Brazil

* gmatioli@uem.br



OPEN ACCESS

Citation: Mangolim CS, Silva TTd, Fenelon VC, Koga LN, Ferreira SBdS, Bruschi ML, et al. (2017) Description of recovery method used for curdlan produced by *Agrobacterium* sp. IFO 13140 and its relation to the morphology and physicochemical and technological properties of the polysaccharide. PLoS ONE 12(2): e0171469. doi:10.1371/journal.pone.0171469

Editor: Krishnendu Acharya, University of Calcutta, INDIA

Received: October 21, 2016

Accepted: January 21, 2017

Published: February 28, 2017

Copyright: © 2017 Mangolim et al. This is an open access article distributed under the terms of the [Creative Commons Attribution License](https://creativecommons.org/licenses/by/4.0/), which permits unrestricted use, distribution, and reproduction in any medium, provided the original author and source are credited.

Data Availability Statement: All relevant data is in the paper and supporting information files.

Funding: The authors are thankful to the Brazilian Agencies CAPES, CNPq and Fundação Araucária for their financial support of this work.

Competing interests: The authors have declared that no competing interests exist.

Abstract

Curdlan is a linear polysaccharide considered a dietary fiber and with gelation properties.

This study evaluated the structure, morphology and the physicochemical and technological properties of curdlan produced by *Agrobacterium* sp. IFO 13140 recovered by pre-gelation and precipitation methods. Commercial curdlan submitted or otherwise to the pre-gelation process was also evaluated. The data obtained from structural analysis revealed a similarity between the curdlan produced by *Agrobacterium* sp. IFO 13140 (recovered by both methods) and the commercial curdlans. The results showed that the curdlans evaluated differed significantly in terms of dispersibility and gelation, and only the pre-gelled ones had significant potential for food application, because this method influence on the size of the particles and in the presence of NaCl. In terms of technological properties, the curdlan produced by *Agrobacterium* sp. IFO 13140 (pre-gelation method) had a greater water and oil holding capacity (64% and 98% greater, respectively) and a greater thickening capacity than the pre-gelled commercial curdlan. The pre-gelled commercial curdlan displayed a greater gel-gelling capacity at 95°C than the others. When applied to food, only the pre-gelled curdlans improved the texture parameters of yogurts and reduced syneresis. The curdlan gels, which are rigid and stable in structure, demonstrated potential for improving the texture of food products, with potential industrial use.

ANEXO 2: Normas (Plos One)

Submission Guidelines

Style and Format

File format	<p>Manuscript files can be in the following formats: DOC, DOCX, RTF, or PDF. Microsoft Word documents should not be locked or protected.</p> <p>LaTeX manuscripts must be submitted as PDFs. Read the LaTeX guidelines.</p>
Length	<p>Manuscripts can be any length. There are no restrictions on word count, number of figures, or amount of supporting information.</p> <p>We encourage you to present and discuss your findings concisely.</p>
Font	Use a standard font size and any standard font, except for Symbol font.
Headings	Limit manuscript sections and sub-sections to 3 heading levels. Make sure heading levels are clearly indicated in the manuscript text.
Layout	<p>Manuscript text should be double-spaced.</p> <p>Do not format text in multiple columns.</p>
Page and line numbers	Include page numbers and line numbers in the manuscript file.
Footnotes	Footnotes are not permitted. If your manuscript contains footnotes, move the information into the main text or the reference list, depending on the content.
Language	<p>Manuscripts must be submitted in English.</p> <p>You may submit translations of the manuscript or abstract as supporting information. Read the supporting information guidelines.</p>
Abbreviations	<p>Define abbreviations upon first appearance in the text.</p> <p>Do not use non-standard abbreviations unless they appear at least three times in the text.</p> <p>Keep abbreviations to a minimum.</p>
Reference style	<p>PLOS uses “Vancouver” style, as outlined in the ICMJE sample references.</p> <p>See reference formatting examples and additional instructions below.</p>
Equations	We recommend using MathType for display and inline equations, as it will provide the most reliable outcome. If this is not possible, Equation Editor is acceptable.

Avoid using MathType or Equation Editor to insert single variables (e.g., “ $a^2 + b^2 = c^2$ ”), Greek or other symbols (e.g., β , Δ , or ' [prime]), or mathematical operators (e.g., x , \geq , or \pm) in running text. Wherever possible, insert single symbols as normal text with the correct Unicode (hex) values.

Do not use MathType or Equation Editor for only a portion of an equation. Rather, ensure that the entire equation is included. Avoid “hybrid” inline or display equations, in which part is text and part is MathType, or part is MathType and part is Equation Editor.

Nomenclature Use correct and established nomenclature wherever possible.

<i>Units of measurement</i>	Use SI units. If you do not use these exclusively, provide parentheses after each value. Read more about SI units.
<i>Drugs</i>	Provide the Recommended International Non-Proprietary Name (RINN). Write in italics (e.g., <i>Homo sapiens</i>). Write out in full title both in the title of the manuscript and at the first mention in a paper. After first mention, the first letter of the genus name and the full species name may be used (e.g., <i>H. sapiens</i>).
<i>Species names</i>	Write in italics. Use the recommended name by consulting the genetic nomenclature database (e.g., HUGO for human genes). It is sometimes advisable to indicate the synonyms for the gene as it appears in the text. Gene prefixes such as those used for cellular localization should be shown in roman typeface (e.g., MYC).
<i>Genes, mutations, genotypes, and alleles</i>	

Copyediting manuscripts

Prior to submission, authors who believe their manuscripts would benefit from professional editing are encouraged to use language-editing and copyediting services. Obtaining this service is the responsibility of the author, and should be done before initial submission. These services can be found on the web using search terms like “scientific editing service” or “manuscript editing service.”

Submissions are not copyedited before publication.

Submissions that do not meet the [PLOS ONE publication criterion for language standards](#) may be rejected.

Manuscript Organization

Manuscripts should be organized as follows. Instructions for each element appear below the list.

Beginning section *The following elements are required, in order:*

- Title page: List title, authors, and affiliations as first page of manuscript
- Abstract

	<ul style="list-style-type: none"> • Introduction
Middle section	<p><i>The following elements can be renamed as needed and presented in any order:</i></p> <ul style="list-style-type: none"> • Materials and Methods • Results • Discussion • Conclusions (optional)
Ending section	<p><i>The following elements are required, in order:</i></p> <ul style="list-style-type: none"> • Acknowledgments • References • Supporting information captions (if applicable)
Other elements	<ul style="list-style-type: none"> • Figure captions are inserted immediately after the first paragraph in which the figure is cited. Figure files are uploaded separately. • Tables are inserted immediately after the first paragraph in which they are cited. • Supporting information files are uploaded separately.

Please refer to our downloadable sample files to make sure that your submission meets our formatting requirements:

Viewing Figures and Supporting Information in the compiled submission PDF

The compiled submission PDF includes low-resolution preview images of the figures after the reference list. The function of these previews is to allow you to download the entire submission as quickly as possible. Click the link at the top of each preview page to download a high-resolution version of each figure. Links to download Supporting Information files are also available after the reference list.

Parts of a Submission

Title

Include a full title and a short title for the manuscript.

Title	Length	Guidelines	Examples
Full title	250 characters	Specific, descriptive, concise, and comprehensible to readers outside the field	Impact of Cigarette Smoke Exposure on Innate Immunity: A <i>Caenorhabditis elegans</i> Model

		Solar Drinking Water Disinfection (SODIS) to Reduce Childhood Diarrhoea in Rural Bolivia: A Cluster-Randomized, Controlled Trial
Short title	100 characters	State the topic of the study Cigarette Smoke Exposure and Innate Immunity SODIS and Childhood Diarrhoea

Titles should be written in sentence case (only the first word of the text, proper nouns, and genus species names are capitalized). Avoid specialist abbreviations if possible. For clinical trials, systematic reviews, or meta-analyses, the subtitle should include the study design.

Author List

Authorship requirements

All authors must meet the criteria for authorship as outlined in the authorship policy. [Read the policy.](#) Those who contributed to the work but do not meet the criteria for authorship can be mentioned in the Acknowledgments. [Read more about Acknowledgments.](#)

The corresponding author must provide an ORCID iD at the time of submission by entering it in the user profile in the submission system. [Read more about ORCID.](#)

Author names and affiliations

Enter author names on the title page of the manuscript and in the online submission system.

On the title page, write author names in the following order:

- First name (or initials, if used)
- Middle name (or initials, if used)
- Last name (surname, family name)

Each author on the list must have an affiliation. The affiliation includes department, university, or organizational affiliation and its location, including city, state/province (if applicable), and country.

If an author has multiple affiliations, enter all affiliations on the title page only. In the submission system, enter only the preferred or primary affiliation.

Author names will be published exactly as they appear in the manuscript file. Please double-check the information carefully to make sure it is correct.

Corresponding author

The submitting author is automatically designated as the corresponding author in the submission system. The corresponding author is the primary contact for the journal office and the only author able to view or change the manuscript while it is under editorial consideration.

The corresponding author role may be transferred to another coauthor. However, note that transferring the corresponding author role also transfers access to the manuscript. (To designate a new corresponding author while the manuscript is still under consideration, watch the video tutorial below.)

Only one corresponding author can be designated in the submission system, but this does not restrict the number of corresponding authors that may be listed on the article in the event of publication. Whoever is designated as a corresponding author on the title page of the manuscript file will be listed as such upon publication. Include an email address for each corresponding author listed on the title page of the manuscript.

Consortia and group authorship

If a manuscript is submitted on behalf of a consortium or group, include the consortium or group name in the author list, and include the full list of members in the Acknowledgments or in a supporting information file. [Read the group authorship policy.](#)

Author Contributions

Enter all author contributions in the submission system during submission. The contributions of all authors must be described using the CRediT Taxonomy of author roles. [Read the policy.](#)

Contributions will be published with the final article, and they should accurately reflect contributions to the work. The submitting author is responsible for completing this information at submission, and it is expected that all authors will have reviewed, discussed, and agreed to their individual contributions ahead of this time.

PLOS ONE will contact all authors by email at submission to ensure that they are aware of the submission.

Cover letter

Upload a cover letter as a separate file in the online system. The length limit is 1 page.

The cover letter should include the following information:

- Summarize the study's contribution to the scientific literature
- Relate the study to previously published work
- Specify the type of article (for example, research article, systematic review, meta-analysis, clinical trial)
- Describe any prior interactions with PLOS regarding the submitted manuscript
- Suggest appropriate Academic Editors to handle your manuscript ([see the full list of Academic Editors](#))

- List any opposed reviewers

IMPORTANT: Do not include requests to reduce or waive publication fees in the cover letter. This information will be entered separately in the online submission system.

[Read about publication fee assistance.](#)

Title page

The title, authors, and affiliations should all be included on a title page as the first page of the manuscript file.

[Download sample title, author list, and affiliations page \(PDF\)](#)

Abstract

The Abstract comes after the title page in the manuscript file. The abstract text is also entered in a separate field in the submission system.

The Abstract should:

- Describe the main objective(s) of the study
- Explain how the study was done, including any model organisms used, without methodological detail
- Summarize the most important results and their significance
- Not exceed 300 words

Abstracts should not include:

- Citations
- Abbreviations, if possible

Introduction

The introduction should:

- Provide background that puts the manuscript into context and allows readers outside the field to understand the purpose and significance of the study
- Define the problem addressed and why it is important
- Include a brief review of the key literature
- Note any relevant controversies or disagreements in the field
- Conclude with a brief statement of the overall aim of the work and a comment about whether that aim was achieved

Materials and Methods

The Materials and Methods section should provide enough detail to allow suitably skilled investigators to fully replicate your study. Specific information and/or protocols for new methods should be included in detail. If materials, methods, and protocols are well established, authors may cite articles where those protocols are described in detail, but the submission should include sufficient information to be understood independent of these references.

We encourage authors to submit detailed protocols for newer or less well-established methods as supporting information. [Read the supporting information guidelines.](#)

Human or animal subjects and/or tissue or field sampling

Methods sections describing research using human or animal subjects and/or tissue or field sampling must include required ethics statements. [See the reporting guidelines for human research, clinical trials, animal research, and observational and field studies for more information.](#)

Data

PLOS journals require authors to make all data underlying the findings described in their manuscript fully available without restriction, with rare exception.

Large data sets, including raw data, may be deposited in an appropriate public repository. [See our list of recommended repositories.](#)

For smaller data sets and certain data types, authors may provide their data within [supporting information files](#) accompanying the manuscript. Authors should take care to maximize the accessibility and reusability of the data by selecting a file format from which data can be efficiently extracted (for example, spreadsheets or flat files should be provided rather than PDFs when providing tabulated data).

For more information on how best to provide data, read our [policy on data availability](#). PLOS does not accept references to “data not shown.”

Cell lines

Methods sections describing research using cell lines must state the origin of the cell lines used. [See the reporting guidelines for cell line research for more information.](#)

New taxon names

Methods sections of manuscripts adding new taxon names to the literature must follow the [reporting guidelines below for a new zoological taxon, botanical taxon, or fungal taxon.](#)

Results, Discussion, Conclusions

These sections may all be separate, or may be combined to create a mixed Results/Discussion section (commonly labeled “Results and Discussion”) or a mixed Discussion/Conclusions section (commonly labeled “Discussion”). These sections may

be further divided into subsections, each with a concise subheading, as appropriate. These sections have no word limit, but the language should be clear and concise.

Together, these sections should describe the results of the experiments, the interpretation of these results, and the conclusions that can be drawn.

Authors should explain how the results relate to the hypothesis presented as the basis of the study and provide a succinct explanation of the implications of the findings, particularly in relation to previous related studies and potential future directions for research.

PLOS ONE editorial decisions do not rely on perceived significance or impact, so authors should avoid overstating their conclusions. See the [PLOS ONE Criteria for Publication](#) for more information.

Acknowledgments

Those who contributed to the work but do not meet our authorship criteria should be listed in the Acknowledgments with a description of the contribution.

Authors are responsible for ensuring that anyone named in the Acknowledgments agrees to be named.

Do not include funding sources in the Acknowledgments or anywhere else in the manuscript file. Funding information should only be entered in the financial disclosure section of the submission system.

References

Any and all available works can be cited in the reference list. Acceptable sources include:

- Published or accepted manuscripts
- Manuscripts on preprint servers, if the manuscript is submitted to a journal and also publicly available as a preprint

Do not cite the following sources in the reference list:

- Unavailable and unpublished work, including manuscripts that have been submitted but not yet accepted (e.g., “unpublished work,” “data not shown”). Instead, include those data as supplementary material or deposit the data in a publicly available database.
- Personal communications (these should be supported by a letter from the relevant authors but not included in the reference list)

References are listed at the end of the manuscript and numbered in the order that they appear in the text. In the text, cite the reference number in square brackets (e.g., “We used the techniques developed by our colleagues [19] to analyze the data”). PLOS uses the numbered citation (citation-sequence) method and first six authors, et al.

Do not include citations in abstracts or author summaries.

Make sure the parts of the manuscript are in the correct order *before* ordering the citations.

Formatting references

Because all references will be linked electronically as much as possible to the papers they cite, proper formatting of the references is crucial.

PLOS uses the reference style outlined by the International Committee of Medical Journal Editors (ICMJE), also referred to as the “Vancouver” style. Example formats are listed below. Additional examples are in the [ICMJE sample references](#).

A reference management tool, EndNote, offers a current [style file](#) that can assist you with the formatting of your references. If you have problems with any reference management program, please contact the source company's technical support.

Journal name abbreviations should be those found in the [National Center for Biotechnology Information \(NCBI\) databases](#).

Source	Format
Published articles	<p>Hou WR, Hou YL, Wu GF, Song Y, Su XL, Sun B, et al. cDNA, genomic sequence cloning and overexpression of ribosomal protein gene L9 (rpL9) of the giant panda (<i>Ailuropoda melanoleuca</i>). Genet Mol Res. 2011;10: 1576-1588.</p> <p>Devaraju P, Gulati R, Antony PT, Mithun CB, Negi VS. Susceptibility to SLE in South Indian Tamils may be influenced by genetic selection pressure on TLR2 and TLR9 genes. Mol Immunol. 2014 Nov 22. pii: S0161-5890(14)00313-7. doi: 10.1016/j.molimm.2014.11.005</p> <p><i>Note: A DOI number for the full-text article is acceptable as an alternative to or in addition to traditional volume and page numbers.</i></p>
Accepted, unpublished articles	Same as published articles, but substitute “Forthcoming” for page numbers or DOI.
Web sites or online articles	Huynen MMTE, Martens P, Hilderlink HBM. The health impacts of globalisation: a conceptual framework. Global Health. 2005;1: 14. Available from: http://www.globalizationandhealth.com/content/1/1/14 .
Books	Bates B. Bargaining for life: A social history of tuberculosis. 1st ed. Philadelphia: University of Pennsylvania Press; 1992.
Book chapters	Hansen B. New York City epidemics and history for the public. In: Harden VA, Risse GB, editors. AIDS and the historian. Bethesda: National Institutes of Health; 1991. pp. 21-28.

Source	Format
Deposited articles (preprints, e-prints, or arXiv)	Krick T, Shub DA, Verstraete N, Ferreiro DU, Alonso LG, Shub M, et al. Amino acid metabolism conflicts with protein diversity; 1991. Preprint. Available from: arXiv:1403.3301v1. Cited 17 March 2014.
Published media (print or online newspapers and magazine articles)	Fountain H. For Already Vulnerable Penguins, Study Finds Climate Change Is Another Danger. The New York Times. 29 Jan 2014. Available from: http://www.nytimes.com/2014/01/30/science/earth/climate-change-taking-toll-on-penguins-study-finds.html . Cited 17 March 2014.
New media (blogs, web sites, or other written works)	Allen L. Announcing PLOS Blogs. 2010 Sep 1 [cited 17 March 2014]. In: PLOS Blogs [Internet]. San Francisco: PLOS 2006 - . [about 2 screens]. Available from: http://blogs.plos.org/plos/2010/09/announcing-plos-blogs/ .
Masters' theses or doctoral dissertations	Wells A. Exploring the development of the independent, electronic, scholarly journal. M.Sc. Thesis, The University of Sheffield. 1999. Available from: http://cumincad.scix.net/cgi-bin/works/Show?2e09
Databases and repositories (Figshare, arXiv)	Roberts SB. QPX Genome Browser Feature Tracks; 2013 [cited 2013 Oct 5]. Database: figshare [Internet]. Available from: http://figshare.com/articles/QPX_Genome_Browser_Feature_Tracks/701214 .
Multimedia (videos, movies, or TV shows)	Hitchcock A, producer and director. Rear Window [Film]; 1954. Los Angeles: MGM.

Supporting Information

Authors can submit essential supporting files and multimedia files along with their manuscripts. All supporting information will be subject to peer review. All file types can be submitted, but files must be smaller than 10 MB in size.

Authors may use almost any description as the item name for a supporting information file as long as it contains an “S” and number. For example, “S1 Appendix” and “S2 Appendix,” “S1 Table” and “S2 Table,” and so forth.

Supporting information files are published exactly as provided, and are not copyedited.

Supporting information captions

List supporting information captions at the end of the manuscript file. Do not submit captions in a separate file.

The file number and name are required in a caption, and we highly recommend including a one-line title as well. You may also include a legend in your caption, but it is not required.

Example caption

S1 Text. Title is strongly recommended. Legend is optional.

In-text citations

We recommend that you cite supporting information in the manuscript text, but this is not a requirement. If you cite supporting information in the text, citations do not need to be in numerical order.

Figures and Tables

Figures

Do not include figures in the main manuscript file. Each figure must be prepared and submitted as an individual file.

Cite figures in ascending numeric order upon first appearance in the manuscript file.

Figure captions

Figure captions must be inserted in the text of the manuscript, immediately following the paragraph in which the figure is first cited (read order). Do not include captions as part of the figure files themselves or submit them in a separate document.

At a minimum, include the following in your figure captions:

- A figure label with Arabic numerals, and “Figure” abbreviated to “Fig” (e.g. Fig 1, Fig 2, Fig 3, etc). Match the label of your figure with the name of the file uploaded at submission (e.g. a figure citation of “Fig 1” must refer to a figure file named “Fig1.tif”).
- A concise, descriptive title

The caption may also include a legend as needed.

Tables

Cite tables in ascending numeric order upon first appearance in the manuscript file.

Place each table in your manuscript file directly after the paragraph in which it is first cited (read order). Do not submit your tables in separate files.

Tables require a label (e.g., “Table 1”) and brief descriptive title to be placed above the table. Place legends, footnotes, and other text below the table.

Data reporting

All data and related metadata underlying the findings reported in a submitted manuscript should be deposited in an appropriate public repository, unless already provided as part of the submitted article.

Repositories may be either subject-specific (where these exist) and accept specific types of structured data, or generalist repositories that accept multiple data types. We recommend that authors select repositories appropriate to their field. Repositories may

be subject-specific (e.g., GenBank for sequences and PDB for structures), general, or institutional, as long as DOIs or accession numbers are provided and the data are at least as open as CC BY. Authors are encouraged to select repositories that meet accepted criteria as trustworthy digital repositories, such as criteria of the Centre for Research Libraries or Data Seal of Approval. Large, international databases are more likely to persist than small, local ones.

To support data sharing and author compliance of the PLOS data policy, we have integrated our submission process with a select set of data repositories. The list is neither representative nor exhaustive of the suitable repositories available to authors. Current repository integration partners include [Dryad](#) and [FlowRepository](#). Please contact data@plos.org to make recommendations for further partnerships.

Instructions for PLOS submissions with data deposited in an integration partner repository:

- Deposit data in the integrated repository of choice.
- Once deposition is final and complete, the repository will provide you with a dataset DOI (provisional) and private URL for reviewers to gain access to the data.
- Enter the given data DOI into the full Data Availability Statement, which is requested in the Additional Information section of the PLOS submission form. Then provide the URL passcode in the Attach Files section.

If you have any questions, please [email us](#).

Accession numbers

All appropriate data sets, images, and information should be deposited in an appropriate public repository. [See our list of recommended repositories](#).

Accession numbers (and version numbers, if appropriate) should be provided in the Data Availability Statement. Accession numbers or a citation to the DOI should also be provided when the data set is mentioned within the manuscript.

In some cases authors may not be able to obtain accession numbers of DOIs until the manuscript is accepted; in these cases, the authors must provide these numbers at acceptance. In all other cases, these numbers must be provided at submission.

Identifiers

As much as possible, please provide accession numbers or identifiers for all entities such as genes, proteins, mutants, diseases, etc., for which there is an entry in a public database, for example:

- [Ensembl](#)
- [Entrez Gene](#)
- [FlyBase](#)
- [InterPro](#)
- [Mouse Genome Database \(MGD\)](#)

- [Online Mendelian Inheritance in Man \(OMIM\)](#)
- [PubChem](#)

Identifiers should be provided in parentheses after the entity on first use.

Striking image

You can choose to upload a “Striking Image” that we may use to represent your article online in places like the journal homepage or in search results.

The striking image must be derived from a figure or supporting information file from the submission, i.e., a cropped portion of an image or the entire image. Striking images should ideally be high resolution, eye-catching, single panel images, and should ideally avoid containing added details such as text, scale bars, and arrows.

If no striking image is uploaded, we will designate a figure from the submission as the striking image.

Striking images should not contain potentially identifying images of people. [Read our policy on identifying information.](#)

[The PLOS licenses and copyright policy](#) also applies to striking images.

Additional Information Requested at Submission

Funding statement

This information should not be in your manuscript file; you will provide it via our submission system.

This information will be published with the final manuscript, if accepted, so please make sure that this is accurate and as detailed as possible. You should not include this information in your manuscript file, but it is important to gather it prior to submission, because your financial disclosure statement cannot be changed after initial submission.

Your statement should include relevant grant numbers and the URL of any funder's web site. Please also state whether any individuals employed or contracted by the funders (other than the named authors) played any role in: study design, data collection and analysis, decision to publish, or preparation of the manuscript. If so, please name the individual and describe their role.

Competing interests

This information should not be in your manuscript file; you will provide it via our submission system.

All potential competing interests must be declared in full. If the submission is related to any patents, patent applications, or products in development or for market, these details, including patent numbers and titles, must be disclosed in full.

Manuscripts disputing published work

For manuscripts disputing previously published work, it is *PLOS ONE* policy to invite input from the disputed author during the peer review process. This procedure is aimed at ensuring a thorough, transparent, and productive review process.

If the disputed author chooses to submit a review, it must be returned in a timely fashion and contain a full declaration of all competing interests. The Academic Editor will consider any such reviews in light of the competing interest.

Authors submitting manuscripts disputing previous work should explain the relationship between the manuscripts in their cover letter, and will be required to confirm that they accept the conditions of this review policy before the manuscript is considered further.

Related manuscripts

Upon submission, authors must confirm that the manuscript, or any related manuscript, is not currently under consideration or accepted elsewhere. If related work has been submitted to *PLOS ONE* or elsewhere, authors must include a copy with the submitted article. Reviewers will be asked to comment on the overlap between related submissions.

We strongly discourage the unnecessary division of related work into separate manuscripts, and we will not consider manuscripts that are divided into “parts.” Each submission to *PLOS ONE* must be written as an independent unit and should not rely on any work that has not already been accepted for publication. If related manuscripts are submitted to *PLOS ONE*, the authors may be advised to combine them into a single manuscript at the editor's discretion.

PLOS does support authors who wish to share their work early and receive feedback before formal peer review. Deposition of manuscripts with preprint servers does not impact consideration of the manuscript at any PLOS journal.

Authors choosing bioRxiv may now concurrently submit directly to select PLOS journals through [bioRxiv's direct transfer to journal service](#).



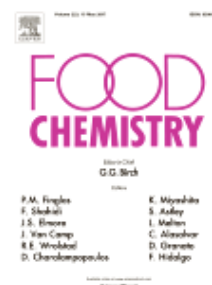
ANEXO 3: Normas (Food Chemistry)

FOOD CHEMISTRY

AUTHOR INFORMATION PACK

TABLE OF CONTENTS

●	Description	p.1
●	Audience	p.2
●	Impact Factor	p.2
●	Abstracting and Indexing	p.2
●	Editorial Board	p.2
●	Guide for Authors	p.4



ISSN: 0308-8146

DESCRIPTION

Food Chemistry publishes original research papers dealing with the advancement of the **chemistry** and **biochemistry** of **foods** or the analytical methods/ approach used. All papers should focus on the novelty of the research carried out.

Topics include:

- Chemistry relating to major and minor **components of food**, their nutritional, physiological, sensory, flavour and microbiological aspects;
- **Bioactive constituents** of foods, including antioxidants, phytochemicals, and botanicals. Data must accompany sufficient discussion to demonstrate their relevance to food and/or food chemistry;
- Chemical and biochemical composition and structure changes in molecules induced by processing, distribution and domestic conditions;
- **Effects of processing** on the composition, quality and safety of foods, other bio-based materials, by-products, and processing wastes;
- Chemistry of **food additives, contaminants**, and other agro-chemicals, together with their metabolism, toxicology and food fate.

Analytical papers related to the microbiological, sensory, nutritional, physiological, authenticity and origin aspects of food. Papers should be primarily concerned with new or novel methods (especially instrumental or rapid) provided adequate validation is described including sufficient data from real samples to demonstrate robustness. Papers dealing with significant improvements to existing methods, or data from application of existing methods to new foods, or commodities produced in unreported geographical areas, will also be considered.

- Methods for the determination of both major and minor components of food especially nutrients and non-nutrient bioactive compounds (with putative health benefits) will be considered.
- Results of method inter-comparison studies and development of food reference materials for use in the assay of food components;

- Methods concerned with the chemical forms in food, nutrient bioavailability and nutritional status;
- General authentication and origin [e.g. Country of Origin Labelling (COOL), Protected Designation of Origin (PDO), Protected Geographical Indication (PGI), Certificate of Specific Character (CSC)] determination of foods (both geographical and production including commodity substitution, and verification of organic, biological and ecological labelling) providing sufficient data from authentic samples should be included to ensure that interpretations are meaningful.

Food Chemistry will not consider papers that focus on purely clinical or engineering aspects without any contribution to chemistry; pharmaceutical or non-food herbal remedies; traditional or folk medicines; or survey/surveillance data.

Papers on therapeutic application of food compounds/isolates for treatment, cure or prevention of human diseases will not be considered for inclusion in *Food Chemistry*.

AUDIENCE

Food technologists, scientists and chemists

IMPACT FACTOR

2015: 4.052 © Thomson Reuters Journal Citation Reports 2016

ABSTRACTING AND INDEXING

BIOSIS
 Chemical Abstracts
 Chemical Engineering Biotechnology Abstracts
 Current Contents
 EMBASE
 FSTA (Food Science and Technology Abstracts)
 Nutrition Abstracts
 Publications in Food Microbiology
 SCISEARCH
 Science Citation Index
 CAB Abstracts
 Sociedad Iberoamericana de Informacion Cientifica (SIIC) Data Bases
 Scopus
 Global Health
 EMBiology

EDITORIAL BOARD

Editor-in-Chief

G.G. Birch, Food and Nutritional Sciences, University of Reading, PO Box 217 Whiteknights, Reading, RG6 6AH, UK

Editor: Analytical, Nutritional and Clinical Methods Section

P. Finglas, Inst. of Food Research, Nutrition Health & Con, Norwich Laboratory, Colney Lane, NR4 7UA, Colney, Norwich, UK

Editors

S.B. Astley, EuroFIR AISBL, Brussels, Belgium
D. Charalampopoulos, University of Reading, Reading, UK
S. Elmore, University of Reading, Reading, England, UK
L. Melton, University of Auckland, Auckland, New Zealand
K. Miyashita, Hokkaido University, Hakodate, Japan
F. Shahidi, Memorial University of Newfoundland, St John's, Canada
J. Van Camp, Universiteit Gent, Gent, Belgium

R.E. Wrolstad, Oregon State University, Corvallis, Oregon, USA

Associate Editors

C. Alasalvar, TÜBİTAK Marmara Research Center, Gebze/Kocaeli, Turkey

S. Baumgartner, Universität für Bodenkultur Wien (BOKU), Vienna, Austria

D. Granato, Universidade Estadual de Ponta Grossa, Ponta Grossa, Brazil

F. Hidalgo, Instituto de la Grasa (IG), Sevilla, Spain

A. Ismail, Universiti Putra Malaysia, Serdang, Selangor, Malaysia

P. Kilmartin, University of Auckland, Auckland Mail Centre, Auckland, New Zealand

R.B. Pegg, University of Georgia, Athens, Georgia, USA

Editorial Board Members

R. Aluko, University of Manitoba, Winnipeg, Manitoba, Canada

R. Amarowicz, Polish Academy of Sciences, Olsztyn, Poland

P. Andrade, University of Porto, Porto, Portugal

S.G. Anema, Fonterra, Palmerston North, New Zealand

J.H. Banoub, Fisheries and Oceans Canada, St. John's, Newfoundland and Labrador, Canada

J.C.M. Barreira, Polytechnic Institute of Bragança, Bragança, Portugal

M. Battino, University of Ancona, Ancona, Italy

J.C. Beaulieu, U.S. Department of Agriculture (USDA), Agricultural Research Service (ARS), New Orleans, Louisiana, USA

R.G. Berger, Leibniz University Hannover, Hannover, Germany

T. Beta, University of Manitoba, Winnipeg, Manitoba, Canada

M. Betti, University of Alberta, Edmonton, Alberta, Canada

J. Birch, University of Otago, Dunedin, New Zealand

Y. Chen, Jiangnan University, Wuxi, China

A. Escarpa, University of Alcalá, Alcalá de Henares, Madrid, Spain

B. Fedrizzi, University of Auckland, Auckland, New Zealand

M. Gidley, University of Queensland, St Lucia, Queensland, Australia

V. Gökmen, Hacettepe University, Ankara, Turkey

M. Jenner, Devon, UK

O.G. Jones, Purdue University, West Lafayette, Indiana, USA

M. Jung, Woosuk University, Jeonbuk, South Korea

S. Kelly, University of East Anglia, Norwich, England, UK

J.F. Kennedy, Chembiochem Laboratories, Worcester, England, UK

J. Lakkis, Pfizer Global Research and Development, Morris Plains, New Jersey, USA

J. Lee, U.S. Department of Agriculture (USDA), Agricultural Research Service (ARS), Parma, Idaho, USA

M. Mathlouthi, Université de Reims Champagne-Ardenne, Reims Cedex, France

S. Polesello, National Research Council of Italy (CNR), Brugherio, Italy

S. Porretta, Stazione Sperimentale per L'Industria delle Conserve Alimentari, Parma, Italy

P. Putnik, University of Zagreb, Zagreb, Croatia

P. Puwastien, Mahidol University, Nakhon Pathom, Thailand

A. Ritieni, Università di Napoli Federico II, Napoli, Italy

B. Saad, Universiti Sains Malaysia, Nibong Tebal, Penang, Malaysia

H. Schönfeldt, University of Pretoria, Pretoria, South Africa

J.-H. Shim, Chonnam National University, Gwangju, South Korea

K. Thurlow, LGC, Teddington, England, UK

F. Toldrá, Institute of Agricultural Chemistry and Food Technology, Paterna (Valencia), Spain

R. Tsao, Agriculture and Agri-Food Canada (AAFC), Guelph, Ontario, Canada

A. Tudos, Shell Global Solutions, Amsterdam, Netherlands

V. Yaylayan, McGill University, Ste Anne de Bellevue, Quebec, Canada

L. Yu, University of Maryland, College Park, Maryland, USA

R. Zeleny, European Commission, Geel, Belgium

J. Zhengyu, Southern Yangtze University, Wuxi, Jiangsu Province, China

Y.J. Zhong, Corbion, Kansas, USA

GUIDE FOR AUTHORS

INTRODUCTION

Ten essential rules to ensure your manuscript is handled promptly

The manuscript fits the Aims and Scope of the journal (<http://www.journals.elsevier.com/food-chemistry>) Manuscript is in accordance with ARTICLE TYPE - GUIDELINES (<http://www.elsevier.com/journals/food-chemistry/0308-8146/guide-for-authors#14000>) The text is written in good English. Authors who feel their manuscript may require editing to conform to correct scientific English may wish to use an English Language Editing service such as the one available from Elsevier's WebShop (<http://webshop.elsevier.com/languageediting/>). Manuscript text is divided into numbered sections; line and page numbers are added and text is double spaced An ethical statement is required for experiments involving humans or animals Conflict of interest statement is included at the end of the manuscript The number of figures and tables combined does not exceed a total of 6; additional tables and figures can be submitted as supplementary material. All relevant references should be provided in the Reference list. Cover letter is prepared, introducing your article and explaining the novelty of the research Highlights are prepared (a birds' eye view of your article in 3-5 points, 85 characters each)

Submission checklist

Checklist can also be downloaded [here](#)

1) Study contents:

The Authors should ensure that The manuscript fits within [Aims & Scope](#) of Food Chemistry. The research is **novel** and has **not been published previously** - please see "Responsible research publication: international standards for authors" from COPE for more information. **Ethical consent** has been obtained in case of work on animals and/or humans.

2) Manuscript preparation:

The Authors should ensure that The number of words and of figures/tables is within limits: • Research article: 7500 words, 6 tables and figures combined

• Review article: 10 000 words, 6 tables and figures combined

• Short communication: 3000 words, 6 tables and figures combined More tables and figures? Submit as [supplementary material](#) The **title page** contains title, author names, affiliations and corresponding author telephone. **Email addresses are required for ALL authors. Authors must provide and use an e-mail address unique to themselves and not one that is shared with another author registered in EES, or a department.** The **highlights** are provided (3-5 bullet points, max 85 characters each including spaces). The manuscript contains a **conflict of interest** statement (before references) The [language](#) follows the requirements of the Guide for Authors . The [formatting](#) of the manuscript follows the requirements of the Guide for Authors . Continuous **line numbering** is provided throughout the manuscript (including captions and references); **page numbering** is provided. All relevant [references](#) are provided in alphabetical order Figures and tables (6 combined) include clear **labels** and are prepared as **individual files**. The manuscript contains appropriate **ethical approval** and **informed consent** (if applicable, include statement). **3) Before submission: Manuscript** file is provided as a Microsoft Word file. Figures and tables are provided as **individual files** A **cover letter** is included. 3 or more suggested **reviewers** are provided (including affiliation and professional email address), **at least 2 of which are from a different country than the Authors. Keywords** are provided.

Now you are ready to submit at <http://ees.elsevier.com/foodchem>

Types of paper

Original research papers; review articles; rapid communications; short communications; viewpoints; letters to the Editor; book reviews.

1. Research papers - original full-length research papers which have not been published previously, except in a preliminary form, and should not exceed 7,500 words (including no more than 6 tables - additional tables and figures can be submitted as supplementary material). Research papers should not contain more than 40 references.

2. Review articles - will be accepted in areas of topical interest, will normally focus on literature published over the previous five years, and should not exceed 10,000 words (including allowance for no more than 6 tables and illustrations). Review articles should not contain more than 80 references.) If it is felt absolutely necessary to exceed this number, please contact the editorial office for advice before submission.

3. Rapid communications - an original research paper reporting a major scientific result or finding with significant implications for the research community, designated by the Editor.

4. Short communications - Short communications of up to 3000 words, describing work that may be of a preliminary nature but which merits immediate publication. These papers should not contain more than 30 references.

5. Viewpoints - Authors may submit viewpoints of about 1200 words on any subject covered by the Aims and Scope.

6. Letters to the Editor - Letters are published from time to time on matters of topical interest.

7. Book reviews

BEFORE YOU BEGIN

Ethics in publishing

Please see our information pages on [Ethics in publishing](#) and [Ethical guidelines for journal publication](#).

Guidelines in the US and Canada, Europe and Australia specifically state that hypothermia (use of ice slurries) is not an acceptable method for killing fish in the research environment.

Declaration of interest

All authors are requested to disclose any actual or potential conflict of interest including any financial, personal or other relationships with other people or organizations within three years of beginning the submitted work that could inappropriately influence, or be perceived to influence, their work. [More information](#).

Submission declaration and verification

Submission of an article implies that the work described has not been published previously (except in the form of an abstract or as part of a published lecture or academic thesis or as an electronic preprint, see '[Multiple, redundant or concurrent publication](#)' section of our ethics policy for more information), that it is not under consideration for publication elsewhere, that its publication is approved by all authors and tacitly or explicitly by the responsible authorities where the work was carried out, and that, if accepted, it will not be published elsewhere in the same form, in English or in any other language, including electronically without the written consent of the copyright-holder. To verify originality, your article may be checked by the originality detection service [CrossCheck](#).

Changes to authorship

Authors are expected to consider carefully the list and order of authors **before** submitting their manuscript and provide the definitive list of authors at the time of the original submission. Any addition, deletion or rearrangement of author names in the authorship list should be made only **before** the manuscript has been accepted and only if approved by the journal Editor. To request such a change, the Editor must receive the following from the **corresponding author**: (a) the reason for the change in author list and (b) written confirmation (e-mail, letter) from all authors that they agree with the addition, removal or rearrangement. In the case of addition or removal of authors, this includes confirmation from the author being added or removed.

Only in exceptional circumstances will the Editor consider the addition, deletion or rearrangement of authors **after** the manuscript has been accepted. While the Editor considers the request, publication of the manuscript will be suspended. If the manuscript has already been published in an online issue, any requests approved by the Editor will result in a corrigendum.

Copyright

Upon acceptance of an article, authors will be asked to complete a 'Journal Publishing Agreement' (see [more information](#) on this). An e-mail will be sent to the corresponding author confirming receipt of the manuscript together with a 'Journal Publishing Agreement' form or a link to the online version of this agreement.

Subscribers may reproduce tables of contents or prepare lists of articles including abstracts for internal circulation within their institutions. [Permission](#) of the Publisher is required for resale or distribution outside the institution and for all other derivative works, including compilations and translations. If excerpts from other copyrighted works are included, the author(s) must obtain written permission from the copyright owners and credit the source(s) in the article. Elsevier has [preprinted forms](#) for use by authors in these cases.

For open access articles: Upon acceptance of an article, authors will be asked to complete an 'Exclusive License Agreement' ([more information](#)). Permitted third party reuse of open access articles is determined by the author's choice of [user license](#).

Author rights

As an author you (or your employer or institution) have certain rights to reuse your work. [More information](#).

Elsevier supports responsible sharing

Find out how you can [share your research](#) published in Elsevier journals.

Role of the funding source

You are requested to identify who provided financial support for the conduct of the research and/or preparation of the article and to briefly describe the role of the sponsor(s), if any, in study design; in the collection, analysis and interpretation of data; in the writing of the report; and in the decision to submit the article for publication. If the funding source(s) had no such involvement then this should be stated.

Funding body agreements and policies

Elsevier has established a number of agreements with funding bodies which allow authors to comply with their funder's open access policies. Some funding bodies will reimburse the author for the Open Access Publication Fee. Details of [existing agreements](#) are available online.

Open access

This journal offers authors a choice in publishing their research:

Open access

- Articles are freely available to both subscribers and the wider public with permitted reuse.
- An open access publication fee is payable by authors or on their behalf, e.g. by their research funder or institution.

Subscription

- Articles are made available to subscribers as well as developing countries and patient groups through our [universal access programs](#).
- No open access publication fee payable by authors.

Regardless of how you choose to publish your article, the journal will apply the same peer review criteria and acceptance standards.

For open access articles, permitted third party (re)use is defined by the following [Creative Commons user licenses](#):

Creative Commons Attribution (CC BY)

Lets others distribute and copy the article, create extracts, abstracts, and other revised versions, adaptations or derivative works of or from an article (such as a translation), include in a collective work (such as an anthology), text or data mine the article, even for commercial purposes, as long as they credit the author(s), do not represent the author as endorsing their adaptation of the article, and do not modify the article in such a way as to damage the author's honor or reputation.

Creative Commons Attribution-NonCommercial-NoDerivs (CC BY-NC-ND)

For non-commercial purposes, lets others distribute and copy the article, and to include in a collective work (such as an anthology), as long as they credit the author(s) and provided they do not alter or modify the article.

The open access publication fee for this journal is **USD 2600**, excluding taxes. Learn more about Elsevier's pricing policy: <https://www.elsevier.com/openaccesspricing>.

Green open access

Authors can share their research in a variety of different ways and Elsevier has a number of green open access options available. We recommend authors see our [green open access page](#) for further information. Authors can also self-archive their manuscripts immediately and enable public access from their institution's repository after an embargo period. This is the version that has been accepted for publication and which typically includes author-incorporated changes suggested during submission, peer review and in editor-author communications. Embargo period: For subscription articles, an appropriate amount of time is needed for journals to deliver value to subscribing customers before an article becomes freely available to the public. This is the embargo period and it begins from the date the article is formally published online in its final and fully citable form. [Find out more.](#)

This journal has an embargo period of 12 months.

Elsevier Publishing Campus

The Elsevier Publishing Campus (www.publishingcampus.com) is an online platform offering free lectures, interactive training and professional advice to support you in publishing your research. The College of Skills training offers modules on how to prepare, write and structure your article and explains how editors will look at your paper when it is submitted for publication. Use these resources, and more, to ensure that your submission will be the best that you can make it.

Language (usage and editing services)

Please write your text in good English (American or British usage is accepted, but not a mixture of these). Authors who feel their English language manuscript may require editing to eliminate possible grammatical or spelling errors and to conform to correct scientific English may wish to use the [English Language Editing service](#) available from Elsevier's WebShop.

Submission

Our online submission system guides you stepwise through the process of entering your article details and uploading your files. The system converts your article files to a single PDF file used in the peer-review process. Editable files (e.g., Word, LaTeX) are required to typeset your article for final publication. All correspondence, including notification of the Editor's decision and requests for revision, is sent by e-mail.

Authors must provide and use an email address unique to themselves and not shared with another author registered in EES, or a department.

Referees

Authors are required to submit with their articles, the names, complete affiliations (spelled out), country and contact details (including current and valid (preferably business) e-mail address) of three potential reviewers. Email addresses and reviewer names will be checked for validity. **Your potential reviewers should not be from your institute, and at least two should be from different countries.** Authors should not suggest reviewers with whom they have collaborated within the past two years. Your submission will be rejected if these are not supplied. Names provided may be used for other submissions on the same topic. Reviewers must have specific expertise on the subject of your article and/or the techniques employed in your study. Briefly state the appropriate expertise of each reviewer.

Review Policy

A peer review system involving two or three reviewers is used to ensure high quality of manuscripts accepted for publication. The Managing Editor and Editors have the right to decline formal review of a manuscript when it is deemed that the manuscript is

- 1) on a topic outside the scope of the Journal;
- 2) lacking technical merit;
- 3) focused on foods or processes that are of narrow regional scope and significance;
- 4) fragmentary and providing marginally incremental results; or
- 5) is poorly written.

PREPARATION

Use of wordprocessing software

General: Manuscripts must be typewritten, double-spaced with wide margins. Each page must be numbered, and lines must be consecutively numbered from the start to the end of the manuscript. Good quality printouts with a font size of 12 or 10 pt are required. The corresponding author should be identified (include a Fax number and E-mail address). Full postal and email addresses must be

given for all co-authors. Authors should consult a recent issue of the journal for style if possible. The Editors reserve the right to adjust style to certain standards of uniformity. Authors should retain a copy of their manuscript since we cannot accept responsibility for damage or loss of papers.

Article structure

Follow this order when typing manuscripts: Title, Authors, Affiliations, Abstract, Keywords, Main text, Acknowledgements, Appendix, References, Vitae, Figure Captions. Do not import the Figures or Tables into your text, figures and tables should be submitted as separate files. The corresponding author should be identified with an asterisk and footnote. All other footnotes (except for table footnotes) should be identified with superscript Arabic numbers. The title of the paper should unambiguously reflect its contents. Where the title exceeds 70 characters a suggestion for an abbreviated running title should be given.

Subdivision - numbered sections

Divide your article into clearly defined and numbered sections. Subsections should be numbered 1.1 (then 1.1.1, 1.1.2, ...), 1.2, etc. (the abstract is not included in section numbering). Use this numbering also for internal cross-referencing: do not just refer to 'the text'. Any subsection may be given a brief heading. Each heading should appear on its own separate line.

Essential title page information

- **Title.** Concise and informative. Titles are often used in information-retrieval systems. Avoid abbreviations and formulae where possible.
- **Author names and affiliations.** Please clearly indicate the given name(s) and family name(s) of each author and check that all names are accurately spelled. Present the authors' affiliation addresses (where the actual work was done) below the names. Indicate all affiliations with a lower-case superscript letter immediately after the author's name and in front of the appropriate address. Provide the full postal address of each affiliation, including the country name and, if available, the e-mail address of each author.
- **Corresponding author.** Clearly indicate who will handle correspondence at all stages of refereeing and publication, also post-publication. **Ensure that the e-mail address is given and that contact details are kept up to date by the corresponding author.**
- **Present/permanent address.** If an author has moved since the work described in the article was done, or was visiting at the time, a 'Present address' (or 'Permanent address') may be indicated as a footnote to that author's name. The address at which the author actually did the work must be retained as the main, affiliation address. Superscript Arabic numerals are used for such footnotes.

Abstract

A concise and factual abstract is required. The abstract should state briefly the purpose of the research, the principal results and major conclusions. An abstract is often presented separately from the article, so it must be able to stand alone. For this reason, References should be avoided, but if essential, then cite the author(s) and year(s). Also, non-standard or uncommon abbreviations should be avoided, but if essential they must be defined at their first mention in the abstract itself.

The abstract should not exceed 150 words.

Highlights

Highlights are mandatory for this journal. They consist of a short collection of bullet points that convey the core findings of the article and should be submitted in a separate editable file in the online submission system. Please use 'Highlights' in the file name and include 3 to 5 bullet points (maximum 85 characters, including spaces, per bullet point). You can view [example Highlights](#) on our information site.

Chemical compounds

You can enrich your article by providing a list of chemical compounds studied in the article. The list of compounds will be used to extract relevant information from the NCBI PubChem Compound database and display it next to the online version of the article on ScienceDirect. You can include up to 10 names of chemical compounds in the article. For each compound, please provide the [PubChem CID](#) of the most relevant record as in the following example: Glutamic acid (PubChem CID:611). Please position the list of compounds immediately below the 'Keywords' section. It is strongly recommended to follow the exact text formatting as in the example below:

Chemical compounds studied in this article

Ethylene glycol (PubChem CID: 174); Plitidepsin (PubChem CID: 44152164); Benzalkonium chloride (PubChem CID: 15865)

[More information.](#)

Formatting of funding sources

List funding sources in this standard way to facilitate compliance to funder's requirements:

Funding: This work was supported by the National Institutes of Health [grant numbers xxxx, yyyy]; the Bill & Melinda Gates Foundation, Seattle, WA [grant number zzzz]; and the United States Institutes of Peace [grant number aaaa].

It is not necessary to include detailed descriptions on the program or type of grants and awards. When funding is from a block grant or other resources available to a university, college, or other research institution, submit the name of the institute or organization that provided the funding.

If no funding has been provided for the research, please include the following sentence:

This research did not receive any specific grant from funding agencies in the public, commercial, or not-for-profit sectors.

Units

Follow internationally accepted rules and conventions: use the international system of units (SI). If other units are mentioned, please give their equivalent in SI.

Temperatures should be given in degrees Celsius. The unit 'billion' is ambiguous and should not be used.

Artwork

Electronic artwork

General points

- Make sure you use uniform lettering and sizing of your original artwork.
- Embed the used fonts if the application provides that option.
- Aim to use the following fonts in your illustrations: Arial, Courier, Times New Roman, Symbol, or use fonts that look similar.
- Number the illustrations according to their sequence in the text.
- Use a logical naming convention for your artwork files.
- Provide captions to illustrations separately.
- Size the illustrations close to the desired dimensions of the published version.
- Submit each illustration as a separate file.

A detailed [guide on electronic artwork](#) is available.

You are urged to visit this site; some excerpts from the detailed information are given here.

Formats

If your electronic artwork is created in a Microsoft Office application (Word, PowerPoint, Excel) then please supply 'as is' in the native document format.

Regardless of the application used other than Microsoft Office, when your electronic artwork is finalized, please 'Save as' or convert the images to one of the following formats (note the resolution requirements for line drawings, halftones, and line/halftone combinations given below):

EPS (or PDF): Vector drawings, embed all used fonts.

TIFF (or JPEG): Color or grayscale photographs (halftones), keep to a minimum of 300 dpi.

TIFF (or JPEG): Bitmapped (pure black & white pixels) line drawings, keep to a minimum of 1000 dpi.

TIFF (or JPEG): Combinations bitmapped line/half-tone (color or grayscale), keep to a minimum of 500 dpi.

Please do not:

- Supply files that are optimized for screen use (e.g., GIF, BMP, PICT, WPG); these typically have a low number of pixels and limited set of colors;
- Supply files that are too low in resolution;
- Submit graphics that are disproportionately large for the content.

Please insert the following text before the standard text - Photographs, charts and diagrams are all to be referred to as "Figure(s)" and should be numbered consecutively in the order to which they are referred. They should accompany the manuscript, but should not be included within the text. All illustrations should be clearly marked with the figure number and the author's name. All figures are to have a caption. Captions should be supplied on a separate sheet.

Color artwork

Please make sure that artwork files are in an acceptable format (TIFF (or JPEG), EPS (or PDF), or MS Office files) and with the correct resolution. If, together with your accepted article, you submit usable color figures then Elsevier will ensure, at no additional charge, that these figures will appear in color online (e.g., ScienceDirect and other sites) regardless of whether or not these illustrations are reproduced in color in the printed version. **For color reproduction in print, you will receive information regarding the costs from Elsevier after receipt of your accepted article.** Please indicate your preference for color: in print or online only. [Further information on the preparation of electronic artwork.](#)

Figure captions

Ensure that each illustration has a caption. Supply captions separately, not attached to the figure. A caption should comprise a brief title (**not** on the figure itself) and a description of the illustration. Keep text in the illustrations themselves to a minimum but explain all symbols and abbreviations used.

Tables

Please submit tables as editable text and not as images. Tables can be placed either next to the relevant text in the article, or on separate page(s) at the end. Number tables consecutively in accordance with their appearance in the text and place any table notes below the table body. Be sparing in the use of tables and ensure that the data presented in them do not duplicate results described elsewhere in the article. Please avoid using vertical rules.

References

Citation in text

Please ensure that every reference cited in the text is also present in the reference list (and vice versa). Any references cited in the abstract must be given in full. Unpublished results and personal communications are not recommended in the reference list, but may be mentioned in the text. If these references are included in the reference list they should follow the standard reference style of the journal and should include a substitution of the publication date with either 'Unpublished results' or 'Personal communication'. Citation of a reference as 'in press' implies that the item has been accepted for publication.

Web references

As a minimum, the full URL should be given and the date when the reference was last accessed. Any further information, if known (DOI, author names, dates, reference to a source publication, etc.), should also be given. Web references can be listed separately (e.g., after the reference list) under a different heading if desired, or can be included in the reference list.

Example: CTAHR (College of Tropical Agriculture and Human Resources, University of Hawaii). Tea (*Camellia sinensis*) a New Crop for Hawaii, 2007. URL http://www.ctahr.hawaii.edu/oc/freepubs/pdf/tea_04_07.pdf. Accessed 14.02.11.

Data references

This journal encourages you to cite underlying or relevant datasets in your manuscript by citing them in your text and including a data reference in your Reference List. Data references should include the following elements: author name(s), dataset title, data repository, version (where available), year, and global persistent identifier. Add [dataset] immediately before the reference so we can properly identify it as a data reference. The [dataset] identifier will not appear in your published article.

Reference management software

Most Elsevier journals have their reference template available in many of the most popular reference management software products. These include all products that support [Citation Style Language styles](#), such as [Mendeley](#) and [Zotero](#), as well as [EndNote](#). Using the word processor plug-ins from these products, authors only need to select the appropriate journal template when preparing their article, after which citations and bibliographies will be automatically formatted in the journal's style. If no template is yet available for this journal, please follow the format of the sample references and citations as shown in this Guide.

Users of Mendeley Desktop can easily install the reference style for this journal by clicking the following link:

<http://open.mendeley.com/use-citation-style/food-chemistry>

When preparing your manuscript, you will then be able to select this style using the Mendeley plug-ins for Microsoft Word or LibreOffice.

All publications cited in the text should be presented in a list of references following the text of the manuscript. See Types of Paper for reference number limits. In the text refer to the author's name (without initials) and year of publication (e.g. "Steventon, Donald and Gladden (1994) studied the effects..." or "...similar to values reported by others (Anderson, Douglas, Morrison & Weiping, 1990)..."). For 2-6 authors all authors are to be listed at first citation. At subsequent citations use first author et al.. When there are more than 6 authors, first author et al. should be used throughout the text. The list of references should be arranged alphabetically by authors' names and should be as full as possible, listing all authors, the full title of articles and journals, publisher and year. The manuscript should be carefully checked to ensure that the spelling of authors' names and dates are exactly the same in the text as in the reference list.

Reference style

Text: Citations in the text should follow the referencing style used by the American Psychological Association. You are referred to the Publication Manual of the American Psychological Association, Sixth Edition, ISBN 978-1-4338-0561-5, copies of which may be [ordered online](#) or APA Order Dept., P.O.B. 2710, Hyattsville, MD 20784, USA or APA, 3 Henrietta Street, London, WC3E 8LU, UK.

List: references should be arranged first alphabetically and then further sorted chronologically if necessary. More than one reference from the same author(s) in the same year must be identified by the letters 'a', 'b', 'c', etc., placed after the year of publication.

Examples:

Reference to a journal publication:

Van der Geer, J., Hanraads, J. A. J., & Lupton, R. A. (2010). The art of writing a scientific article. *Journal of Scientific Communications*, 163, 51–59.

Reference to a book:

Strunk, W., Jr., & White, E. B. (2000). *The elements of style*. (4th ed.). New York: Longman, (Chapter 4).

Reference to a chapter in an edited book:

Mettam, G. R., & Adams, L. B. (2009). How to prepare an electronic version of your article. In B. S. Jones, & R. Z. Smith (Eds.), *Introduction to the electronic age* (pp. 281–304). New York: E-Publishing Inc.

Reference to a website:

Cancer Research UK. Cancer statistics reports for the UK. (2003). <http://www.cancerresearchuk.org/aboutcancer/statistics/cancerstatsreport/> Accessed 13.03.03.

Reference to a dataset:

[dataset] Oguro, M., Imahiro, S., Saito, S., Nakashizuka, T. (2015). *Mortality data for Japanese oak wilt disease and surrounding forest compositions*. Mendeley Data, v1. <http://dx.doi.org/10.17632/xwj98nb39r.1>.

Supplementary material

Supplementary material can support and enhance your scientific research. Supplementary files offer the author additional possibilities to publish supporting applications, high-resolution images, background datasets, sound clips and more. Please note that such items are published online exactly as they are submitted; there is no typesetting involved (supplementary data supplied as an Excel file or as a PowerPoint slide will appear as such online). Please submit the material together with the article and supply a concise and descriptive caption for each file. If you wish to make any changes to supplementary data during any stage of the process, then please make sure to provide an updated file, and do not annotate any corrections on a previous version. Please also make sure to switch off the 'Track Changes' option in any Microsoft Office files as these will appear in the published supplementary file(s). For more detailed instructions please visit our [artwork instruction pages](#).

RESEARCH DATA

Data in Brief

Authors have the option of converting any or all parts of their supplementary or additional raw data into one or multiple Data in Brief articles, a new kind of article that houses and describes their data. Data in Brief articles ensure that your data, which is normally buried in supplementary material, is actively reviewed, curated, formatted, indexed, given a DOI and publicly available to all upon publication. Authors are encouraged to submit their Data in Brief article as an additional item directly alongside the revised version of their manuscript. If your research article is accepted, your Data in Brief article will automatically be transferred over to *Data in Brief* where it will be editorially reviewed

and published in the new, open access journal, *Data in Brief*. Please note an open access fee is payable for publication in *Data in Brief*. Full details can be found on the [Data in Brief website](#). Please use [this template](#) to write your Data in Brief.

Database linking

Elsevier encourages authors to connect articles with external databases, giving readers access to relevant databases that help to build a better understanding of the described research. Please refer to relevant database identifiers using the following format in your article: Database: xxxx (e.g., TAIR: AT1G01020; CCDC: 734053; PDB: 1XFN). [More information and a full list of supported databases](#).

AudioSlides

The journal encourages authors to create an AudioSlides presentation with their published article. AudioSlides are brief, webinar-style presentations that are shown next to the online article on ScienceDirect. This gives authors the opportunity to summarize their research in their own words and to help readers understand what the paper is about. [More information and examples are available](#). Authors of this journal will automatically receive an invitation e-mail to create an AudioSlides presentation after acceptance of their paper.

Interactive plots

This journal enables you to show an Interactive Plot with your article by simply submitting a data file. [Full instructions](#).

Additional information

Abbreviations for units should follow the suggestions of the British Standards publication BS 1991. The full stop should not be included in abbreviations, e.g. m (not m.), ppm (not p.p.m.), % and '/' should be used in preference to 'per cent' and 'per'. Where abbreviations are likely to cause ambiguity or may not be readily understood by an international readership, units should be put in full.

Current recognised (IUPAC) chemical nomenclature should be used, although commonly accepted trivial names may be used where there is no risk of ambiguity.

The use of proprietary names should be avoided. Papers essentially of an advertising nature will not be accepted.

AFTER ACCEPTANCE

Online proof correction

Corresponding authors will receive an e-mail with a link to our online proofing system, allowing annotation and correction of proofs online. The environment is similar to MS Word: in addition to editing text, you can also comment on figures/tables and answer questions from the Copy Editor. Web-based proofing provides a faster and less error-prone process by allowing you to directly type your corrections, eliminating the potential introduction of errors.

If preferred, you can still choose to annotate and upload your edits on the PDF version. All instructions for proofing will be given in the e-mail we send to authors, including alternative methods to the online version and PDF.

We will do everything possible to get your article published quickly and accurately. Please use this proof only for checking the typesetting, editing, completeness and correctness of the text, tables and figures. Significant changes to the article as accepted for publication will only be considered at this stage with permission from the Editor. It is important to ensure that all corrections are sent back to us in one communication. Please check carefully before replying, as inclusion of any subsequent corrections cannot be guaranteed. Proofreading is solely your responsibility.

Offprints

The corresponding author will, at no cost, receive a customized [Share Link](#) providing 50 days free access to the final published version of the article on [ScienceDirect](#). The Share Link can be used for sharing the article via any communication channel, including email and social media. For an extra charge, paper offprints can be ordered via the offprint order form which is sent once the article is accepted for publication. Both corresponding and co-authors may order offprints at any time via Elsevier's [Webshop](#). Corresponding authors who have published their article open access do not receive a Share Link as their final published version of the article is available open access on ScienceDirect and can be shared through the article DOI link.

AUTHOR INQUIRIES

Visit the [Elsevier Support Center](#) to find the answers you need.

